

## **Subglacial carbonate deposits as a potential proxy for glacier's existence**

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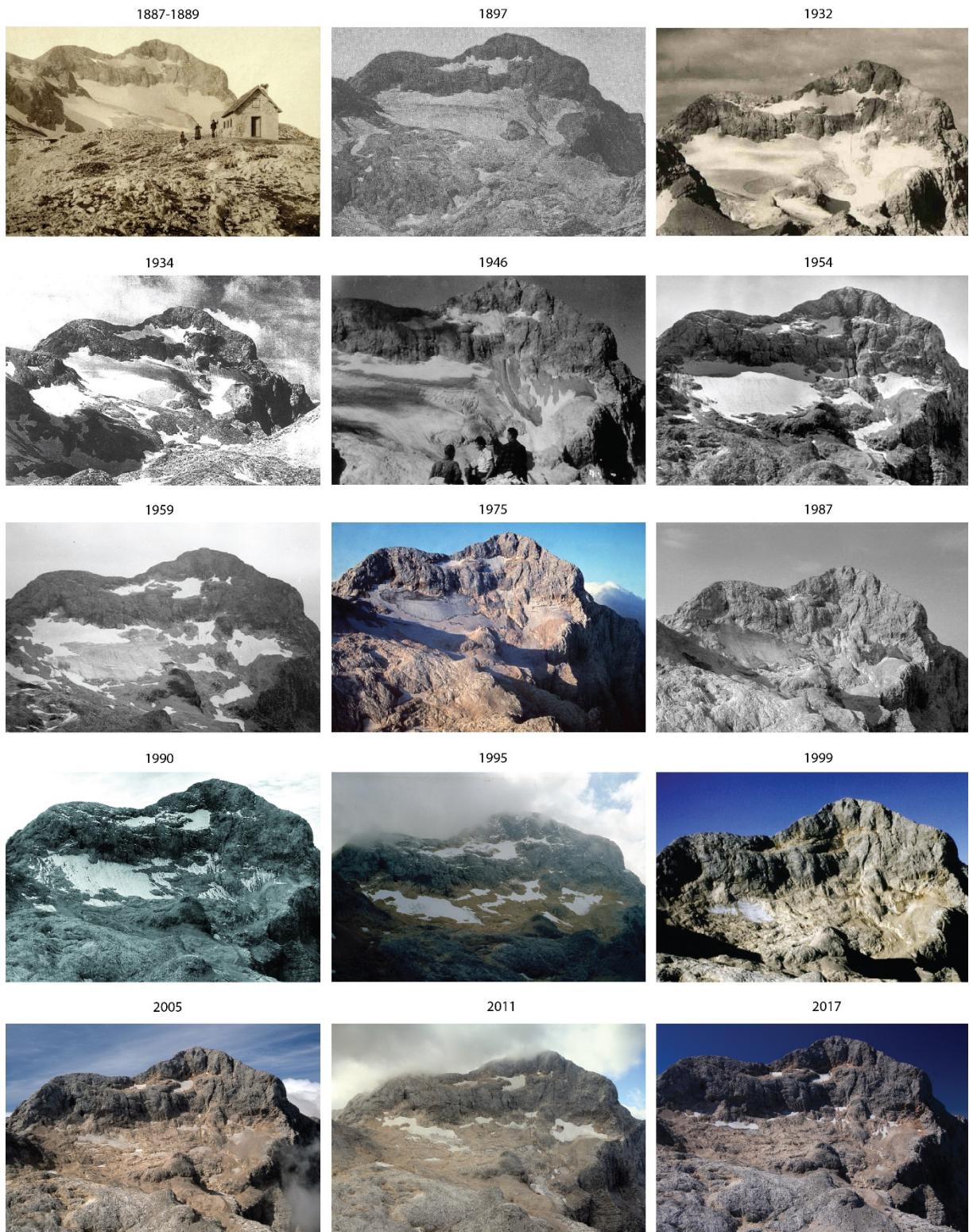
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## **Supplementary Information**

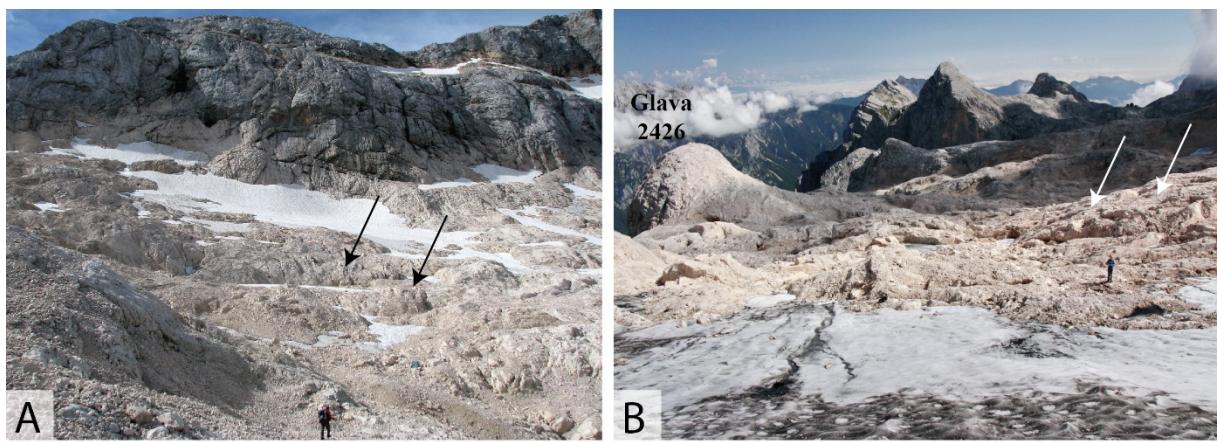
Supplementary Figure S1: Photographic material of the retreating Triglav Glacier (© ZRC SAZU Anton Melik Geographical Institute archive).



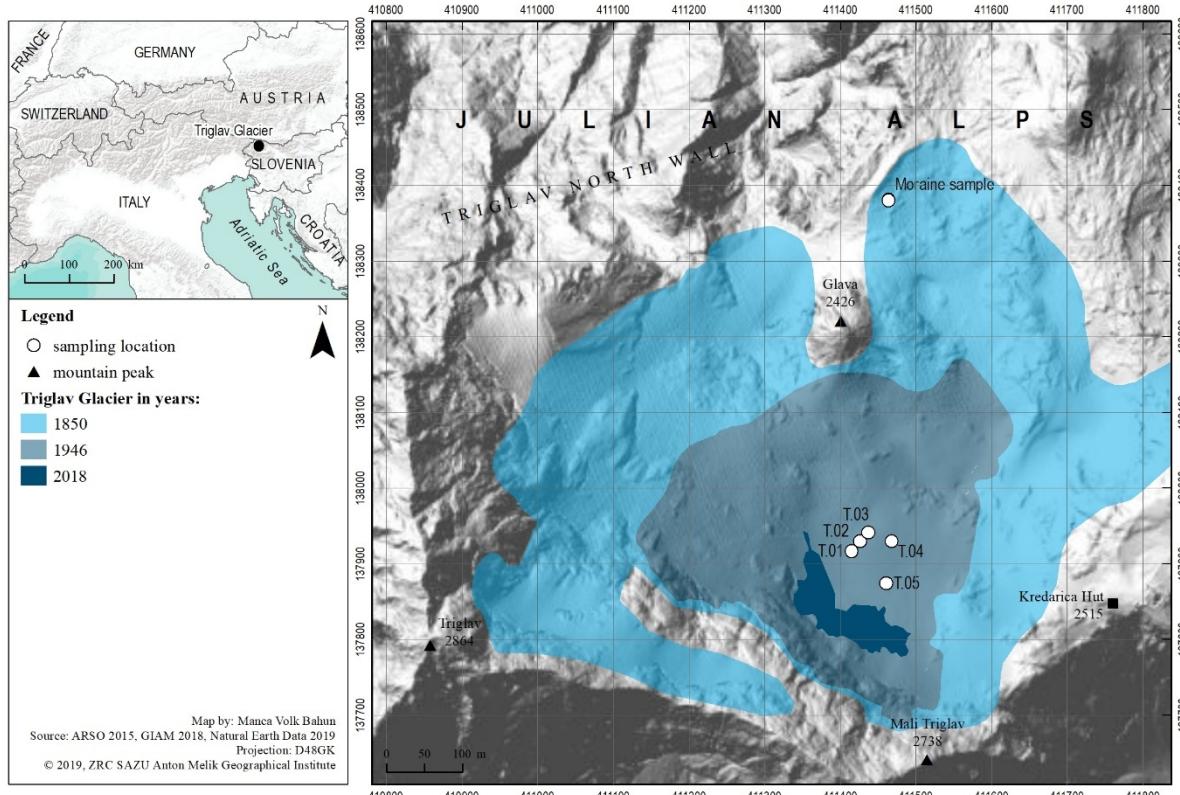
Supplementary Figure S2: The recently exposed surface with shafts and subglacial carbonate deposits (© ZRC SAZU Anton Melik Geographical Institute archive).



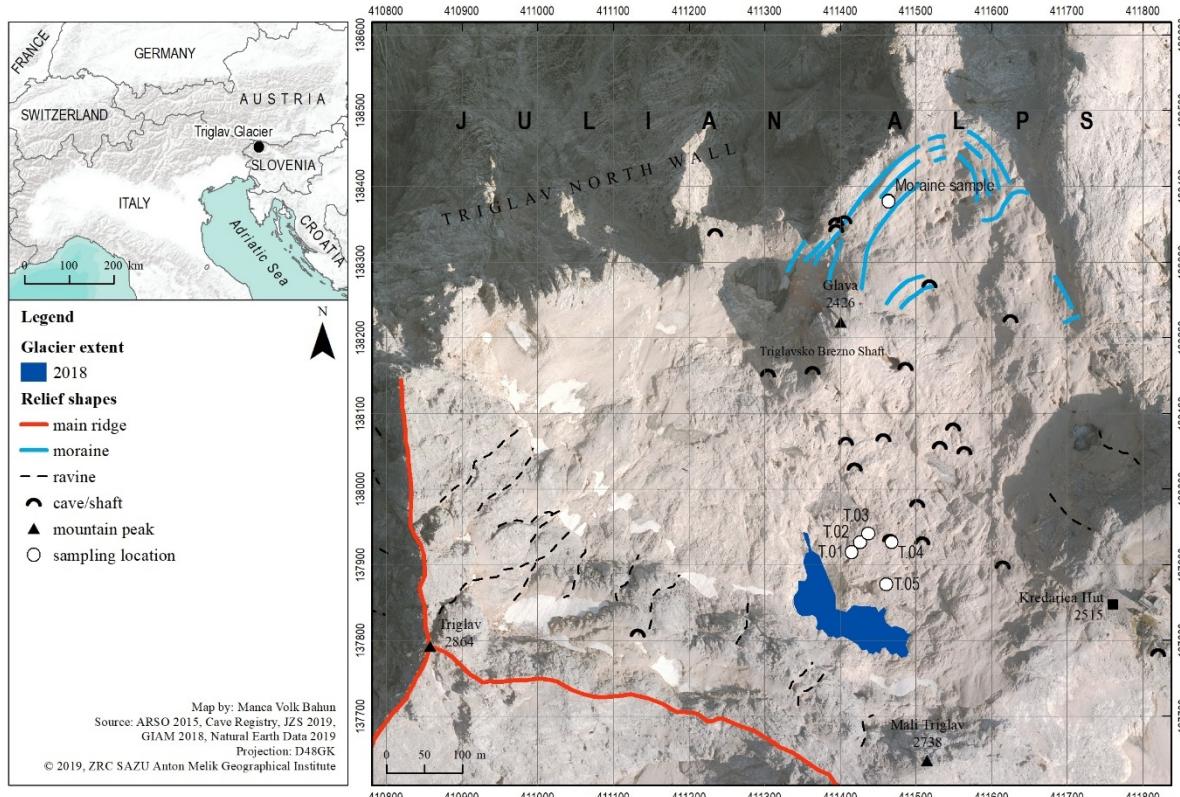
Supplementary Figure S3: General view of the subglacial carbonate location (© ZRC SAZU Anton Melik Geographical Institute archive).



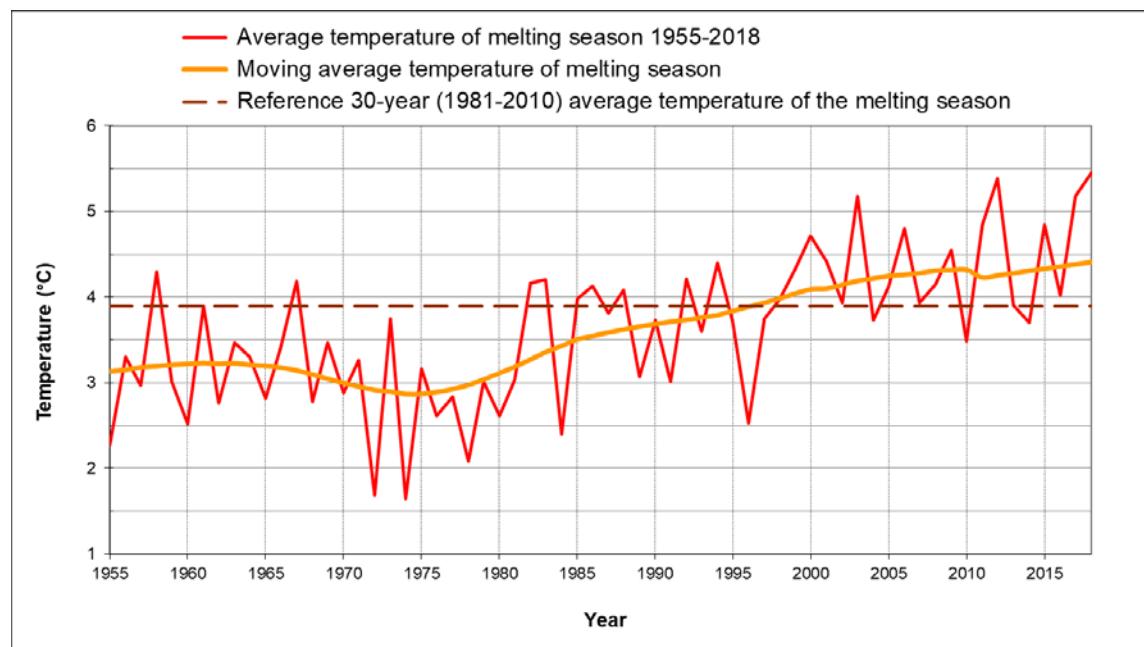
Supplementary Figure S4: Locality LiDAR DTM map of the Triglav Glacier extent and the sampling sites discussed in the text.



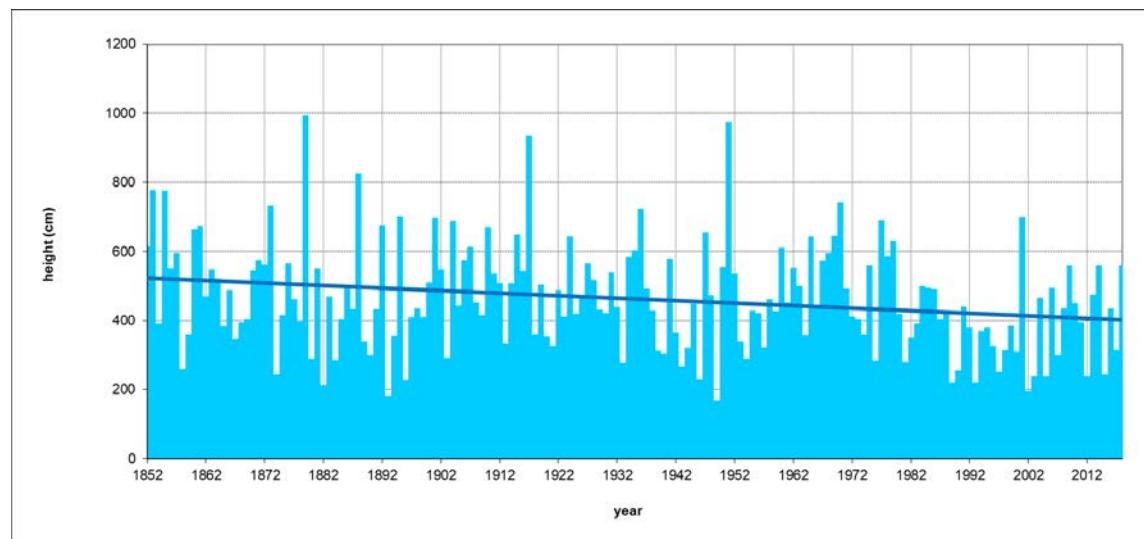
Supplementary Figure S5: Locality aerial map of the Triglav Glacier, the sampling sites discussed in the text, and general geomorphology.



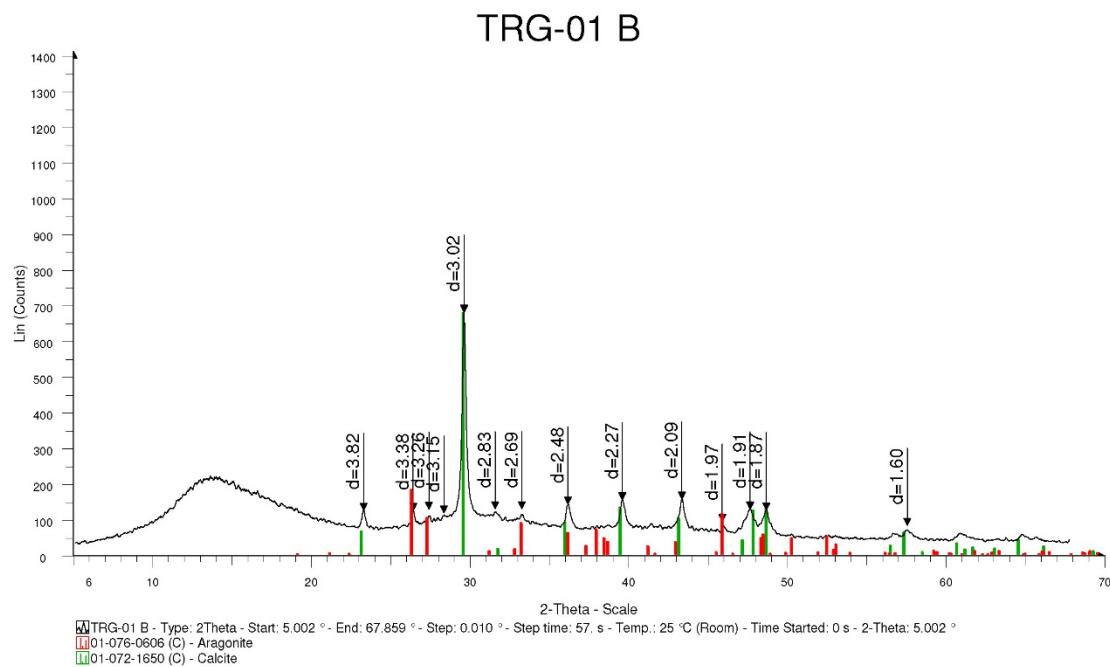
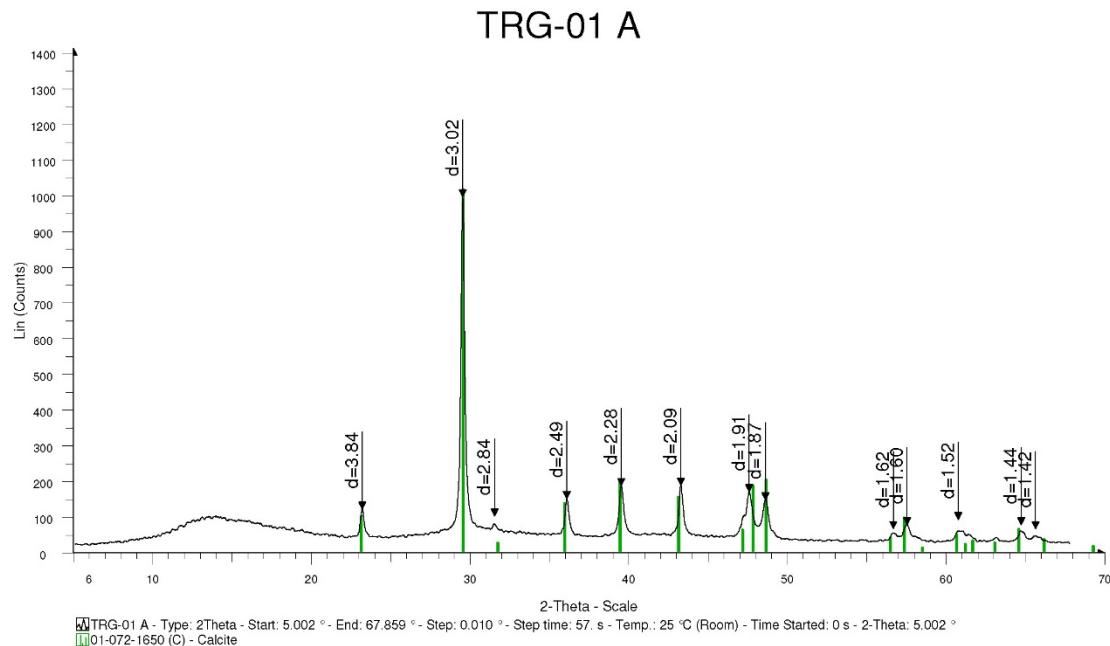
Supplementary Figure S6: Average melting season temperature (May – October) on meteorological station Kredarica (Fig. 1) between 1955 and 2018. Source: Slovenian Environment Agency.



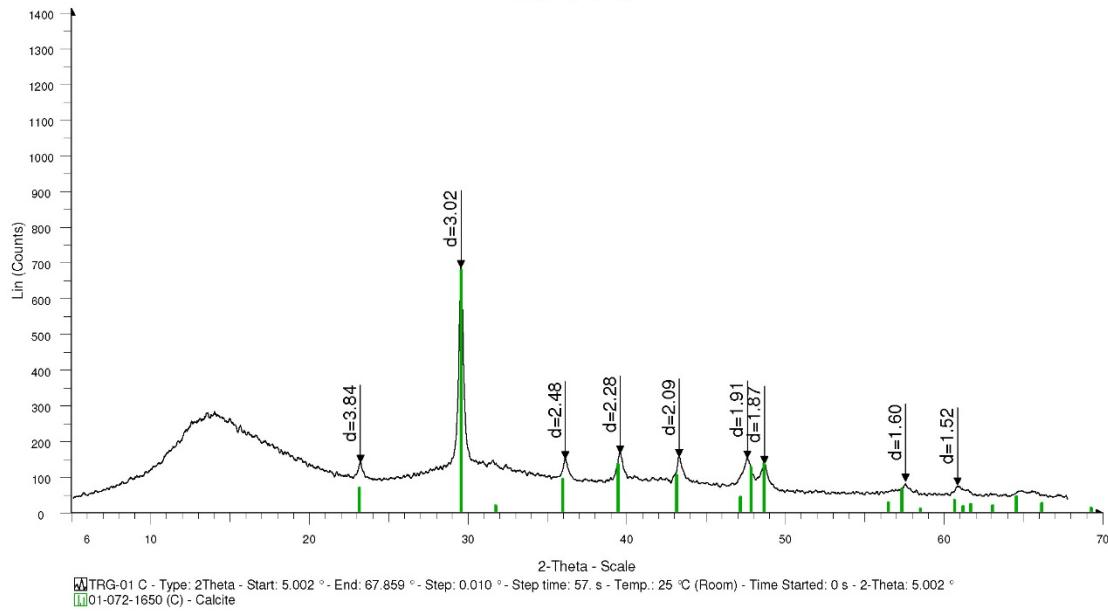
Supplementary Figure S7: Reconstruction of the highest seasonal snow height on the edge of the Triglav Glacier between 1852 and 2018. Updated from Gabrovec et al. 2014(Gabrovec et al., 2014). Source: Slovenian Environment Agency (from 1955 on), reconstruction/reanalysis before 1955 done by Jaka Ortar.



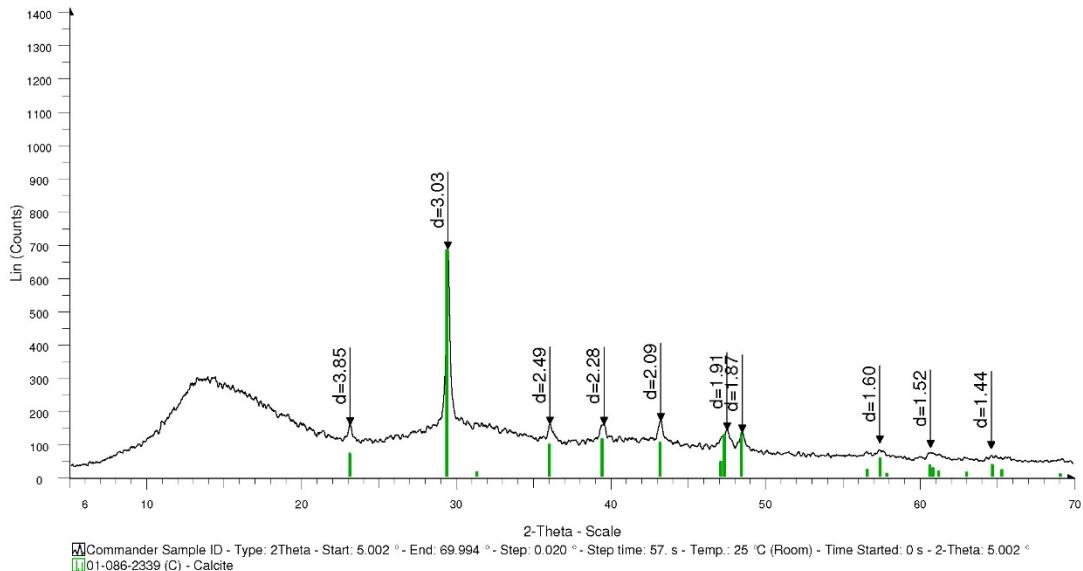
Supplementary Figure S8: XRD graphs



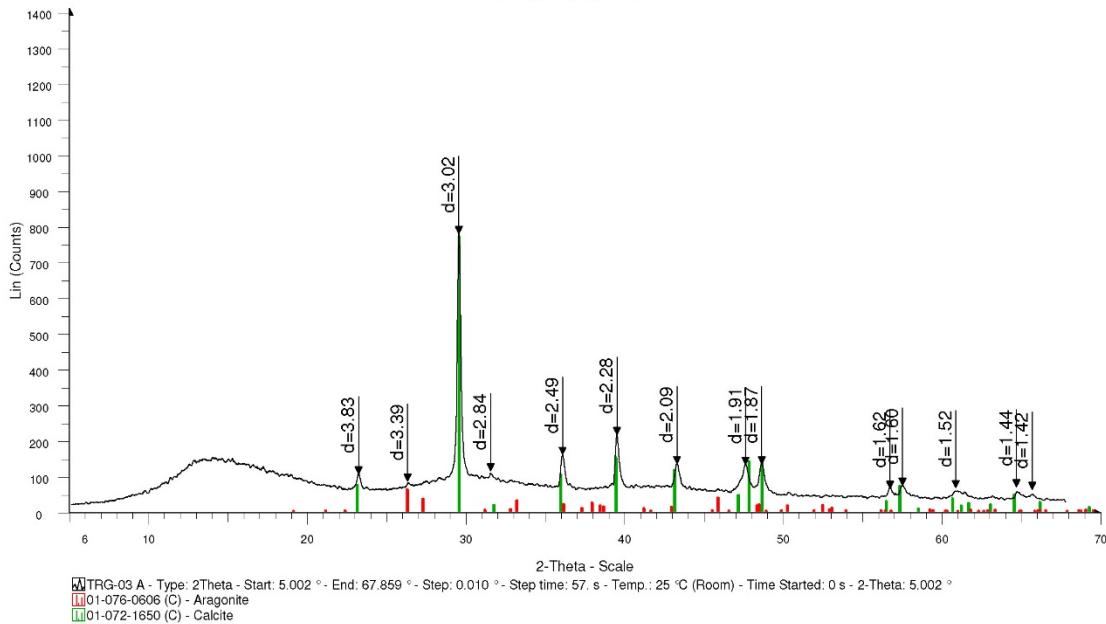
### TRG-01 C



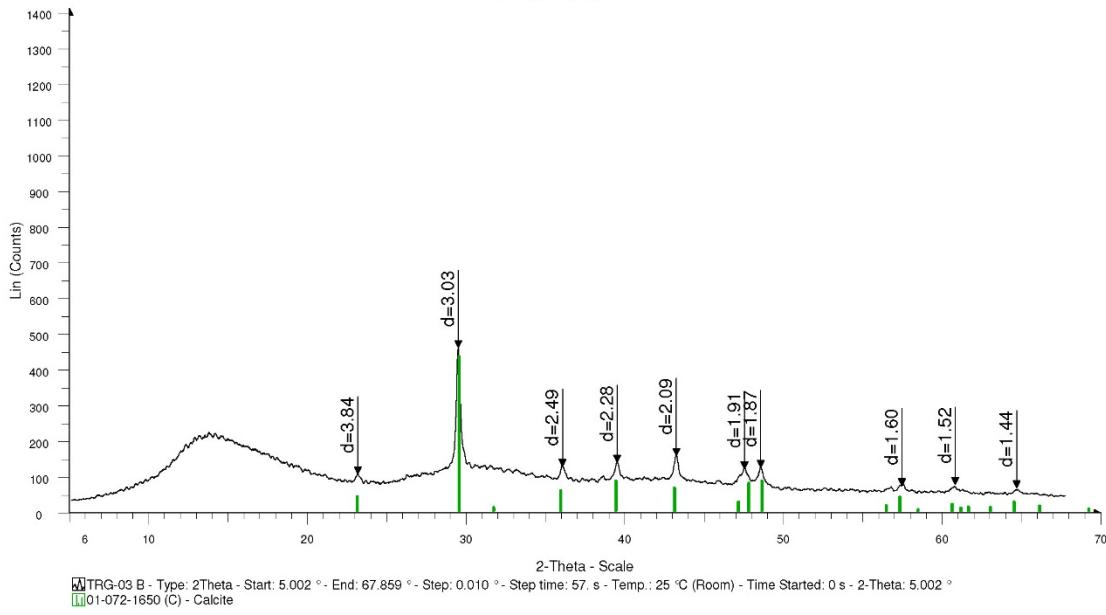
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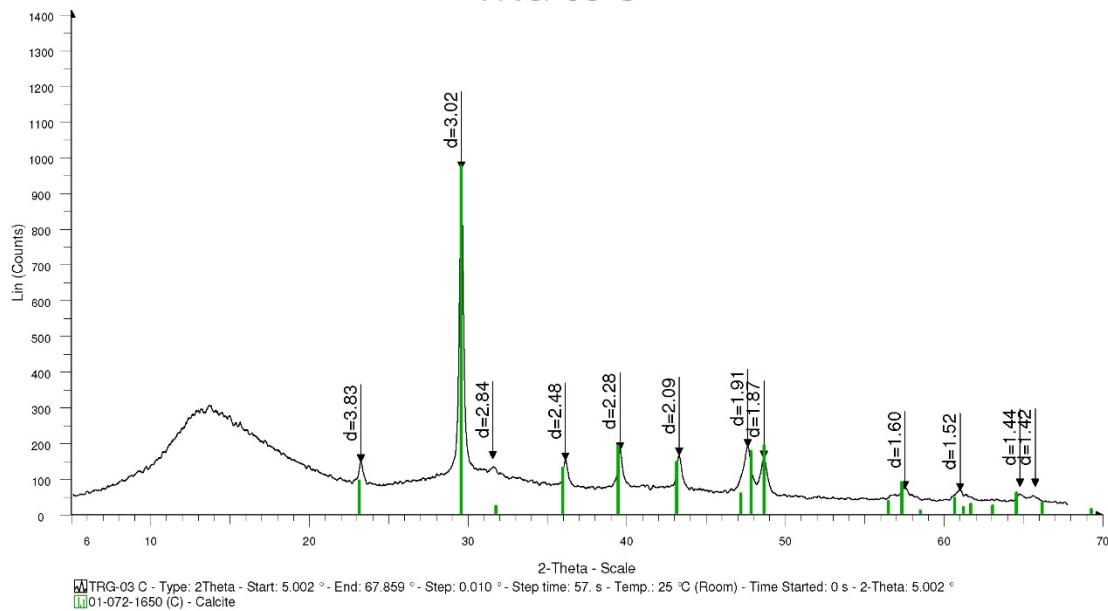
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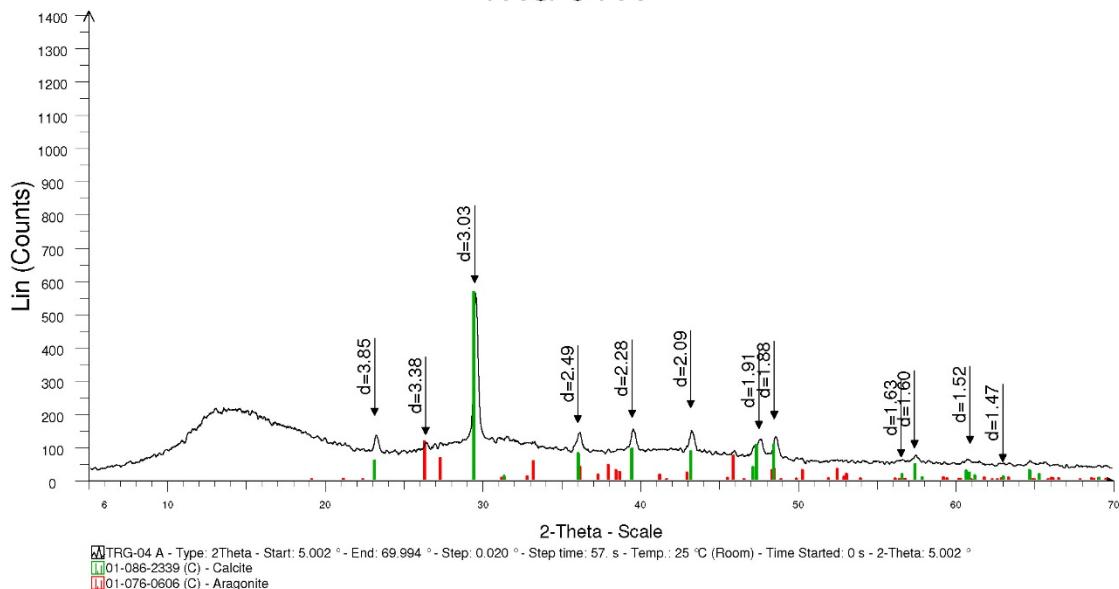
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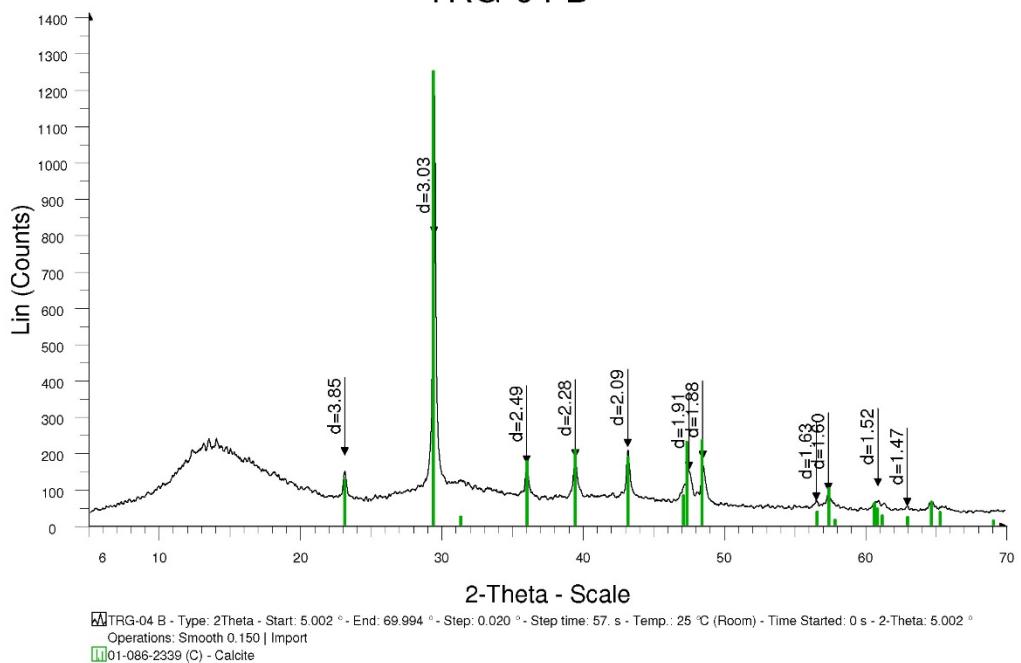
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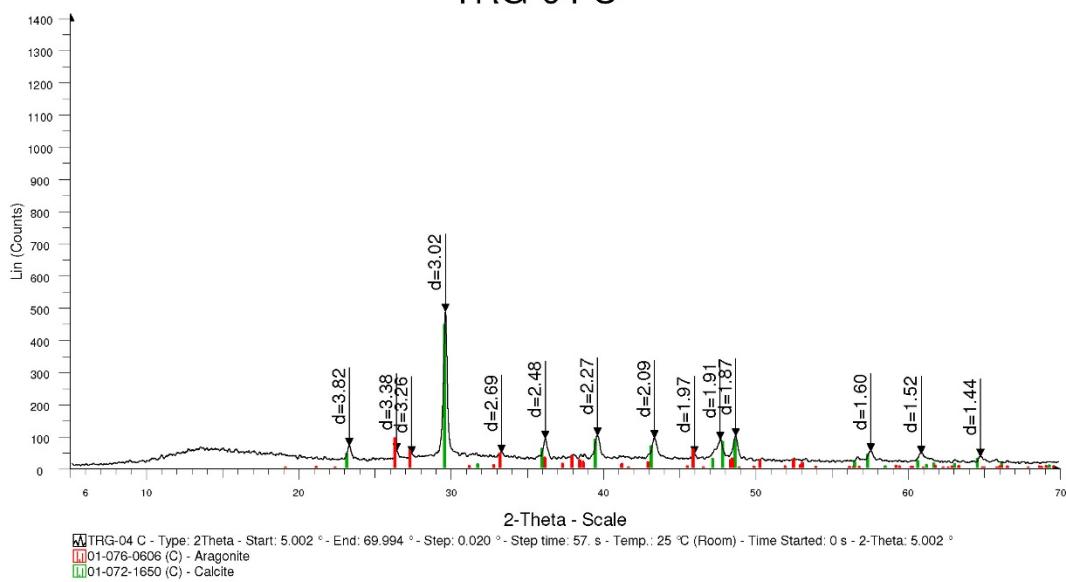
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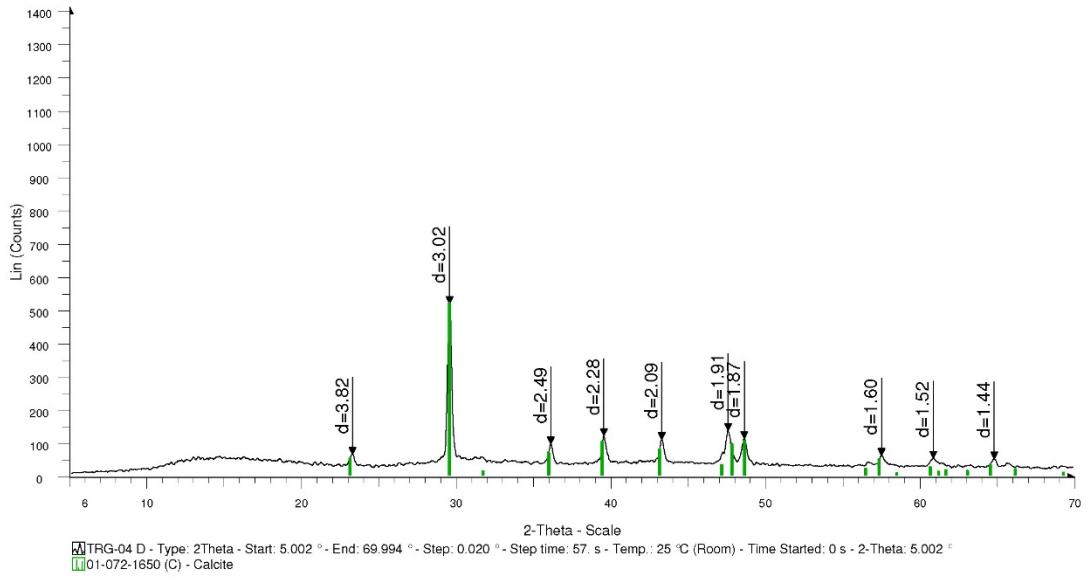
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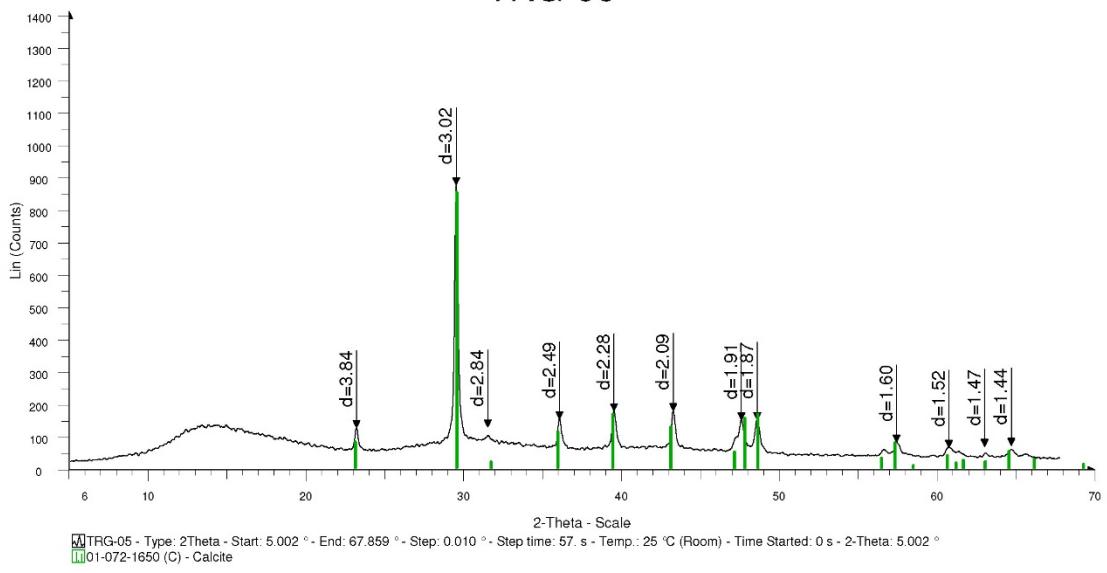
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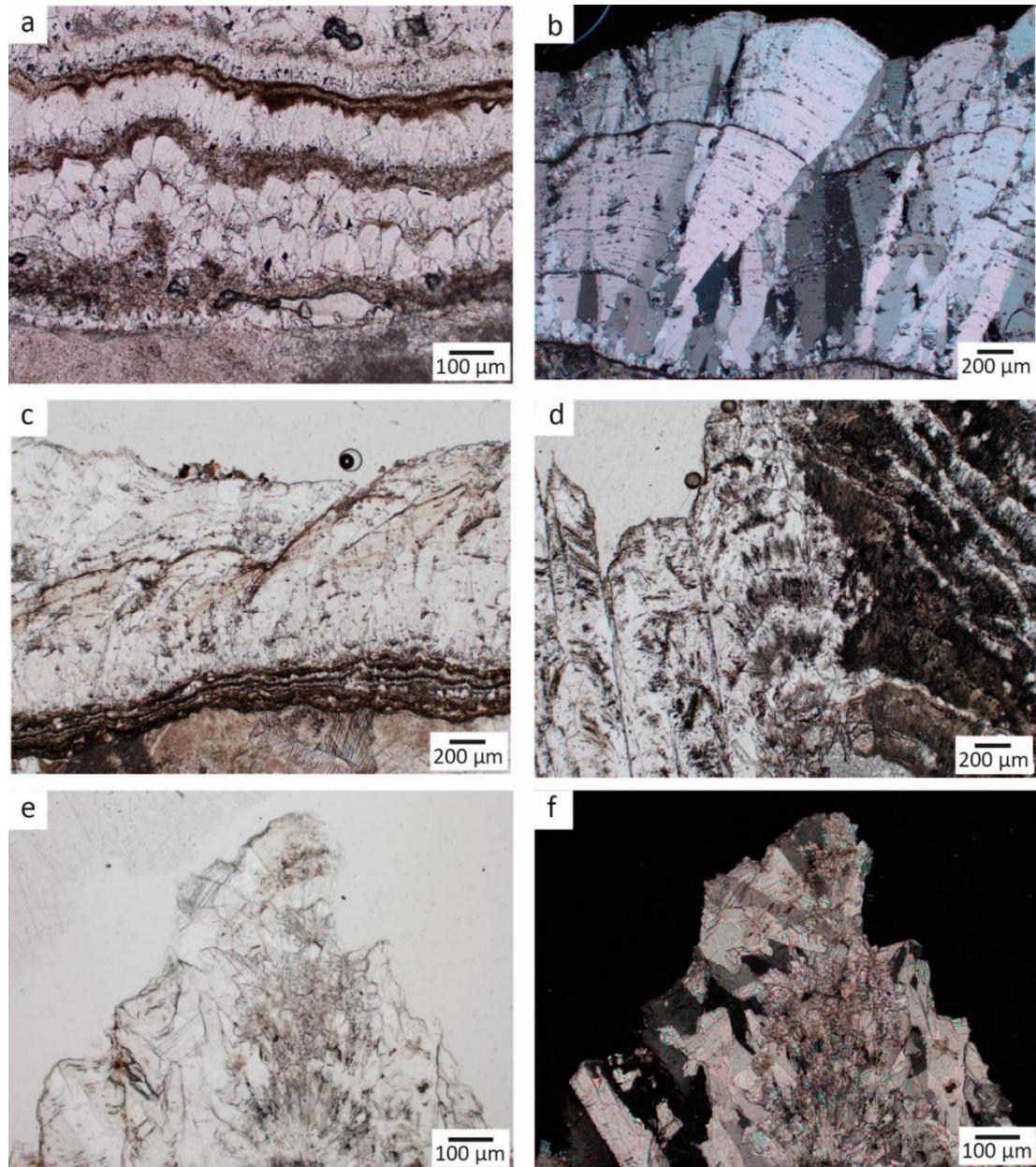
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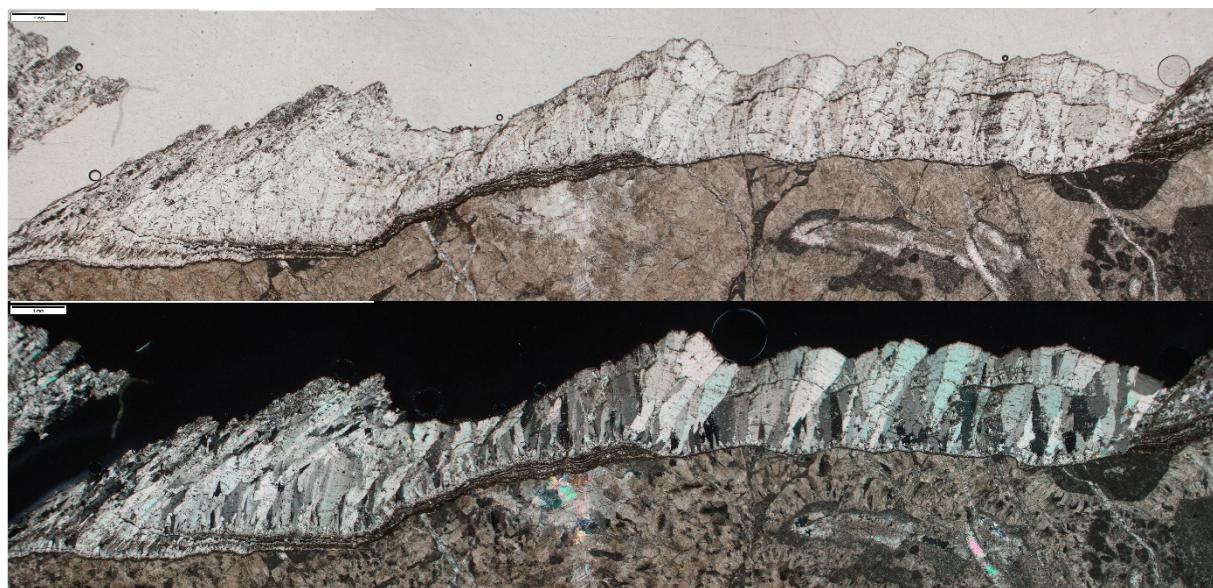
### TRG-05



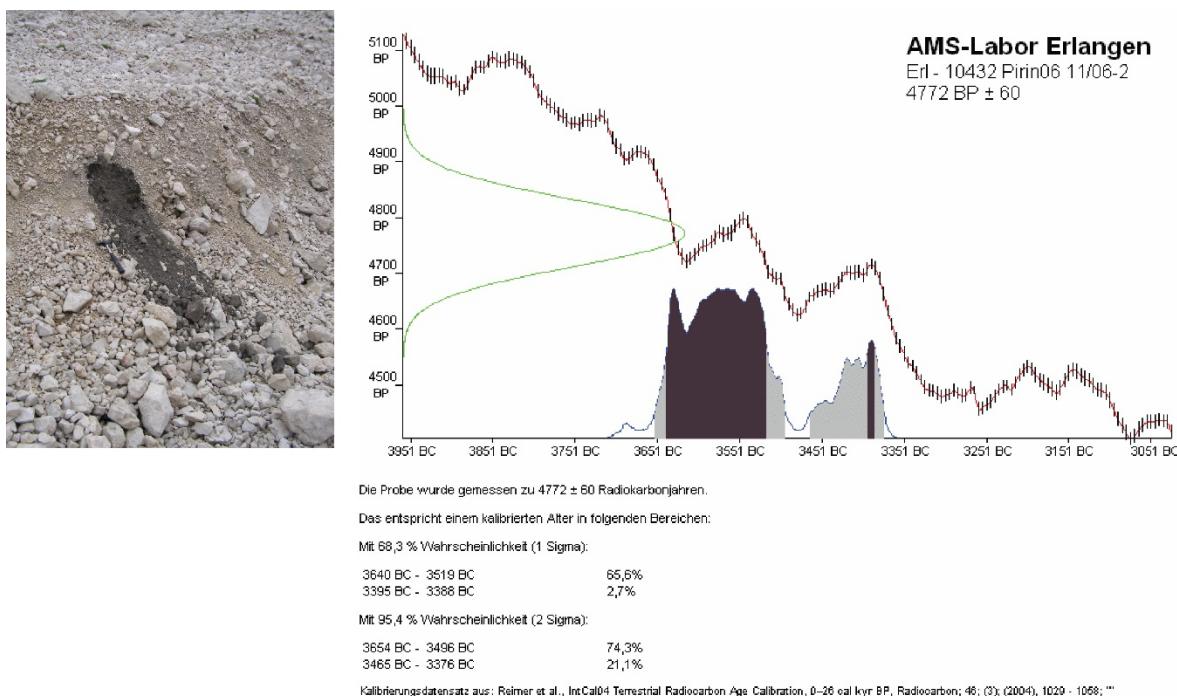
Supplementary Figure S9: a) Short columnar calcite crystals alternating with brown micritic bands constitute the first calcite precipitates over the host rock. Plane polarised light (PPL); b) columnar calcite crystals predominantly oriented towards the right (downslope). The growth of the crystal on the center crosscut the direction of growth of previous crystals. Crossed polarised light (XPL); c) oriented columnar calcite crystals displaying signs of dissolution. Occasionally, both crystal layers are separated by a thin, wavy micritic layer possibly indicating dissolution processes (PPL); d) aragonite-calcite layered textures (right) and secondary calcite crystals with aragonite relicts, some of them arranged in fans aligned forming a band (PPL); e) secondary calcite showing abundant aragonite relicts and very anhedral textures (PPL); f) same image as e, under crossed polarised light.



Supplementary Figure S10: The columnar and elongated columnar calcites oriented downslope (plane polarised light – above, and crossed polarised light – below)



Supplementary Figure S11: The photograph and  $^{14}\text{C}$  results of dated moraine organic matter by Karsten Grunewald and his team in the laboratory in Erlangen.



Supplementary Table S1:  $\delta^{13}\text{C}$  and  $\delta^{18}\text{O}$  stable isotope ratios.

<b>Sample</b>	<b><math>\delta^{13}\text{C}</math> (‰ PDB)</b>	<b><math>\delta^{18}\text{O}</math> (‰ PDB)</b>
T.01_a	2.41	-2.84
T.03_a	1.35	-4.71
T.03_b	0.51	-5.45
T.04_c	1.50	-3.96
T.05_a	0.98	-4.63

Supplementary Table S2: U-Th data for the subglacial calcite. Note: Ratios in parentheses are activity ratios calculated from the atomic ratios. Errors are at 2 $\delta$  level. The ages are calculated using the Isoplot 3.75 Program of Ludwig (2012)(Ludwig, 2012) with decay constants from Cheng et al. (2000)(Cheng et al., 2000). Corrected Ages were calculated assuming initial/detrital  $^{230}\text{Th}/^{232}\text{Th}$  activity ratio equal  $0.825 \pm 50\%$  (the bulk-Earth value, which is the most commonly used for initial/detrital  $^{230}\text{Th}$  corrections).

U-Th dating was carried out using a Nu Plasma multi-collector inductively-coupled plasma mass spectrometer (MC-ICP-MS) in the Radiogenic Isotope Facility (RIF) at the School of Earth and Environmental Sciences, The University of Queensland (UQ) following chemical treatment procedures and MC-ICP-MS analytical protocols described elsewhere(Clark et al., 2012; Clark et al., 2014; Zhao et al., 2009). Powdered or chipped sub-samples weighing 16–78 mg were spiked with a mixed  $^{229}\text{Th}-^{233}\text{U}$  tracer and then completely dissolved in concentrated  $\text{HNO}_3$ . After digestion, each sample was treated with  $\text{H}_2\text{O}_2$  to decompose trace amounts of organic matters (if any) and to facilitate complete sample-tracer homogenisation. U and Th were separated using conventional anion-exchange column chemistry using Bio-Rad AG 1-X8 resin. After stripping off the matrix from the column using double-distilled 7N  $\text{HNO}_3$  as eluent, 3 ml of a 2%  $\text{HNO}_3$  solution mixed with trace amount of HF was used to elute both U and Th into a 3.5-ml pre-cleaned test tube, ready for MC-ICP-MS analyses, without the need for further drying down and re-mixing. After column chemistry, the U-Th mixed solution was injected into the MC-ICP-MS through a DSN-100 desolvation nebuliser system with an uptake rate of around 0.07 ml per minute. U-Th isotopic ratio measurement was performed on the MC-ICP-MS using a detector configuration to allow simultaneous measurements of both U and Th isotopes(Clark et al., 2014; Zhou et al., 2011). The  $^{230}\text{Th}/^{238}\text{U}$  and  $^{234}\text{U}/^{238}\text{U}$  activity ratios of the samples were calculated using the decay constants given in(Cheng et al., 2000). The non-radiogenic  $^{230}\text{Th}$  was corrected using an assumed bulk-Earth atomic  $^{230}\text{Th}/^{232}\text{Th}$  ratio of  $4.4 \pm 2.2 \times 10^{-6}$ . U-Th ages were calculated using the Isoplot/Ex 3.75 Program(Ludwig, 2012).

Sample Name	Sample wt.(g)	U (ppm)	$\pm 2s$	$^{232}\text{Th}$ (ppb)	$\pm 2s$	$(^{230}\text{Th}/^{232}\text{Th})$	$\pm 2s$	$(^{230}\text{Th}/^{238}\text{U})$	$\pm 2s$	$(^{234}\text{U}/^{238}\text{U})$	$\pm 2s$	uncorr. $^{230}\text{Th}$ Age (ka)	$\pm 2s$	corr. $^{230}\text{Th}$ Age (ka)	$\pm 2s$	corr. Initial $(^{234}\text{U}/^{238}\text{U})$	$\pm 2s$
T.01_a1	0,06022	0,45584	0,00019	20,522	0,019	10,945	0,047	0,1624	0,0007	0,9787	0,0012	19,83	0,10	<b>18,45</b>	<b>0,70</b>	0,9773	0,0013
T.03_a1	0,07823	0,41066	0,00017	7,454	0,012	19,219	0,103	0,1150	0,0006	1,0053	0,0005	13,26	0,07	<b>12,72</b>	<b>0,28</b>	1,0055	0,0006
T.03_a2	0,02380	0,8933	0,0003	4,929	0,005	19,99	0,23	0,0363	0,0004	1,0062	0,0011	4,012	0,047	<b>3,849</b>	<b>0,094</b>	1,0063	0,0011
T.03_b1	0,01620	1,7737	0,0005	4,080	0,005	24,49	0,23	0,0186	0,0002	1,0079	0,0008	2,027	0,019	<b>1,959</b>	<b>0,039</b>	1,0079	0,0008
T.05	0,02151	1,3305	0,0004	68,59	0,11	12,20	0,05	0,2072	0,0008	1,0060	0,0010	25,15	0,11	<b>23,62</b>	<b>0,78</b>	1,0065	0,0011
HU-1 standard	0,03057	0,7511	0,0003	0,0660	0,0006	34680	342	1,0036	0,0026	1,0013	0,0011	<b>Secular equilibrium</b>					
HU-1 standard	0,03538	0,7542	0,0003	0,1994	0,0008	11439	53	0,9969	0,0024	1,0006	0,0009	<b>Secular equilibrium</b>					
YB-1 standard	0,09094	0,1293	0,0001	0,5889	0,0012	291,3	1,3	0,4371	0,0018	1,7452	0,0018	30,79	0,15	<b>30,71</b>	<b>0,15</b>	1,8136	0,0020
YB-1 standard	0,10259	0,1292	0,0001	0,6118	0,0012	278,6	1,3	0,4348	0,0019	1,7445	0,0020	30,62	0,15	<b>30,54</b>	<b>0,16</b>	1,8126	0,0022
SRM-960 standard	0,10570	5,5396	0,0046	0,01	0,000	1485	28	0,00071	0,00001	0,9636	0,0007	0,0805	0,0009	<b>0,0805</b>	<b>0,0009</b>	0,9636	0,0007
YB-1 speleothem std	0,10214	0,1253	0,0000	0,49	0,001	339,8	1,2	0,4343	0,0015	1,7492	0,0013	30,53	0,12	<b>30,46</b>	<b>0,12</b>	1,8174	0,0014

## REFERENCES:

- Cheng, H., Edwards, R. L., Hoff, J., Gallup, C. D., Richards, D. A., and Asmerom, Y.: The half-lives of uranium-234 and thorium-230, *Chemical Geology*, 169, 17-33, doi:10.1016/s0009-2541(99)00157-6, 2000.
- Clark, T. R., Zhao, J.-x., Feng, Y.-x., Done, T. J., Jupiter, S., Lough, J., and Pandolfi, J. M.: Spatial variability of initial  $^{230}\text{Th}/^{232}\text{Th}$  in modern Porites from the inshore region of the Great Barrier Reef, *Geochimica et Cosmochimica Acta*, 78, 99-118, doi:10.1016/j.gca.2011.11.032, 2012.
- Clark, T. R., Zhao, J.-x., Roff, G., Feng, Y.-x., Done, T. J., Nothdurft, L. D., and Pandolfi, J. M.: Discerning the timing and cause of historical mortality events in modern Porites from the Great Barrier Reef, *Geochimica et Cosmochimica Acta*, 138, 57-80, doi:10.1016/j.gca.2014.04.022, 2014.
- Gabrovec, M., Hrvatin, M., Komac, B., Ortar, J., Pavšek, M., Topole, M., Triglav Čekada, M., and Zorn, M.: *Triglavski ledenik*, Založba ZRC, Ljubljana, 2014.
- Ludwig, K. R.: User's manual for Isoplot version 3.75–4.15: a geochronological toolkit for Microsoft Excel, Berkeley Geochronological Center Special Publication, 5, 2012.
- Zhao, J.-x., Yu, K.-f., and Feng, Y.-x.: High-precision  $^{238}\text{U}$ - $^{234}\text{U}$ - $^{230}\text{Th}$  disequilibrium dating of the recent past: a review, *Quaternary Geochronology*, 4, 423-433, doi:10.1016/j.quageo.2009.01.012, 2009.
- Zhou, H., Zhao, J., Qing, W., Feng, Y., and Tang, J.: Speleothem-derived Asian summer monsoon variations in Central China, 54-46 ka, *Journal of Quaternary Science*, 26, 781-790, doi:10.1002/jqs.1506, 2011.