

1 **Intrusion, retention, and snowpack chemical effects from exhaust emissions at**  
2 **Concordia Station, Antarctica**

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12 **Abstract**

13 ~~Continuous-The chemistry measurements~~ of reactive gases ~~in the snowpack and above the snowpack~~  
14 ~~surface inside the snowpack and in the lower atmosphere was investigated, were conducted~~ at Concordia  
15 Station (Dome C), Antarctica, from December 2012 - January 2014. Measured species included ozone,  
16 nitrogen oxides, gaseous elemental mercury, and formaldehyde, for study of photochemical reactions,  
17 surface exchange, and the seasonal cycles and atmospheric chemistry of these gases. The experiment  
18 was installed  $\approx$  1 km from the station main infrastructure inside the station clean air sector and within  
19 the station electrical power grid boundary. Ambient air was sampled continuously from inlets mounted  
20 above the surface on a 10 m meteorological tower. In addition, snowpack air was collected at 30 cm  
21 intervals to 1.2 m depth from two manifolds that had both above and below surface sampling  
22 inlets.~~Air was sampled continuously from three inlets mounted above the surface on a 10 m~~  
23 ~~meteorological tower. In addition, snowpack air was collected in 30 cm intervals to 1.2 m depth from two~~  
24 ~~manifolds that had both above and below surface sampling inlets, as well as from (two) above and four~~  
25 ~~below the surface sampling inlets from within the snowpack.~~ Despite being in the clean air sector, over  
26 the course of the 1.2-year study, we observed on the order of ~~45-50~~ occasions when exhaust plumes from  
27 the camp, most notably from the power generation system, were transported to the study site.  
28 Continuous monitoring of nitrogen oxides (NO<sub>x</sub>) provided a measurement of a chemical tracer for exhaust  
29 plumes. Highly elevated levels of NO<sub>x</sub> (up to 1000 x background) and lowered ozone (down to  $\approx$ 50%),  
30 most likely from titration with nitric oxide, were measured in air from above and within the snowpack.  
31 Within 5-15 minutes from observing elevated pollutant levels above the snow, rapidly increasing and long-  
32 lasting concentration enhancements were measured in snowpack air. While pollution events typically  
33 lasted only a few minutes to an hour above the snow surface, elevated NO<sub>x</sub> levels were observed in the  
34 snowpack lasting from a few days to  $\approx$  one week. These observations add important new insight to the  
35 discussion if and how snow-photochemical experiments within reach of the power grid of polar research  
36 sites are possibly compromised by the snowpack being chemically influenced (contaminated) by gaseous

37 and particulate emissions from the research camp activities. This question is critical for evaluating if  
38 snowpack trace chemical measurements from within the camp boundaries are representative for the vast  
39 polar ice sheets.

40

## 41 Introduction

42 Research conducted during the past  $\approx 15$  years has revealed an active and remarkable spatial diversity  
43 of atmospheric oxidation chemistry in the polar lower atmosphere [Grannas et al., 2007]. Ozone plays a  
44 fundamental role in controlling the lifetime of many atmospheric trace gases directly and indirectly by  
45 modulating atmospheric OH. Unlike the episodic ozone depletion events observed at coastal sites, the  
46 opposite effect (i.e. ozone production) has been observed in the Antarctic interior [Crawford et al., 2001;  
47 Helmig et al., 2007b; Helmig et al., 2008a; Legrand et al., 2009; Legrand et al., 2016]. The discovery of  
48 ozone production chemistry in the remote and pristine Antarctic environment was rather surprising,  
49 because hitherto photochemical production in the lower atmosphere had exclusively been associated  
50 with polluted urban environments [Molina and Molina, 2004]. Photochemical production and snowpack  
51 emissions of nitric oxides ( $\text{NO}_x$ ) have been identified as underlying processes driving this chemistry.  $\text{NO}_x$   
52 has been shown to be formed from photochemical reactions in the snowpack [Honrath et al., 1999; Jones  
53 et al., 2000], with deposited nitrate constituting the reservoir of this chemistry.  $\text{NO}_x$  play a crucial role in  
54 snow photochemical reactivity [Murray et al., 2015].  $\text{NO}_x$  mixing ratios in interstitial air resulting from  
55 photochemical reactions can exceed those in the air above the snowpack by a factor of  $\approx 50$  [Van Dam et  
56 al., 2015].

57 This concentration gradient is driving  $\text{NO}_x$  emission fluxes out of the snowpack into the overlying  
58 atmosphere [Jones et al., 2001; Honrath et al., 2002], which, under stable atmospheric conditions, can  
59 cause large  $\text{NO}_x$  enhancements in the atmospheric surface layer [Helmig et al., 2008b; Neff et al., 2008;  
60 Frey et al., 2011; Frey et al., 2013], and in the presence of solar irradiance trigger photochemical ozone  
61 production, with resulting peak ozone levels that can be double those in the boundary layer [Crawford et  
62 al., 2001; Helmig et al., 2008a; Legrand et al., 2016]. Experiments on reactive nitrogen chemistry  
63 investigating this rather unexpected ozone production chemistry have built on a variety of atmospheric  
64 research strategies, including snowpack air sampling [Dibb et al., 2002; Jacobi et al., 2004; Helmig et al.,  
65 2007a; Van Dam et al., 2015], snow chambers [Dibb et al., 2002], snow chemical analyses [Dassau et al.,  
66 2002; Dibb et al., 2007b; France et al., 2011; Erbland et al., 2013], atmospheric monitoring [Frey et al.,  
67 2011; Kramer et al., 2015; Legrand et al., 2016], surface fluxes [Jones et al., 2001; Honrath et al., 2002;  
68 Frey et al., 2011; Frey et al., 2015], and boundary layer vertical profiling [Helmig et al., 2008a; Frey et al.,  
69 2015].

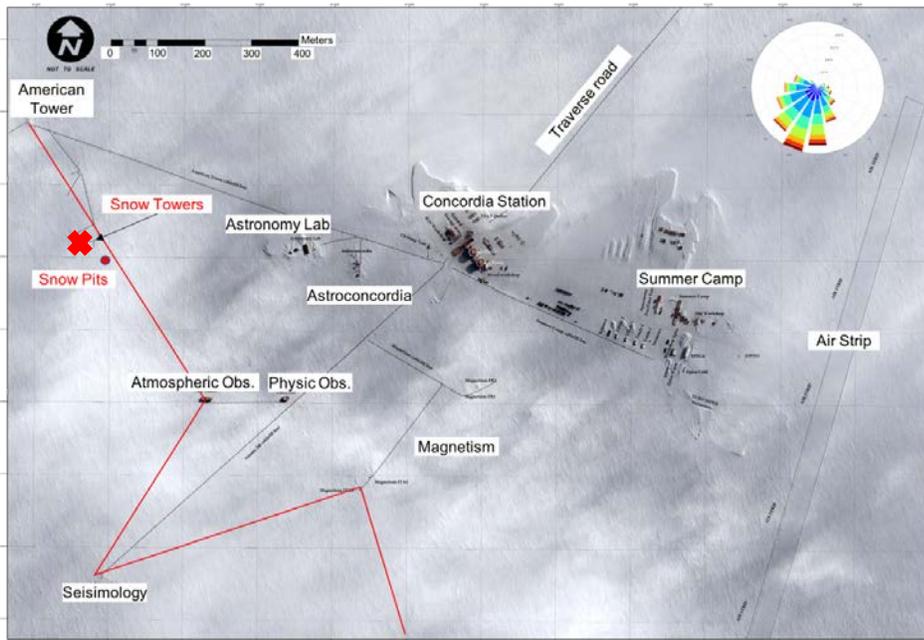
70 Most of these studies rely on observations from dedicated campaigns at research stations, including  
71 photochemistry campaigns at Summit, Greenland [Dibb et al., 2007a], the Antarctic Tropospheric  
72 Chemistry Investigation [ANTCI; Eisele and Davis, 2008] at the South Pole, the Chemistry of the Antarctic  
73 Boundary Layer and the Interface with Snow (CHABLIS) experiment at Halley [Jones et al., 2008], and the  
74 Oxidant Production over Antarctic Land and its Export (OPALE) campaign at Concordia Station [Preunkert  
75 et al., 2012]. A common limitation of these studies is that experiments were conducted in proximity to  
76 ~~large~~ research stations, where use of fuel-powered engines in generators and vehicles cause exhaust

77 emissions with highly elevated concentrations of particulates and gases, particularly of volatile organic  
78 compounds (VOCs) and NO<sub>x</sub>. -A critical question is if and how this pollution, and possibly secondary  
79 products formed during the atmospheric transport and deposition, impact the snow chemical position  
80 and reactivity, and potentially the findings from this aforementioned literature. -This is of particular  
81 importance for oxidized nitrogen species. This experiment-study yielded for the first time a year-long  
82 record of NO<sub>x</sub> and O<sub>3</sub> in an Antarctic snowpack at Concordia and the atmosphere above it. This experiment  
83 also gave us the opportunity to study and evaluate occurrences of pollution episodes, using the NO<sub>x</sub>  
84 monitoring as a sensitive chemical tracer for identification of exhaust plumes.

#### 85 **Methods**

86 Location: This experiment was conducted at the French/Italian Antarctic research station Concordia,  
87 located at the Dome Circe or Dome Charlie (Dome C, 75.10°S/123.35°E 3233 m asl, mean temperature -  
88 55°C). An experimental site was established at the border of the clean air sector, approximately ≈1 km to  
89 the west of the station common buildings (Figure 1). The clean air sector is located in the opposite  
90 direction of the prevailing wind direction. The site consisted of a 8 m x 2 m x 2.5 m underground laboratory

91



**Figure 1:** Satellite image of research station Concordia with location of the snow photochemistry experiment indicated by the red text and cross marker. Its location was ~1 km west of the station main buildings and power generation plant. The map also indicates the location where the snow pits were collected, the border of the clean air sector (which is west of the red line), and a wind rose for a full year of wind data collected from the met tower that was adjacent to the snow tower manifolds. Satellite image of research station Concordia with location of the snow photochemistry experiment indicated by the red cross and text. Its location was ~1 km west of the station main buildings and power generation plant.

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93 positioned at the border of the clean air area, a 10 m tall meteorological tower, and two snow air sampling

94 manifolds placed 15 m into clean snow ('snow tower') for sampling the atmosphere and the snow

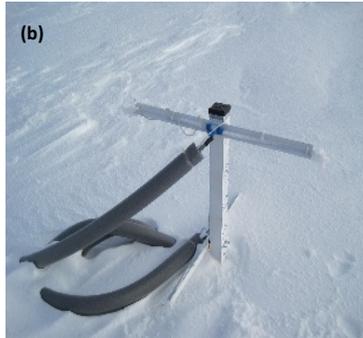
95 interstitial air (Figure 2). The installation was in late November 2012 with continuous monitoring

96 conducted until January 2014 (14 months).

97 **Meteorological Tower:** A 10 m meteorological tower (Figure 2a) was equipped with two sonic

98 anemometers for atmospheric turbulence measurements, and three gas sampling inlets (0.5 m, 2 m, 10

99 m) with sampling lines inside a heated conduit running to the laboratory. The upper inlet was attached



**Figure 2:** (a) Meteorological tower with the station infrastructure in the background. The wooden box to the left is the entry hatch to the underground laboratory. Air sampling inlets were located at 0.5 m, 2 m and 10 m above the surface on the tower. Two snowpack air sampling towers were located approximately 7 m to the left and 10 m to the right of the base of the meteorological tower. (b) One of the two snow pack air sampling manifold (snow tower), with one pair of inlets right on the snow surface, and one inlet pair at ~30 cm height. Four more equivalent sampling inlet pairs are below the snow surface at 30 cm depth intervals extending to a maximum depth of 1.2 m- [see Figure 1 in -Van Dam et al. \[2015\] for a schematic of a similar installation at Summit, Greenland\).](#)

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100  
101 to a manual pulley allowing the ~~upper~~ inlet to be lowered for side by side sampling of both inlets for  
102 tracking and correcting sampling inlet/lines biases.

103 **Snowpack Air Sampling:** Two identical multi-inlet snow sampling manifolds ('snow tower') for collection  
104 of interstitial and ambient air ~~at six ~20 cm vertical intervals~~ were constructed, with a similar design to  
105 that described by [Seok et al., 2009] (Figure 2b). The snow tower consisted of a vertical post of square  
106 aluminum alloy (3.8 x 3.8 cm) with eight 60 cm long cross arms spaced vertically every ~~30-20~~ cm. Each of  
107 the cross bars supported a pair of sampling inlets. The inlets were fitted with 25 mm Acrodisc hydrophobic  
108 polytetrafluoroethylene (PTFE) syringe filters (Pall Life Sciences, Ann Arbor, Michigan, USA) to prevent  
109 snow and ice crystals from being pulled into the sampling line. For the installation, a snow pit was dug  
110 and the ~~inlets were lines~~ inserted horizontally into the clean untouched walls of the pit. ~~The hole was~~  
111 ~~then loosely refilled with the excavated snow, re-establishing the stratigraphy as much as possible.~~  
112 ~~Blowing snow then refilled any remaining gaps within the following 2-3 days. The snow tower was kept~~  
113 ~~in place after the campaign, so we have no data on the potential changes in porosity and air flow dynamics~~  
114 ~~that resulted from the installation and subsequent changes in the snow morphology as the snowpack re-~~  
115 ~~equilibrated. The hole was then loosely refilled with the excavated snow, respecting the stratigraphy as~~  
116 ~~much as possible. Blowing snow then refilled any remaining gaps within the following 2-3 days. The snow~~  
117 ~~tower was kept in place after the campaign, so we have no data on the potential changes in porosity and~~  
118 ~~air flow dynamics that resulted from the snow tower installation and subsequent changes in the snow~~  
119 ~~morphology as the snowpack re equilibrated. The hole was then refilled with the excavated snow,~~  
120 ~~respecting the stratigraphy as much as possible.~~ Insulated and heated sampling lines connected the  
121 sampling inlets to the chemical analyzers in the underground laboratory. All sampling lines were of 0.64  
122 cm o.d. x 30 m long pre-conditioned PFA tubing, except the lines to the gradient inlets on the

123 meteorological tower, which were of 0.78 cm o.d. because they were continuously pumped to maintain  
124 a flow of at least 2 l min<sup>-1</sup>. Air was pulled through the snow tower sampling lines by the combined flow of  
125 the gas analyzers (ozone monitor at 1 l min<sup>-1</sup>, a gaseous elemental mercury (GEM) analyzer at 1 l min<sup>-1</sup>,  
126 NO<sub>x</sub> monitor at 1 l min<sup>-1</sup>). A maximum of two monitors sampled from the snow tower inlets together at  
127 a given time to limit the maximum snow air sampling. Since each line connected to a pair of inlets at equal  
128 height, the effective flow through each inlet was a maximum of  $\approx 1 \text{ l min}^{-1}$ . Each height was sampled for  
129 10 min every 2 hours, resulting in an approximately volume of a sphere with a radius of 25 cm around  
130 each inlet every 2 hours. ~~Sampling from the two snow towers was alternated every 24 hr.~~ Sampling  
131 from the two snow towers was alternated every 24 hours. Each sampling inlet had a thermocouple  
132 wire attached for monitoring of the snowpack temperature gradient.

133 Ozone Measurements: Ozone was measured with a Thermo Environmental (TEI) 49i UV absorption  
134 monitor that was calibrated against a NOAA Global Monitoring Division reference standard before field  
135 shipment.

136 NO<sub>x</sub> Measurements: Nitrogen oxides were monitored with a TEI chemiluminescence analyzer (TEI 42C-  
137 TL). The TEI 42C-TL has two channels. The first channel measures NO via NO + O<sub>3</sub> chemiluminescence.  
138 The second channel measures total nitrogen oxides (NO<sub>x</sub> = NO + NO<sub>2</sub>) by redirecting air through a heated  
139 (325°C) molybdenum converter, which causes NO<sub>2</sub>—including other oxidized nitrogen compounds—to be  
140 converted to NO. NO<sub>2</sub> is then determined by subtracting NO, obtained from the first channel, from the  
141 resulting NO<sub>x</sub> signal. There are a number of other oxidized nitrogen species that can contribute to the  
142 NO<sub>2</sub> measurement [Steinbacher *et al.*, 2007]. The error in the NO<sub>2</sub> measurement increases with rising  
143 levels of interfering gases such as nitrous acid (HONO), peroxyacetyl nitrate (PAN), and alkyl nitrates that  
144 contribute to the NO<sub>2</sub>-mode signal. Consequently, NO<sub>2</sub> concentrations obtained with the TEI 42C-TL  
145 represent an estimate for the sum of these oxidized nitrogen species. Field calibrations were conducted  
146 with a NIST-traceable 1 ppm NO in N<sub>2</sub> gas standard (Scott-Marine, Inc., Riverside, CA, USA) that was  
147 dynamically diluted to low ppb mixing ratios ~~with NO<sub>x</sub>-scrubbed ambient air.~~ We did not bring a zero air  
148 compressed gas cylinder to Concordia. Instead, a low NO<sub>x</sub> dilution gas was prepared by pumping ambient  
149 air (which had significantly lower NO<sub>x</sub> levels than snowpack air) through a cartridge filled with  $\approx 1 \text{ dm}^3$  of  
150 granular Chemisorbent (Purafill, Doraville, GA). Calibration ranges were from 0.1-25 ppb, and the  
151 instrument response was linear within this range. Intercept values of the linear regression, and zero  
152 values from sampling of the scrubbed air, were below 0.1 ppb.

153 Snow Sampling and NO<sub>3</sub><sup>-</sup> Determination:

154 The snow pit NO<sub>3</sub><sup>-</sup> data stem from sampling that was done at and near Concordia between January 2009  
155 and December 2010, and at  $\approx 3 \text{ m}$  distance from snow tower 2 in January 2014. Snow was collected in  
156 pre cleaned 50 ml centrifuge tubes inserted directly on a newly scraped wall of ~~the a~~ snow pit. Nitrate  
157 concentration in snow samples wasere measured directly in the field, at the wet chemistry laboratory of  
158 Concordia station. Each sample was melted at room temperature<sub>s</sub> and NO<sub>3</sub><sup>-</sup> concentrations were  
159 determined using a colorimetric method employed routinely at Concordia [Frey *et al.*, 2009].

## 160 Results and Discussion

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161 Results of the year-round snowpack and ambient monitoring, including interpretations on photochemistry  
162 will be presented elsewhere [Helmig et al., 2019]. Here, we primarily focus on occurrences of pollution  
163 transport to the sampling site and its penetration into the snowpack.

164 **Figure 3** shows a photograph of the station main buildings. The power plant is adjacent to the two column  
165 structure. Approximately 300 m<sup>3</sup> of Special Antarctic Blend (SAB) diesel fuel are burned in the plant for  
166 electricity and heat generation per year. The exhaust plume from the 5-m high stack of the power plant  
167 can be seen in the picture, blowing towards the west. Due to the typical strong stratification and stability  
168 of the atmosphere near the surface, the plume does not rise far above the stack height, but instead gets  
169 transported horizontally at a height of ≈ 5-10 m above the snow surface. This is a typical exhaust plume  
170 dispersion behavior for a cold regions environment, seen at many other polar research stations. The  
171 plume typically does not hit the surface within the immediate distance of the stack location. Depending  
172 on the actual turbulent mixing conditions, it may take several hundreds of meters before the stack  
173 emissions are encountered right at the surface.

174 Of the gases monitored in this experiment, NO<sub>x</sub> were the most sensitive tracer for pollution impact. We  
175 chose to concentrate on NO<sub>x</sub> as the total of NO+NO<sub>2</sub>, as this is a more representative indicator for the  
176 total amount of oxidized nitrogen, whereas NO would only indicate a fraction. Further, the fractionation  
177 between NO and NO<sub>x</sub> is sensitive to other gases (such as ozone) and residence time and snowpack depth.  
178 Therefore, interpretation of NO data is more ambiguous. NO<sub>x</sub> in ambient air at Concordia remained well  
179 below 1 ppb during background conditions year-round [Helmig et al., 2019], in agreement with  
180 observations from prior shorter campaign NO<sub>x</sub> measurement at Concordia

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**Figure 3:** Photograph of ~~Dome C~~ Concordia S-station illustrating the dispersion of the exhaust plume from the  
electrical power generating plant during conditions with a strong surface temperature inversion. The plume  
is ~~dissipates~~ blown towards the west in the direction of the experimental site. This is a typical situation for a  
contamination event.

181

182 [Frey et al., 2011; Frey et al., 2013; Frey et al., 2015]. We did not observe highly elevated NO levels in the  
183 surface layer outside of pollution events, as what has been observed at South Pole [Helmig et al., 2008b;  
184 Neff et al., 2008]. During pollution events, the 1 ppb<sub>s</sub> threshold was exceeded very quickly in  
185 measurements taken from the meteorological tower and from above-surface snow tower inlets, with  
186 resulting NO<sub>x</sub> mixing ratios rising to as high as close to 200 ppb, representing an up to 1000-fold  
187 enhancement over background conditions (Figure 4).

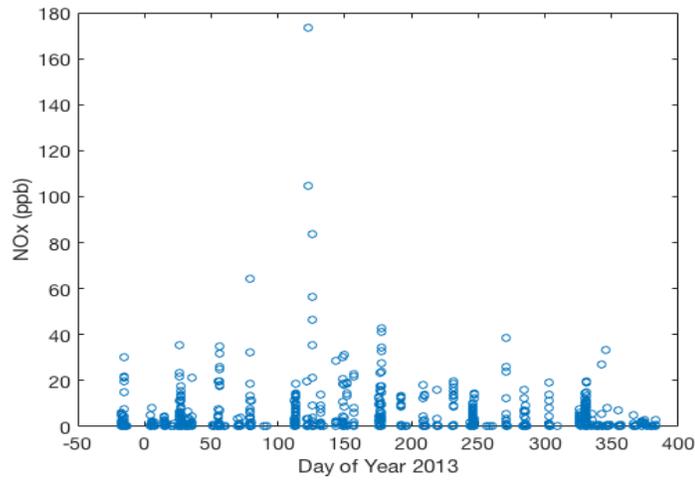
188 During the course of the ~~≈~~14 month study, a total of ~~≈~~50 pollution events were observed, although  
189 some events overlapped (Figure 4). Most of these occurrences were relatively short, with elevated mixing  
190 ratios above the snowpack lasting from minutes to a few hours. We counted 15 events in total when  
191 there was a spike in the above surface NO<sub>x</sub> measurements followed by an increase of at least 1 ppb of NO<sub>x</sub>  
192 in snowpack air. It took up to in excess of 7 days for NO<sub>x</sub> in the snowpack air to return to pre-event  
193 levels. Within the snowpack, NO<sub>x</sub> mixing ratios remained elevated for 1-5 days after the event. Integrated  
194 over the entire campaign, pollution episodes constituted < 2.0 % of the measurements above the  
195 snowpack, and <10% of the measurements within the snowpack. The correlation analyses of pollution  
196 occurrences with wind direction clearly defines the direction of the transport. The predominant wind  
197 direction sector at Concordia is southeast to northwest (Figure 5a), with southeasterly winds having the  
198 overall largest share. NO<sub>x</sub> levels were consistently well below 1 ppb when winds were from east to  
199 northeast. The sector with pollution transport is rather well defined, with wind directions covering  
200 approximately 45-120° (Figure 5b). These sectors perfectly line up with the upwind direction of the station  
201 power plant (Figure 1), clearly identifying the plant as the source of these pollution occurrences.

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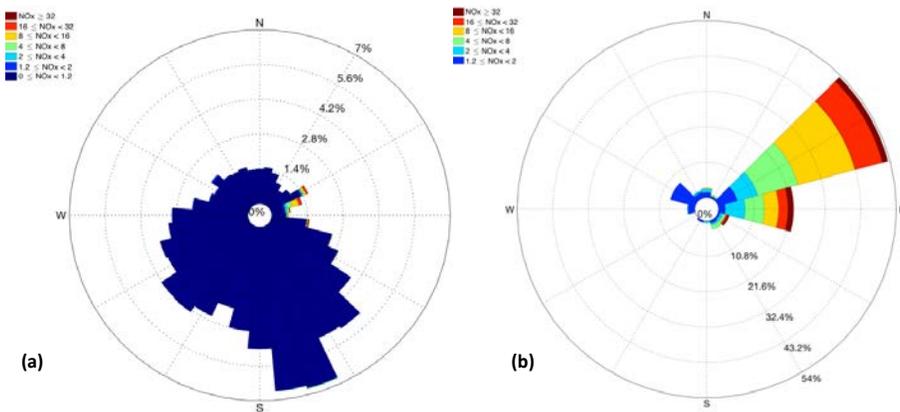
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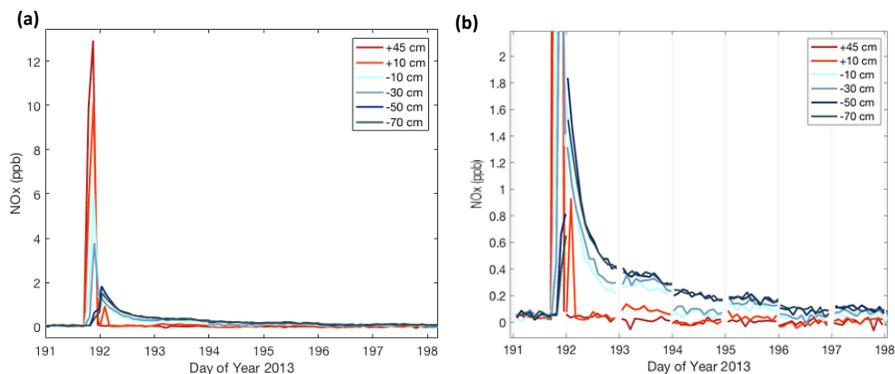
**Figure 4:** NO<sub>x</sub> measured from the above surface inlets on the two snow towers, and from all inlets on the meteorological tower. Circles represent 10-min averaged data. Plotted are data that were extracted from occurrences when NO<sub>x</sub> was above 1 ppb, well above the background mixing ratio, and any time when the wind direction was from the polluted wind sector.

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204



**Figure 5:** (a) Concentration wind rose with the relative frequency of NO<sub>x</sub> mixing ratio data from the above surface inlets of the two snow towers segregated by 10° sectors for the full year of observation data. This panel shows all data. (b) The same analysis, with wind direction data binned in 20° sectors for events when NO<sub>x</sub> in ambient air exceeded 1.2 ppb.

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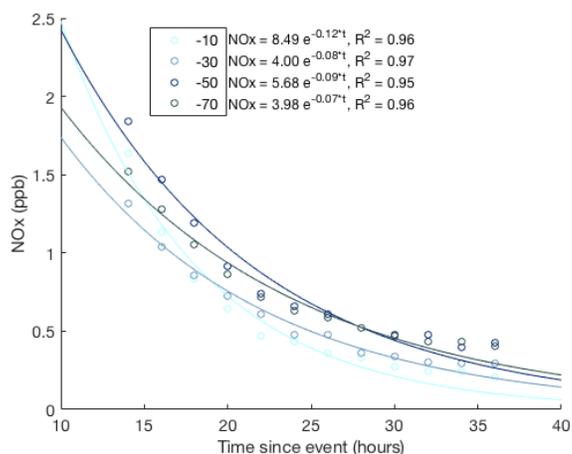


**Figure 6.** Combined measurements from the two snow towers capturing a pollution event at Concordia during the middle of the winter (Day of Year 191 = July 11). Plotted time series traces correspond to the sampling heights indicated in the legend, with positive numbers giving the height above the snow surface, and negative numbers the depth below the snow surface. The right graph (b) is an enlargement of the data shown in the left (a) with the transitions between the connecting lines between two snow tower measurements removed to show the level of agreement in the data from the two sampling manifolds. The sampling switched between the two snow towers every 24 hours, leading to some abrupt shifts in the snowpack NO<sub>x</sub> measurements from within the snowpack.

206 One of these elevated  $\text{NO}_x/\text{NO}_x$  events is further investigated in Figure 6. Here, we show the measurements  
207 from six inlets on the snow tower over a one-week period. The sampling of a polluted plume is first  
208 observed in the two above surface inlets (orange/red colored data; +10 and +45 cm), by the sudden  
209 increase of  $\text{NO}_x$  from well below 1 ppb to a mixing ratio of  $\approx 13$  ppb. This spike in  $\text{NO}_x$  lasted for  $\approx 3$   
210 hours. After that time,  $\text{NO}_x$  in air sampled above the surface dropped very quickly and equilibrated to  
211 prior mixing ratios within less than 0.5 hours.

212  
213 A much different behavior was found in the air sampled from within the snowpack, indicated by the  
214 data in the blueish colors. The onset of the pollution signal is delayed, by 1-3 hours, with progressively  
215 later times towards deeper in the snowpack. Maximum mixing ratios that are reached in the snowpack  
216 are lower, i.e. 10-50% of those that were measured above the surface, with mixing ratio maxima becoming  
217 progressively smaller with increasing depth. The most remarkable difference between the above and  
218 below surface measurements is the longer residence time of the pollution signal in the snowpack.  $\text{NO}_x$   
219 mixing ratios in air withdrawn from all sampling inlets in the snowpack dropped steadily, but remained  
220 elevated in comparison to levels seen before the pollution event for  $\approx$  one week. The behavior seen in  
221 the measurements from snow tower 1 were in full qualitative, and within  $\approx 30\%$  quantitative agreement  
222 with the concurrent observations from the second snow tower. After the pollution event,  $\text{NO}_x$  in the  
223 snowpack air gradually steadily declined over several days (Figure 6). Fitting of the data to an exponential  
224 decay function yields similar results for all snowpack depths (Figure 7), with exponential regression fit  $R^2$   
225 results of  $\geq 0.95$ .

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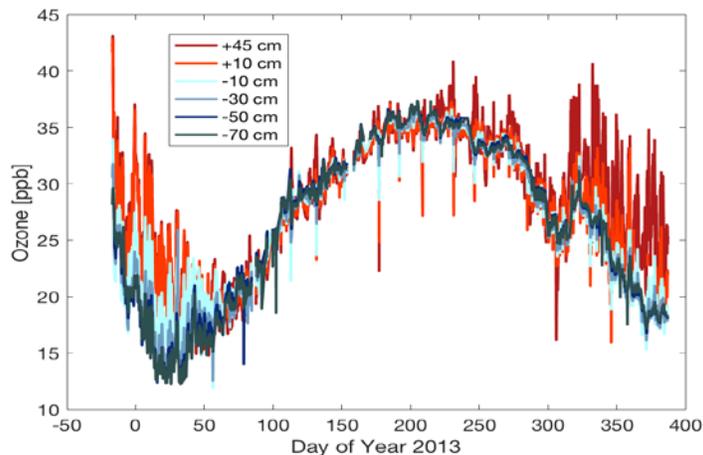
**Figure 7.** Exponential decay function fits to the  $\text{NO}_x$  snowpack measurements versus time at four depths for the event starting on Day of Year 191 shown in Figure 6. The start of the event was defined as the time when high  $\text{NO}_x$  was detected above the snowpack. Solutions for the best fit exponential decay functions are given in the legend.

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229 Effects of the exhaust transport were also observed in the ozone signal. The ozone record, shown in  
 230 Figure 8, shows a plethora of short positive and negative spikes superimposed on the annual cycle. The  
 231 up to 15-20 ppb sudden ozone increases seen during the austral summer months are attributable to the  
 232 photochemical ozone production events that occur in the surface layer of the Antarctic Plateau (see  
 233 discussion in the introduction section). Occurrences of these elevated ozone events at Concordia Station  
 234 have previously been investigated by Legrand et al. [2009] and Cristofanelli et al. [2018]. Besides these  
 235 positive ozone spikes, this annual record also Here, the signal shows numerous was sudden negative ozone  
 236 changes that can be attributed to from destruction of ozone by titration of NO in the exhaust plume. Up  
 237 to 50% of the ambient ozone was destroyed in air sampled from the above surface inlets. Similar to  $\text{NO}_x$ ,  
 238 this signal, albeit weaker and attenuated in time, was also seen in the air sampled from within the  
 239 snowpack (Supplement Figure S5-1).

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**Supplement Figure 8S-1.** Ozone measured from the snow tower inlets throughout the year. Negative spikes in the data coincide with elevated  $\text{NO}_x$  from exhaust infiltration in the snowpack.

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242 Enhancements in formaldehyde in ambient samples suspected to be influenced by station emissions  
 243 have previously been noted in measurements taken during the OPALE campaign [Preunkert et al., 2015].  
 244 We therefore investigated if pollution signatures were present in formaldehyde measurements taken in  
 245 the record from our study. Furthermore, we revisited the gaseous elemental mercury (GEM)  
 246 measurements taken during the campaign [Angot et al., 2016]. Both of these measurements did not  
 247 quite have the time resolution and sensitivity as the  $\text{NO}_x$  and ozone monitoring. Formaldehyde  
 248 measurements overlapped with the ozone and  $\text{NO}_x$  monitoring only for a short period during the  
 249 2012/2013 austral summer, and this measurement period suffered from a number of instrument  
 250 problems. The remaining data did not allow a conclusive evaluation on the behavior of formaldehyde in  
 251 the snowpack during pollution events. Similarly, we did not identify a clear signature of GEM changes in  
 252 snowpack air that correlated with the  $\text{NO}_x$  enhancements during pollution events.

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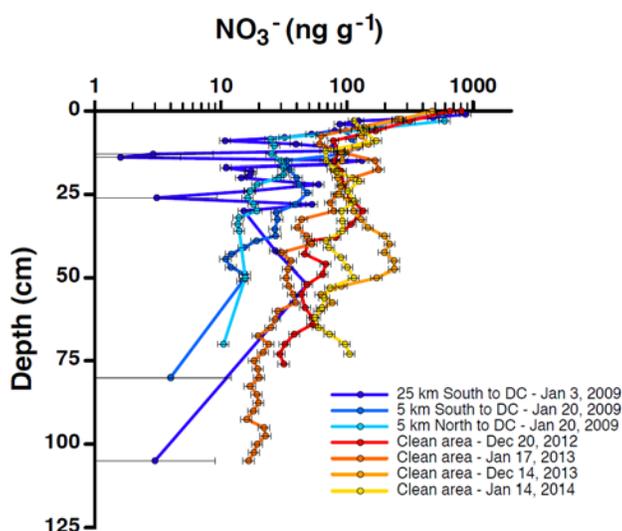
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253 Nitrogen oxides undergo reaction with atmospheric oxidants, primarily the OH radical (summer only)  
 254 and ozone, yielding higher oxidized nitrogen species (including  $\text{NO}_3$ ,  $\text{N}_2\text{O}_5$ ,  $\text{HNO}_3$ ,  $\text{HNO}_4$ ) that can further  
 255 react and partition into the snowpack aqueous and solid phases. The frequency, large enhancement, and  
 256 long duration (in the snowpack) of  $\text{NO}_x$  pollution events constitute an apparent unnatural source of  $\text{NO}_x$   
 257 to the snowpack. One can hypothesize that further reaction of NO and  $\text{NO}_2$  with oxidants, such as with  
 258  $\text{OH}$ , may be a source of  $\text{HNO}_2$  and  $\text{HNO}_3$  in the snowpack, which would add acidity to the snow. This then  
 259 also poses the question if and to what degree photochemical processes, building for instance on  $\text{NO}_2^-$  or  
 260  $\text{NO}_3^-$  as a substrate, may be altered from natural conditions. Further, the transformations of  $\text{NO}_x$  into

261 these higher oxidized species may potentially leave a long-lasting-term chemical signature in the snowpack  
262 (such as of NO<sub>3</sub><sup>-</sup>).

263 We investigated this question by comparing NO<sub>3</sub><sup>-</sup> results from snow pit sampling at different locations  
264 within the camp and at up to 25 km distance of the Dome C-Concordia Station. Nitrate in the snowpack  
265 shows a steep vertical gradient, with highest levels observed right at the surface, and progressively lower  
266 concentrations with increasing depth (Figure 99; and Supplement Figure S-1 for a summary graph where  
267 data from both groups were combined and binned in 15 cm depth intervals). The results from the seven  
268 snow pits are consistent in the depth profile;

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269  
270 **Figure 99.** Nitrate concentration in snow pits in proximity ( $\approx 20$  m) to the snow towers (warm colors) at the  
271 border of the clean air sector (warm colors), and sampling locations up and downwind of Concordia (cold  
272 colors). Sampling dates are indicated in the figure legend. Horizontal error bars depict the estimated  
273 uncertainty of the chemical analysis, i.e. 10% at  $> 10$  ng g<sup>-1</sup>, 50% between 5-10 ng g<sup>-1</sup>, and 100% at  $< 5$  ng g<sup>-1</sup>.

274 however, there appears to be a tendency that the four snow pits within the camp have somewhat higher,  
275 as well as more variable snow NO<sub>3</sub><sup>-</sup> at intermediate depths (around 50 cm). Deeper into the snowpack  
276 the difference between the two groups of data becomes weaker/stronger. At 1 m depth, the snow age is  
277 approximately ten years, and therefore approximately corresponds to the start of permanent and year-  
278 round activities in 2005 (note that summer activities at Dome Concordia Station were established in 1996).  
279 This behavior in the similar vertical NO<sub>3</sub><sup>-</sup> vertical gradient concentration profiles in the Antarctic snowpack  
280 has been have previously investigated and been documented [France et al., 2011]. Further, relatively  
higher NO<sub>3</sub><sup>-</sup> concentrations (approximately 5 times) are observed in the upper layers of the snowpack-is  
observed during the Antarctic summer (Supplement Figure S-2-2), with average NO<sub>3</sub><sup>-</sup> levels approximately  
5 times higher than during the winter. This NO<sub>3</sub><sup>-</sup> enhancement in surface snow and the seasonal cycle

281 have been linked to the production of HNO<sub>3</sub> in the photoactive summer months from reaction of OH with  
282 NO<sub>2</sub>, with the NO<sub>3</sub><sup>-</sup> enhancement being the result of HNO<sub>3</sub> deposition to the snow surface [Erbland *et al.*,  
283 2013]. Elevated NO<sub>3</sub><sup>-</sup> concentrations observed at the surface of the snowpack are a common feature at  
284 low accumulation regions, with the concentration values depending strongly on the strategy for collection  
285 of the first few cm of the snow-pack [Erbland *et al.*, 2013; Shi *et al.*, 2018]. With the sensitivity of surface  
286 snow NO<sub>3</sub><sup>-</sup> to the seasonal cycle and ~~sampling strategy~~, the high variability of NO<sub>3</sub><sup>-</sup> observed ~~at the in~~  
287 ~~surface snow and vertical profiles, the currently available data do not allow surface precludes a~~ conclusive  
288 ~~quantitative assessment interpretation to what of the~~ degree ~~to which~~ the snowpack at Concordia is  
289 ~~chemically~~ impaired by ventilation of the snowpack with pollution-NO<sub>x</sub> enriched air. ~~Contamination of~~  
290 ~~the snowpack around Concordia station has been noted in previous investigations.~~ Warren *et al.* [2006]  
291 ~~reported an > 3-fold increase in snowpack black carbon concentration after the station was established in~~  
292 ~~2003, with black carbon levels in pre-2003 snow (from deeper depths) also being in closer agreement with~~  
293 ~~snow collected from further distance to the station. Black carbon is one of the contributing factors for a~~  
294 ~~decrease in light penetration into the snowpack.~~ [Warren *et al.*, 2006]-[France *et al.*, 2011]-[Libois *et al.*,  
295 2013]. ~~Consequently, the increased presence of black carbon causes a shallower e-folding depth~~  
296 ~~compared to pristine, uncontaminated snow. Contamination of the snow pack around the Concordia~~  
297 ~~station has been noted in previous investigation of black carbon, with a > 3 fold increase in concentration~~  
298 ~~between pre and post 2003 [Warren et al., 2006]. This increase in black carbon was shown to result in a~~  
299 ~~significant decrease in light penetration into the snowpack [Warren et al., 2006; France et al., 2011; Libois~~  
300 ~~et al., 2013].~~ In addition to the experiments described above, during the 2014 campaign a number of  
301 ~~dynamic flow-through snow chamber photochemistry experiments were conducted to investigate if there~~  
302 ~~were differences in the reactive chemistry in the snow from near the snow tower site compared to snow~~  
303 ~~sampled 25 km away from camp. These measurements showed on the order of 10-20% higher NO<sub>x</sub> and~~  
304 ~~less ozone in the outflow of chambers filled with the snow from further away from the camp. We did not~~  
305 ~~conduct a high enough number of repeats for evaluating the repeatability and statistical significance of~~  
306 ~~these results to gauge if and how much of this signal was due to the experimental setup or due to~~  
307 ~~differences in the snowpack chemical composition. Nonetheless, these preliminary findings point towards~~  
308 ~~possible differences in the chemical behavior that potentially are linked to differences in the snow~~  
309 ~~sampling locations and contaminant levels resulting from camp influences that warrant further~~  
310 ~~investigation.~~In addition to the experiments described thus far, during the 2014 campaign a number of  
311 ~~dynamic flow through snow chamber photochemistry experiments were conducted comparing snow from~~  
312 ~~near the snow tower site to snow sampled 25 km away from camp. These experiments showed on the~~  
313 ~~order of 15% differences in ozone loss in the chambers (unpublished results). We did not conduct a high~~  
314 ~~enough number of repeats for evaluating the repeatability of these experiments to gauge if and how much~~  
315 ~~of this signal was due to the experimental setup or the due to differences in the snowpack chemical~~  
316 ~~composition. Nonetheless, these preliminary findings point towards possible chemical behavior that~~  
317 ~~potentially is linked to differences in the snow sampling locations and camp influences.~~

318

## 319 Summary and Conclusions

320 With our snowpack sampling manifold we were able to sample snowpack air to a maximum depth of 70  
321 cm below the surface. Up to  $\approx$  2 ppb enhancements in NO<sub>x</sub> ~~were~~ observed at that depth from exhaust

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322 infiltration. While our experiment was not able to access air deeper (than 70 cm), the observed  
323 concentration gradients observed imply that this transport and contamination extends well beyond the  
324 depth that was probed in these measurements.

325 Our This experiment was a one spot measurement, at  $\approx$  1 km distance from the camp main facilities. We  
326 have no data that would allow us to assess to what distance from the camp the snowpack pollution from  
327 exhaust infiltration would be noticeable and of importance, but it likely extends well beyond the  
328 distance of our site.

329 Several other previous studies have noted challenges in sampling clean air at polar research sites that  
330 stem from pollution caused by camp exhaust. For atmospheric sampling, this interference can be  
331 mitigated by careful postprocessing/filtering of the monitoring data, or by interrupting the sampling  
332 during unfavorable wind conditions, which is particularly critical for integrated aerosol collection [Wolff  
333 and Cachier, 1998; Wolff et al., 1998]. Our experiments measurements from Concordia Station emphasize  
334 the pronounced and longer-lasting influence that station exhaust can have on NO<sub>x</sub> levels inside the  
335 snowpack air (compared to ambient air). A tendency of potentially enhanced snowpack NO<sub>3</sub><sup>-</sup> levels in  
336 two snow pits collected at the camp, compared to data from three sites at further distance, supports the  
337 suspicion that the snowpack chemical oxidized nitrogen composition at the station may be compromised  
338 (i.e. contaminated) from the re-occurring ventilation of the snowpack with polluted, NO<sub>x</sub>-enriched air. A  
339 similar conclusion was derived from d<sup>15</sup>N nitrate analyses of snow at Summit. Samples collected in the  
340 predominant downwind direction of the station generator showed an isotopic signature that had a  
341 stronger association with engine exhaust than samples collected at further distance [Fibiger et al., 2016].  
342 Even in the Summit clean air sector elemental carbon in snow was 1.8 – 2.4 times higher than in snow  
343 collected at 10-20 km distance from the camp [Hagler et al., 2008]. A tendency of potentially enhanced  
344 snowpack NO<sub>3</sub><sup>-</sup> levels in two snow pits collected at the camp, compared to data from three sites at further  
345 distance supports the suspicion that the snowpack composition at the station may be compromised (i.e.  
346 contaminated) from the re-occurring ventilation of the snowpack with polluted (NO<sub>x</sub>-enriched) air, which  
347 adds to the previous findings on black carbon, and organic gases snowpack contamination from engine  
348 exhaust. A tendency of potentially enhanced snowpack NO<sub>3</sub><sup>-</sup> levels in two snow pits collected at the camp,  
349 compared to data from three sites at further distance supports the suspicion that the snowpack  
350 composition at the station may be compromised (i.e. contaminated) from the re-occurring ventilation of  
351 the snowpack with polluted (NO<sub>x</sub>-enriched) air. Chemical signatures of other trace species that have  
352 enhanced concentration levels in engine exhaust, such as black carbon, organic gases are left behind in  
353 the snowpack is already demonstrated, resulting in a decrease of the e-folding depth of the light  
354 penetration. The associations shown in our study argue for further investigation, for instance by a high  
355 resolution spatial survey of surface snow composition within and beyond camp boundaries. Given the  
356 strong seasonality of NO<sub>3</sub><sup>-</sup>, this survey should be done with as close as possible concurrently conducted  
357 snow sampling at selected locations to minimize the influence of temporal changes on the NO<sub>3</sub><sup>-</sup> signature.

358 These observations emphasize concerns about the representativeness of experimental snow chemistry  
359 data collected within a Polar research camp periphery. This raises the question how interpretations from  
360 such experiments reflect conditions in the remote Polar environment. Furthermore, our findings should  
361 motivate comparison studies with sampling along transects to further distance from the main camp

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362 facilities. Comparison of these observations will likely yield new insights for evaluating prior polar  
363 research site observations and interpretation of snow photochemistry in the glacial snowpack.

364

365 **Author contribution:**

366 DH oversaw the study, participated in field work, conducted data analyses and quality control, and  
367 prepared the manuscript.

368 DL conducted data analyses, quality control, prepared figures, and contributed to the manuscript  
369 preparation.

370 JH fabricated the instrumental equipment and data acquisition system, and participated in the field work.

371 JS contributed to the study design, participated in field work, conducted data analyses and interpretation,  
372 and contributed to the manuscript preparation.

373

374

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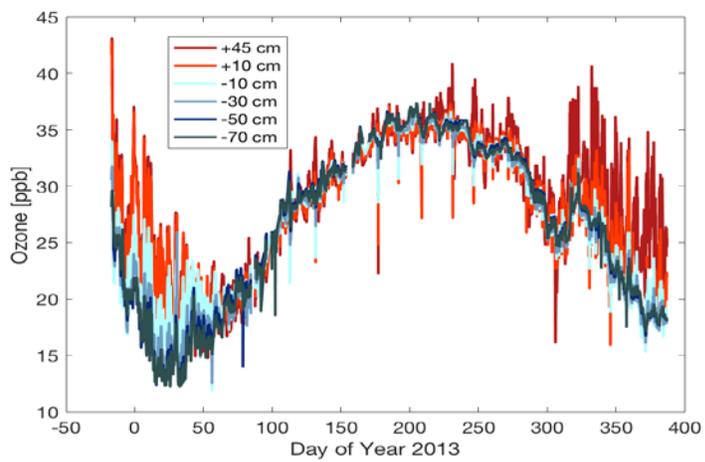
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524

525 **Supplemental Materials**

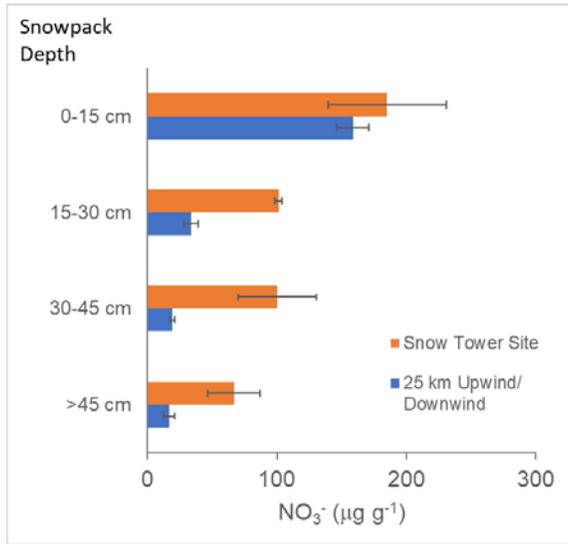
526



**Supplement Figure S-1.** Ozone measured from the snow tower inlets throughout the year. Negative spikes in the data coincide with elevated  $\text{NO}_x$  from exhaust infiltration in the snowpack.

527

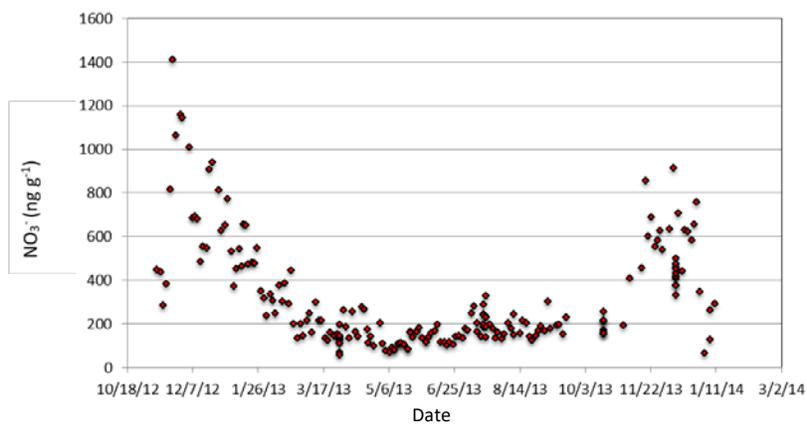
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**Supplement Figure S-12.** Seasonal cycle of nitrate in skin layer surface snow collected at Dome C three times per week during October 2012–January 2014. Data from manuscript Figure 9, with the two types of samples (snow pits in proximity ( $\approx 20$  m) to the snow towers (orange), and sampling locations up and downwind of Concordia (blue)) combined and binned in 15 cm depth intervals. Error bars depict the 1- $\sigma$  standard deviation of available data within each bin.

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**Supplement Figure S-22.** Seasonal cycle of nitrate in skin layer surface snow collected at Dome C three times per week during October 2012 - January 2014.

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530

531 **Response to Anonymous Referee #1 Comments**

532

533 **We thank this reviewer for the constructive feedback on our manuscript. Our responses and changes**  
534 **implemented in the manuscript are summarized below in blue bold font after each of the reviewer's**  
535 **comment points.**

536

537 GENERAL COMMENTS

538 This study reports observations of air and snow chemistry from a 14 month measurement campaign  
539 at Dome C, East Antarctica with a focus on pollution episodes caused by the year-round operated  
540 Concordia station. Gasphase concentrations of nitrogen oxides (NO<sub>x</sub>) increased by several orders of  
541 magnitude during a total of 15 events of pollution transport from the station during the study period,  
542 with concurrent decreases in ozone indicating titration by nitric oxide. While near-surface air  
543 concentrations of NO<sub>x</sub> settled to background levels within hours, firn air concentrations remained  
544 elevated for days to up to a week. The snow chemical composition may be affected as indicated by  
545 nitrate concentrations in snow below the top layer being higher near the station than further away.  
546 The authors conclude that these observations provide insight to the discussion on how  
547 representative snow photo-chemistry studies near research stations are for the wider Antarctic ice  
548 sheet.

549 The detection of elevated NO<sub>x</sub> and associated O<sub>3</sub> decrease when sampling station power  
550 generator exhaust, and of course also any kind of motorised traffic (air plane, ski doo, tractors etc)  
551 is not new, but a well-known limitation of studies of background atmospheric chemistry at most  
552 polar research stations. This is typically mitigated by careful filtering of atmospheric measurements.

553 A new insight from this study is the observation that pollution plumes diffuse into the upper  
554 snowpack and remain there for several days, with the potential to alter chemical composition of  
555 the upper snowpack near the station and therefore compromise local snow photochemistry being  
556 representative for the wider ice sheet. Understanding changes in air and snowpack chemistry near  
557 Concordia station would be potentially very useful for the design of future experiments at Dome C  
558 and elsewhere. Rather than just casting doubt on previous measurement campaigns I suggest the  
559 authors improve the manuscript by broadening the range of considered chemical species and by  
560 providing more quantitative detail in their discussion of pollution impacts on snow chemistry as  
561 promised by the title:

562 The abstract states that other reactive chemical species were also measured, notably formaldehyde  
563 and gaseous elemental mercury (GEM). Even though not as sensitive to pollution as NO<sub>x</sub> I urge the  
564 authors to present and discuss those data during pollution events to achieve a more general picture  
565 of pollution impacts near the station. For example formaldehyde has been discussed as a  
566 snowpack HO<sub>x</sub> radical precursor at Dome C (Preunkert et al., 2015), and it would be important to  
567 understand if and how station pollution changes also nearby CH<sub>2</sub>O profiles in snow.

568 **We re-examined the available formaldehyde and GEM data from the campaign. Those**  
569 **measurements do not quite have the time resolution and sensitivity as the NO<sub>x</sub> and ozone**  
570 **monitoring. Enhancements in ambient samples suspected to be influenced by station emissions have**  
571 **previously been noted by in measurements taken during the OPALE campaign in 2011 [Preunkert et**  
572 **al., 2015]. Formaldehyde measurements overlapped with the ozone and NO<sub>x</sub> monitoring only for a**

573 short period during the 2012/2013 austral summer, and this measurement period suffered from a  
574 number of instrument problems. The remaining data did not allow a conclusive evaluation on the  
575 behavior of formaldehyde in the snowpack during pollution events. We also revisited the gaseous  
576 elemental mercury (GEM) measurements taken during the campaign [Angot et al., 2016]. These data  
577 did not show a clear signature of GEM in snowpack air that correlated with the NO<sub>x</sub> enhancements  
578 during pollution events. This information has been added to the manuscript text. There are no other  
579 chemical gas measurements available from the campaign for a further investigation.

580 Snow nitrate profiles observed in the clean-air sector near the station show a less steep decline  
581 with depth and slightly higher concentrations than at sites further away (>5km). Please discuss the  
582 role of spatial variability in snow accumulation and stratigraphy. The presence of sastrugi as well  
583 as substantial snow drift around buildings/obstacles as seen on Fig.1 certainly have an impact on  
584 snow chemistry profiles. Accumulation rate determines exposure time of a given snow layer to  
585 actinic flux as well as stratigraphy/ snow optical properties.

586 **Authors' response:** The skin layer of the surface snow has been collected at Concordia every 3 days  
587 for more than 8 years. Furthermore, every three months surface snow is collected randomly in an  
588 area of ca. 100 m<sup>2</sup> at the same time. The standard deviation in the nitrate concentration is ca. 20-30%.  
589 Manuscript Figure 8, and the new Supplement Figure 3 provide the available data for evaluation of  
590 the variability in nitrate results in snow samples collected from snow pits at different locations.

591 Can the authors provide an estimate of at least the order of magnitude of how much snow nitrate  
592 could be produced from pollution episodes throughout a year, knowing O<sub>3</sub> and NO<sub>x</sub> in firn air &  
593 assuming HO<sub>x</sub> levels in summer?

594 **Authors' response:** We agree that this is an interesting and compelling question, but we do not  
595 see a way to possibly develop a nitrate production estimate given the available data and analysis  
596 tools.

597 I'd suggest to double-check with co-author Savarino, who has measured the stable nitrogen isotope  
598 <sup>15</sup>N in atmospheric and snow nitrate over several years now at Dome C. Assuming <sup>15</sup>N in NO<sub>x</sub> from  
599 the station plume is different from other sources, did they ever detect any evidence from existing  
600 measurements of <sup>15</sup>N in nitrate in snow (or in atmospheric particulate no<sub>3</sub>) that would indicate a  
601 contribution from local pollution?

602 **Authors' response:** Thus far there has been no indication of <sup>15</sup>N signals from plume pollution. It  
603 needs to be realized that that the amount of nitrate in the snowpack is orders of magnitude larger  
604 than the sum of total oxidized nitrogen gases in the atmospheric boundary layer. Therefore,  
605 influences on the differences in the isotopic signature of atmospheric nitrogen would be very difficult  
606 to detect in snowpack samples.

607

608 SPECIFIC COMMENTS

609 Abstract & I162: in the text and Fig.4 a total of 50 pollution events are reported, whereas the abstract  
610 states 15 occasions of plume transport, how do the different numbers arise? Please explain.

611 **Authors' response:** In total there were about 50 pollution events. Of those, we counted 15 events  
612 in total when there was a spike in the above surface NO<sub>x</sub> measurements followed by an increase of at  
613 least 1 ppb of NO<sub>x</sub> within the snowpack. This detail has been added to the text.

614 196-115 Please clarify the schedule of the firm air sampling. Apparently inlets were switched  
615 between the 2 snow towers every 24 hr (caption of Fig.6)?

616 **Authors' response: Yes, the reviewer is correct. We added this information to the method**  
617 **description as well.**

618 1235-239 This section requires further clarification: e-folding depth depends in a non- linear fashion  
619 on both, absorbing impurities (BC, HULIS etc) as well as on snow density & grain size/shape  
620 (scattering). Thus, the variability observed (Warren et al., 2006; France et al., 2011) is not only due  
621 to BC deposition but in part also due to the snow stratigraphy present at DC.

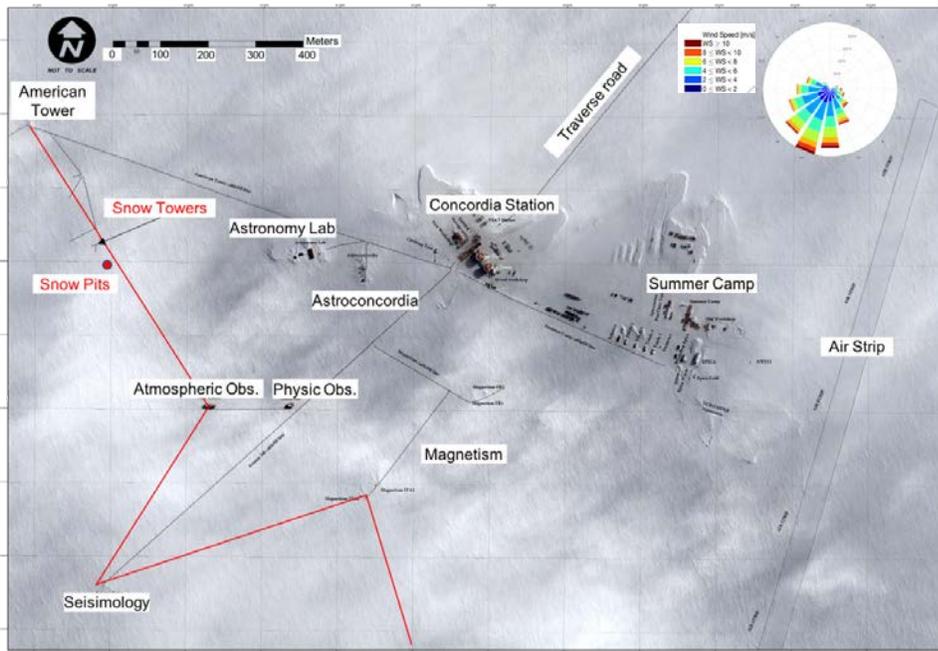
622 **Authors' response: The sentence was reworded to: "Contamination of the snowpack around**  
623 **Concordia station has been noted in previous investigations. [Warren et al., 2006] reported an > 3-fold**  
624 **increase in snowpack black carbon concentration after the station was established in 2003, with black**  
625 **carbon levels in pre-2003 snow (from deeper depths) also being in closer agreement with collected**  
626 **snow from further distance to the station. Black carbon is one of the contributing factors for a**  
627 **decrease in light penetration into the snowpack [Warren et al., 2006; France et al., 2011; Libois et al.,**  
628 **2013]. Consequently, the black carbon snowpack contamination results in a shallower e-folding depth**  
629 **compared to pristine, uncontaminated snow.**

630 Fig.1: To better illustrate the location of the Atmospheric Observatory add a wind rose and  
631 boundaries of the clean air sector.

632 **Authors' response: A wind rose and lines that indicate the border of the clean air sector were**  
633 **added to the figure. The revised figure is shown below.**

634

635



636  
637

638 **Figure 1:** Satellite image of research station Concordia with location of the snow photochemistry  
639 experiment indicated by the red text. Its location was ~1 km west of the station main buildings and  
640 power generation plant. The map also indicates the location where the snow pits were collected, the  
641 border of the clean air sector (which is west of the red line), and a wind rose for a full year of wind data  
642 collected from the met tower that was adjacent to the snow tower manifolds.

643

644 Fig.2: Text only is hard to follow so please add a schematic indicating location and height/depth of  
645 all inlets and sensors, as well as of the snow pit profile.

646 **Authors' response:** We added this text to the figure caption: "(see Figure 1 in van Dam et al. (2015)  
647 for a schematic of a similar installation at Summit, Greenland)". Indication of the location of the snow  
648 pit sampling was added to Figure 1, and the snow pit profile data are shown in Figure 9.

649 Fig.6: caption "... measurements FROM the snow tower.." Supplement-Fig S-1 and S-2: in the  
650 online version no supplement was available. I suggest to show the O3 mixing ratios and the  
651 surface snow no3 also in the main manuscript.

652 **Authors' response:** The ozone graph was moved from the Supplement to the main manuscript.  
653 But we prefer keeping the surface nitrate graph in the Supplement as these data are purely  
654 ancillary supporting information.

655

656

657 REFERENCES

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678 **Response to Anonymous Reviewer #2 Comments**

679 **We thank reviewer #2 for the constructive comments and suggestions for improving our manuscripts.**  
680 **Our responses are inserted below and highlighted in dark blue font for easier identification.**  
681

682  
683 In “Intrusion, retention, and snowpack chemical effects from exhaust emissions at Concordia Station,  
684 Antarctica,” Helmig, et al. show the impact station exhaust can have on even the clear air sector, at  
685 a remote, high-latitude, research station. They show incidents of high NO<sub>x</sub> above the snowpack  
686 correspond with winds from camp. They also show the lag in higher NO<sub>x</sub> concentrations in the  
687 snowpack and that the higher concentrations at depth persist for longer than those over the  
688 snowpack.

689 This work is important, but not unique. It reemphasizes that more researchers need to look at the  
690 potential impacts of station activities on snow and air chemistry, even in the designated “clean air  
691 sector” at these stations. Others have shown the impact of station emissions on snow and air  
692 chemistry at remote stations, including the impact of NO<sub>x</sub> from station power generation.

693 General comments:

694 The authors do not cite other work showing the impact of station exhaust on NO<sub>x</sub> in the air and  
695 nitrate in the snow. Fibiger et al. (2016) show the impact of station exhaust on NO<sub>x</sub>  
696 concentrations above the snowpack and on nitrate in snow at Summit, Greenland. They use δ<sup>15</sup>N to  
697 quantify this influence on snow nitrate. More discussion of how the influence at Dome C compares  
698 with these findings would greatly strengthen this analysis.

699 **Authors’ response: We have included discussion and a reference to this publication in the section**  
700 **on our results of the snow pit data in our revised manuscript.**

701 Wolff et al. (1998), also show this issue with filter collections at various stations, including Durmont  
702 d’Urville, which could result in deposited nitrate.

703 **Authors’ response: We have also added reference to two articles published by this group of**  
704 **authors in the discussion and conclusion sections (Wolff et al., 1998a,b).**

705 The authors state in the methods section that the NO measurement is more accurate than NO<sub>x</sub>,  
706 which contains several other NO<sub>y</sub> species. Throughout the manuscript, however, the authors use  
707 the NO<sub>x</sub> measurement. Why has this been chosen over NO? The enhancements should still show up  
708 in NO, without the interference issues.

709 **Authors’ response: We chose NO<sub>x</sub> as it is a more representative indicator for the total of the**  
710 **oxidized nitrogen, whereas NO would only indicate a fraction. Further, the fractionation between**  
711 **NO and NO<sub>x</sub> is sensitive to other gases (such as ozone) and residence time and snowpack depth.**  
712 **Therefore, interpretation of NO data is more ambiguous.**

713 In figure 6 the authors show the impact on firn NO<sub>x</sub> continues past day 198, which is 6 days after  
714 event onset. Separately, the authors state there are 15 (abstract) or 50 (text) occurrences of plumes  
715 over the sampling site. They conclude that less than 10% of the measurements were contaminated  
716 by the plume, but this is difficult to reconcile. The authors should provide more detail on how many  
717 events there are, how long they last.

718 **Authors’ response: We counted 15 events in total when there was a spike in the above air NO<sub>x</sub>**  
719 **measurements followed by an increase of at least 1 ppb within the snowpack. This detail has been**  
720 **added to the text.**

721 Also, the abstract states the NO<sub>x</sub> levels last “a few days to one week,” but in the text it says “1-5  
722 days.” These statements need to be verified and data needs to be presented more consistently and  
723 clearly.

724 **Authors’ response:** As can be seen in Figure 6, even after 7 days NO<sub>x</sub> in the snowpack still hadn’t  
725 completely returned to the pre-event levels. We have corrected the text accordingly, and now state  
726 that “..... it took up to in excess of 7 days for NO<sub>x</sub> in the snowpack air to return to pre-event levels.”

727

728 Specific comments:

729 Lines 18-20. This sentence needs to be reworded. It is not clear how many inlets there are or how  
730 many are above or below the snowpack.

731 **Authors’ response:** Sentence was reworded to: “Ambient air was sampled continuously from inlets  
732 mounted above the surface on a 10 m meteorological tower. In addition, snowpack air was collected  
733 in 30 cm intervals to 1.2 m depth from two manifolds that had both above and below surface  
734 sampling inlets.”

735 Line 102. The authors explain how they set inlets in the snow for sampling firn air.

736 They do not explain how confident they are that digging a pit and refilling it does not change the  
737 porosity of the snow or the dynamics of air flow. This could influence how applicable the firn results  
738 are.

739 **Authors’ response:** We added some more explanation. The text now reads: “The hole was then  
740 loosely refilled with the excavated snow, re-establishing the stratigraphy as much as possible. Blowing  
741 snow then refilled any remaining gaps within the following 2-3 days. The snow tower was kept in  
742 place after the campaign, so we have no data on the potential changes in porosity and air flow  
743 dynamics that resulted from the snow tower installation and subsequent changes in the snow  
744 morphology as the snowpack re-equilibrated.”

745 Line 130. The authors say their calibration gas was diluted with NO<sub>x</sub> scrubbed air. Was this tested  
746 against dilution with true zero air? Most NO<sub>x</sub> scrubbers emit a low, constant, level of NO<sub>x</sub>, which  
747 must be accounted for in calibrations. This tends not to be important in polluted areas, but is for 1  
748 ppb NO<sub>x</sub> measurements.

749 **Authors’ response:** We did not bring zero air compressed gas cylinder to Concordia. Instead, we  
750 prepared a low NO<sub>x</sub> dilution gas by pumping ambient air (which had significantly lower NO<sub>x</sub> levels  
751 than snowpack air) through a cartridge filled with ~ 1 dm<sup>3</sup> of granular Chemisorbent (Purafil,  
752 <https://www.purafil.com/products/chemical-filtration/chemical-media/#chemisorbant>).  
753 Calibration ranges were from 0.1-25 ppb and the instrument response was linear within this  
754 range. Intercepts of the linear regression, and zero values from sampling of the scrubbed air were  
755 below 0.1 ppb.

756 Line 192. Show this data in the supplement. It’s hard to know what these qualitative and  
757 quantitative comparisons look like without showing the data.

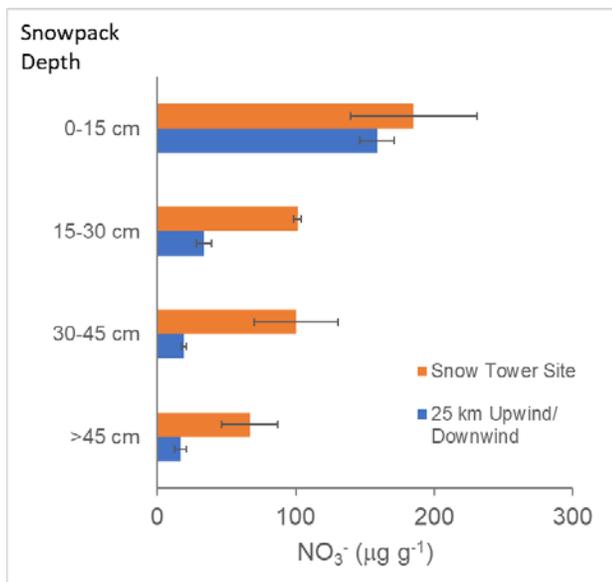
758 **Authors’ response:** The results shown in Figure 6 are actually the combined data from both snow  
759 towers. Wording has been added to the figure caption to make this clearer: “Combined  
760 measurements from the two snow towers capturing a pollution event at Concordia during the middle  
761 of the winter (Day of Year 191 = July 11). Plotted time series traces correspond to the sampling  
762 heights indicated in the legend, with positive numbers giving the height above the snow surface, and  
763 negative numbers the depth below the snow surface. The right graph (b) is an enlargement of the  
764 data shown in the left (a), with the transitions between the connecting lines between the two snow

765 tower measurements removed to better show the level of agreement in the data from the two  
766 sampling manifolds. The sampling switched between the two snow towers every 24 hours leading to  
767 some abrupt shifts in  $\text{NO}_x$  measurements from within the snowpack.”

768 Line 220. The authors say the difference becomes weaker at depth, but only one point seems to have  
769 a smaller difference. Support this statement with statistics. What are the differences at the surface  
770 and at depth? Is that difference significant?

771 **Authors' response:** Motivated by this reviewer comment we did some further analysis of these  
772 data. The within camp and 25 km distance data sets were combined and binned in 15 cm-depth  
773 increments. Results shown below indicate that the difference between the two data sets actually  
774 becomes larger with depth, and not weaker as we had mistakenly stated in the manuscript. This  
775 analysis does not show clear evidence for  $\text{NO}_3^-$  contamination at the snow tower site in the upper  
776 snowpack layers. However, deeper in the snowpack,  $\text{NO}_3^-$  levels appear to be higher at the snow  
777 tower site. A hypothesis explaining this behavior may be that the photolysis and loss of nitrate is  
778 weaker in the more contaminated snow, possibly due to the fact that the snowpack contains more  
779 chromophores than the unspoiled snow. Certainly, more snow pit sampling and chemical analyses  
780 are needed to investigate this stipulation. The figure was added to the Supplement, and this data  
781 discussion was added to the manuscript text.

782



783

784

785 Line 242. Do not cite unpublished results. Either publish them here, or do not include them. Also,  
786 which way is the 15% difference in ozone loss? Using data that others cannot see is bad scientific  
787 practice.

788 **Authors' response:** The section was changed to: “In addition to the experiments described above,  
789 during the 2014 campaign a number of dynamic flow-through snow chamber photochemistry

790 experiments were conducted to investigate if there were differences in the reactive chemistry in the  
791 snow from near the snow tower site compared to snow sampled 25 km away from camp. These  
792 measurements showed on the order of 10-20% higher NO<sub>x</sub> and less ozone in the outflow of chambers  
793 filled with the snow from further away from the camp. We did not conduct a high enough number of  
794 repeats for evaluating the repeatability and statistical significance of these results to gauge if and how  
795 much of this signal was due to the experimental setup or the due to differences in the snowpack  
796 chemical composition. Nonetheless, these preliminary findings point towards possible differences in  
797 the chemical behavior that potentially are linked to differences in the snow sampling locations and  
798 contaminant levels resulting from camp influences that warrant further investigation.

799 Line 264. Something is wrong in this sentence. You need to delete “organic gases” or reword.

800 **Authors’ response:** Sentence was shortened and connected to the previous sentence. It now  
801 reads: “A tendency of potentially enhanced snowpack NO<sub>3</sub><sup>-</sup> levels in two snow pits collected at the  
802 camp, compared to data from three sites at further distance, supports the suspicion that the  
803 snowpack composition at the station may be compromised (i.e. contaminated) from the re-occurring  
804 ventilation of the snowpack with polluted (NO<sub>x</sub>-enriched) air, which adds to the previous findings on  
805 black carbon, and organic gases snowpack contamination from engine exhaust at polar research  
806 stations.”

807

#### 808 **References**

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