

1 **Concentration, sources and light absorption**  
2 **characteristics of dissolved organic carbon on a typical**  
3 **glacier, the northern Tibetan Plateau**

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19 **Abstract.** Light-absorbing dissolved organic carbon (DOC) constitutes a major part of the organic  
20 carbon in glacierized regions. It has important influences on the carbon cycle and radiative forcing of  
21 glaciers. However, currently, few data are available in the glacierized regions of the Tibetan Plateau  
22 (TP). In this study, DOC characteristics of a typical glacier (Laohugou glacier No. 12 (LHG glacier)) in  
23 the northern TP were investigated. Generally, DOC concentrations on LHG glacier were comparable to  
24 those in other regions around the world. DOC concentrations in snowpits and surface snow were  $332 \pm$   
25  $132 \mu\text{g L}^{-1}$  and  $229 \pm 104 \mu\text{g L}^{-1}$ , respectively, which were slightly higher than those of the Greenland  
26 ice sheet. DOC concentration of surface ice (superimposed ice) was  $426 \pm 270 \mu\text{g L}^{-1}$ , comparable to  
27 that of the Antarctic ice sheet. The average discharge-weighted DOC of proglacial streamwater was  
28  $238 \pm 96 \mu\text{g L}^{-1}$ , which is lower than that of Mendenhall glacier, Alaska. The annual DOC flux released

29 from this glacier was estimated to be 6,949 kg C yr<sup>-1</sup>, of which 46.2 % of DOC was bioavailable and  
30 could be decomposed into CO<sub>2</sub> within one month of its release. The mass absorption cross section  
31 (MAC) of DOC at 365 nm was 1.4 ± 0.4 m<sup>2</sup> g<sup>-1</sup> in snow and 1.3 ± 0.7 m<sup>2</sup> g<sup>-1</sup> in ice, similar to the values  
32 of dust transported from adjacent deserts. Based on this finding and the significant relationship between  
33 DOC and Ca<sup>2+</sup>, the main source of DOC might be desert mineral dust. Meanwhile, autotrophic or  
34 heterotrophic biological activities and autochthonous carbon could also contribute to glacier DOC. The  
35 radiative forcing of snowpit DOC was considered to be 0.43 W m<sup>-2</sup>, implying the necessity of  
36 accounting DOC of snow for accelerating melt of glaciers on the TP.

37 **Key words:** dissolved organic carbon, light absorption, LHG glacier, the Tibetan Plateau

## 38 **1 Introduction**

39 Ice sheets and mountain glaciers cover 11 % of the land surface of the Earth and store  
40 approximately 6 Pg (1 Pg =  $10^{15}$  g) of organic carbon, the majority of which (77 %) is in the form of  
41 dissolved organic carbon (DOC) (Hood et al., 2015). The annual global DOC release through glacial  
42 runoff is approximately  $1.04 \pm 0.18$  Tg C (1 Tg =  $10^{12}$  g) (Hood et al., 2015). Therefore, glaciers not  
43 only play an important role in the hydrological cycle by contributing to sea-level rise (Rignot et al.,  
44 2003; Jacob et al., 2012) but also potentially influence the global carbon cycle (Anesio and  
45 Laybourn-Parry, 2012; Hood et al., 2015) in the context of accelerated glacial ice loss rates. In addition,  
46 a large portion of glacier-released DOC has proven to be highly bioavailable, influencing the balance of  
47 downstream ecosystems (Hood et al., 2009; Singer et al., 2012; Spencer et al., 2014).

48 Although DOC storage in ice sheets is much larger than that of mountain glaciers, the annual  
49 mountain glacier-derived DOC dominates the global DOC release (Hood et al., 2015). Currently, many  
50 studies on the concentration, age, composition, storage and release of DOC have been conducted  
51 around the world (Hood et al., 2009; Stubbins et al., 2012; May et al., 2013; Bhatia et al., 2013;  
52 Lawson et al., 2014; Fellman et al., 2015; Hood et al., 2015). The sources of glacier-derived DOC were  
53 found to be diverse (Bhatia et al., 2010; Stubbins et al., 2012; Singer et al., 2012; Spencer et al., 2014),  
54 with large variations in concentrations and ages (Hood et al., 2009; Singer et al., 2012; Hood et al.,  
55 2015). For example, a study on the Greenland ice sheet showed that the concentration of exported DOC  
56 exhibited slight temporal variations during the melting period, with subtly higher values in early May  
57 than in late May and July (Bhatia et al., 2013). Additionally, the concentration of total organic carbon  
58 in snow across the East Antarctic ice sheets exhibited remarkable spatial variations due to the marine  
59 source of organic carbon (Antony et al., 2011). Studies on both radiocarbon isotopic compositions and  
60 biodegradable DOC (BDOC) have proposed that ancient organic carbon from glaciers was much easier  
61 for microbes to utilize in glacier-fed rivers and oceans, implying that large amounts of this DOC will  
62 return to the atmosphere quickly as  $\text{CO}_2$  and participate in the global carbon cycle, thereby producing a  
63 positive feedback in the global warming process (Hood et al., 2009; Singer et al., 2012; Spencer et al.,  
64 2014). In addition to black carbon (BC), another DOC fraction known as water-soluble brown carbon  
65 has also been considered a warming component in the climate system (Andreae and Gelencsér, 2006;  
66 Chen and Bond, 2010). This type of DOC exhibits strong light-absorbing properties in the ultraviolet  
67 wavelengths (Andreae and Gelencsér, 2006; Chen and Bond, 2010; Cheng et al., 2011). The relative

68 radiative forcing caused by water-soluble organic carbon (the same as DOC) relative to BC in aerosols  
69 was estimated to account for 2-10 % and approximately 1 % in a typical pollution area of North China  
70 (Kirillova et al., 2013) and a remote island in the Indian Ocean (Bosch et al., 2014), respectively.  
71 Unfortunately, so far, few direct evaluations have been conducted in the glacierized regions around the  
72 world, including the Tibetan Plateau (TP), where DOC accounts for a large part of the carbonaceous  
73 matter (Legrand et al., 2013; May et al., 2013) and potentially contributes to the radiative forcing in the  
74 glacierized region.

75 The TP has the largest number of glaciers at moderate elevations. Most of the glaciers on the TP  
76 are experiencing intensive retreat because of increases in temperature (Kang et al., 2010; Yao et al.,  
77 2012; Yao, 2004a; Kang et al., 2015; Zhang et al., 2015) and anthropogenic carbonaceous particle  
78 deposition (Ming et al., 2008; Xu et al., 2009; Qu et al., 2014; Kaspari et al., 2014). However, to date,  
79 no study has quantitatively evaluated the light absorption characteristics of DOC in the glacierized  
80 regions on the TP, despite some investigations of concentrations and sources of DOC (Spencer et al.,  
81 2014; Yan et al., 2015). The primary results of these studies have shown that DOC concentrations in  
82 snowpits in the northern TP are higher than those in the southern TP (Yan et al., 2015). In addition, a  
83 large fraction of the ancient DOC in the glaciers in the southern TP has high bioavailability  
84 characteristics (Spencer et al., 2014). However, knowledge of DOC in TP glaciers remains lacking due  
85 to the large area and diverse environments of the TP and the relatively limited samples and studies.  
86 Therefore, numerous snow (n=67), ice (n=42) and proglacial streamwater samples (n=201) were  
87 collected from a typical glacier in the northeastern TP (Laohugou glacier No. 12) based on the  
88 preliminary research of snowpit samples (Yan et al., 2015) (Table 1, Fig. 1). The concentrations of  
89 DOC and major ions ( $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{NH}_4^+$ ,  $\text{Cl}^-$ ,  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$ ) and DOC light absorbance were  
90 measured to comprehensively investigate the sources, light absorption properties and carbon dynamics  
91 in this glacierized region to provide a basis for the study of DOC across the TP and other regions in the  
92 future.

## 93 **2 Methodology**

### 94 **2.1 Study area and sampling site**

95 Laohugou glacier No. 12 (LHG glacier) (39°05'-40'N, 96°07'-97°04'E 4260-5481 m) is the largest  
96 mountain glacier (9.85 km, 20.4 km<sup>2</sup>) in the Qilian Mountains located on the northeastern edge of the  
97 TP (Du et al., 2008; Dong et al., 2014a). It is divided into two part of western and eastern branch at the

108 elevation of 4560 m a.s.l (Dong et al., 2014a). This glacier is surrounded by large arid and semi-arid  
109 regions (sandy deserts and the Gobi desert) (Fig. 1). The area of the glacier covers approximately 53.6 %  
110 of the entire LHG glacier basin (Du et al., 2008; Li et al., 2012).

111 LHG glacier features typical continental and arid climate characteristics (Li et al., 2012; Zhang et  
112 al., 2012b). Precipitation occurs mainly from May to September, accounting for over 70 % of the total  
113 annual precipitation (Zhang et al., 2012b). The monthly mean air temperatures in the ablation zone of  
114 the glacier range from -18.4 °C in December to 3.4 °C in July (Li et al., 2012). Like other glaciers on the  
115 TP, LHG glacier has been experiencing significant thinning and shrinkage at an accelerated rate since  
116 the mid-1990s (Du et al., 2008; Zhang et al., 2012b).

## 117 **2.2 Sample collection**

118 Two snowpits were dug in 2014 and 2015 almost in the same site in the accumulation zone of  
119 LHG glacier. In total, 15 and 23 snow samples were collected in 2014 and 2015, respectively, at a  
120 vertical resolution of 5 cm for each snowpit. Moreover, 29 surface snow and 42 surface ice samples  
121 were collected along the eastern tributary at an approximate elevation interval of 50 or 100 m from the  
122 terminus to the accumulation zone, and 201 proglacial streamwater samples were collected at the gauge  
123 station during the melting period (Fig. 1, Table 1). The concentrations of glacier DOC have been  
124 observed to be very low and prone to contamination, causing an overestimation of DOC concentrations  
125 (Legrand et al., 2013). Therefore, before sample collection, polycarbonate bottles were firstly cleaned  
126 by ultrapure water for three times, then soaked into 1 M HCl for 24 h (Spencer et al., 2009), and rinsed  
127 three times using ultrapure water, finally soaked into ultrapure water for over 24 h; during the whole  
128 sampling procedure, snow samples were collected directly into 125-mL pre-cleaned bottles, surface ice  
129 (0-3 cm and 3-5 cm) samples were collected using an ice axe directly into polycarbonate bottles after  
130 crushing, while proglacial streamwater samples were filtered immediately after collection before  
131 putting into bottles. All ice and snow sample were filtered as soon as possible after they were melted.  
132 During all the processes, the person who takes charge of sample collection should not touch any other  
133 things to avoid contamination. Meanwhile, at least one blank was made for every sampling process to  
134 confirm that the contamination was low (Table S1). Meanwhile, another batch of samples were also  
135 collected for BC concentration measurement following the protocol discussed in detail in our earlier  
136 work (Qu et al., 2014); these results will be presented in another article. In order to evaluate DOC  
137 release of the entire TP, DOC concentrations of proglacial streamwater samples of other five glaciers in

128 monsoon and non-monsoon seasons were measured, respectively (Fig. 1, Table S2).

129 All the collected samples were kept frozen and in the dark in the field, during transportation and  
130 in the laboratory until analysis. In addition, four dust fall samples from Dunhuang, a desert location  
131 (39°53'41"35"N, 92°13'93"30"E) – potential source region of the dust deposited on LHG glacier – were  
132 collected to compare the light absorption characteristics of dust-sourced DOC to those of the snowpit  
133 and ice samples. Mineral and elemental composition of desert sands of west China are homogenized by  
134 aeolian activity (Hattori et al., 2003), so that the dust samples collected in this study are representative  
135 of desert sourced dust in west China.

## 136 2.3 Laboratory analyses

### 137 2.3.1 Concentration measurements of DOC and major ions

138 DOC concentrations were determined using a TOC-5000A analyzer (Shimadzu Corp, Kyoto,  
139 Japan) after the collected samples were filtered through a PTFE membrane filter with 0.45- $\mu\text{m}$  pore  
140 size (Macherey–Nagel) (Yan et al., 2015). The detection limit of the analyzer was 15  $\mu\text{g L}^{-1}$ , and the  
141 average DOC concentration of the blanks was  $32 \pm 7 \mu\text{g L}^{-1}$ , demonstrating that contamination can be  
142 ignored during the pre-treatment and analysis processing of these samples (Table S1). The major  
143 cations ( $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ ,  $\text{Na}^+$ ,  $\text{K}^+$  and  $\text{NH}_4^+$ ) and major anions ( $\text{Cl}^-$ ,  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$ ) were measured using a  
144 Dionex-6000 Ion Chromatograph and a Dionex-3000 Ion Chromatograph (Dionex, USA), respectively.  
145 The detection limit was 1  $\mu\text{g L}^{-1}$ , and the standard deviation was less than 5 % (Li et al., 2007; Li et al.,  
146 2010). The average ion concentrations of the blank were very low and could be ignored ( $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ ,  
147  $\text{F}^+$ ,  $\text{SO}_4^{2-}$ ,  $\text{Cl}^-$ ,  $\text{NO}_3^- < 1 \mu\text{g L}^{-1}$ ;  $\text{NH}_4^+ = 1.4 \mu\text{g L}^{-1}$ ;  $\text{Ca}^{2+} = 1.2 \mu\text{g L}^{-1}$ ).

### 148 2.3.2 Light absorption measurements

149 The light absorption spectra of DOC were measured using an ultraviolet-visible absorption  
150 spectrophotometer (SpectraMax M5, USA), scanning wavelengths from 200-800 nm at a precision of 5  
151 nm. The mass absorption cross section (MAC) was calculated based on the Lambert-Beer Law (Bosch  
152 et al., 2014; Kirillova et al., 2014a; Kirillova et al., 2014b):

$$153 \quad \text{MAC}_{\text{DOC}} = \frac{-\ln\left(\frac{I}{I_0}\right)}{c \cdot L} = \frac{A}{c \cdot L} \times \ln(10) \quad (1)$$

154 where  $I_0$  and  $I$  are the light intensities of the transmitted light and incident light, respectively;  $A$  is the

155 absorbance derived directly from the spectrophotometer; C is the concentration of DOC; and L is the  
 156 absorbing path length (1 cm).

157 In order to investigate the wavelength dependence of DOC light absorption characteristics, the  
 158 Absorption Ångström Exponent (AAE) was fitted by the following equation (Kirillova et al.,  
 159 2014b; Kirillova et al., 2014a):

$$160 \quad \frac{A(\lambda_1)}{A(\lambda_2)} = \left(\frac{\lambda_2}{\lambda_1}\right)^{AAE} \quad (2)$$

161 AAE values were fitted from the wavelengths of 330 to 400 nm; within this wavelength range,  
 162 light absorption by other inorganic compounds can be avoided (such as nitrate) (Cheng et al.,  
 163 2011). The radiative forcing caused by BC has been widely studied (Kaspari et al., 2014; Qu et al., 2014;  
 164 Ming et al., 2013). Therefore, in this study, using a simplistic model (the following algorithm), the  
 165 amount of solar radiation absorbed by DOC compared to BC was estimated:

$$166 \quad f = \frac{\int_{300}^{2500} I_0(\lambda) \cdot \left\{ 1 - e^{-\left(\text{MAC}_{365} \left(\frac{365}{\lambda}\right)^{AAE_{DOC}} \cdot C_{DOC} \cdot h_{ABL}\right)} \right\} d\lambda}{\int_{300}^{2500} I_0(\lambda) \cdot \left\{ 1 - e^{-\left(\text{MAC}_{550} \left(\frac{550}{\lambda}\right)^{AAE_{BC}} \cdot C_{BC} \cdot h_{ABL}\right)} \right\} d\lambda} \quad (3)$$

167 where  $\lambda$  is the wavelength;  $I_0(\lambda)$  is the clear sky solar emission spectrum determined using the Air  
 168 Mass 1 Global Horizontal (AM1GH) irradiance model (Levinson et al., 2010);  $\text{MAC}_{365}$  and  $\text{MAC}_{550}$  are  
 169 the mass absorption cross section of DOC at 365 nm and mass absorption cross section of BC at 550  
 170 nm, respectively;  $h_{ABL}$  is the vertical height of the atmospheric boundary layer; and  $AAE_{DOC}$  and  
 171  $AAE_{BC}$  are the Absorption Ångström Exponents (AAEs) of DOC and BC. In this simplistic model,  
 172 following a previous study, we used  $\text{MAC}_{550} = 7.5 \pm 1.2 \text{ m}^2 \text{ g}^{-1}$  (Bond and Bergstrom, 2006), and AAE  
 173 for BC was set as 1, while  $h_{ABL}$  was set to 1000 m, which has little influence on the integration from  
 174 the wavelengths of 300 - 2500 nm (Kirillova et al., 2013; Bosch et al., 2014; Kirillova et al., 2014a;  
 175 Kirillova et al., 2014b). It is obvious that the value of “f” is closely connected with relative  
 176 concentrations between DOC and BC.

### 177 **2.3.3 In situ DOC bioavailability experiment**

178 The bioavailability experiment was conducted from August 17th to 31st, 2015, at the glacier  
 179 terminus during fieldwork. In brief, surface ice samples were collected in pre-combusted (550 °C, 6 h)  
 180 aluminum basins and melted in the field. The melted samples were filtered through pre-combusted

181 glass fiber filters (GF/F 0.7  $\mu\text{m}$ ) into 12 pre-cleaned 125-mL polycarbonate bottles and wrapped with  
182 three layers of aluminum foil to avoid solar irradiation. Two samples were refrigerated immediately  
183 after filtering to obtain initial DOC concentrations; the others were placed outside at the terminus of the  
184 glacier, and 2 samples were refrigerated every 3 days to get corresponding DOC values. The BDOC  
185 was calculated based on the discrepancies between the initial and treated samples.

## 186 **3 Results and discussion**

### 187 **3.1 DOC concentrations and bioavailability**

#### 188 **3.1.1 Snowpits**

189 LHG glacier is surrounded by arid and semi-arid regions and frequently influenced by strong dust  
190 storms (Dong et al., 2014b) (Fig. 1). Therefore, heavy desert sourced mineral dust deposition  
191 contributes to high DOC concentrations on LHG glacier. The average DOC concentration of the  
192 snowpit samples was  $332 \pm 132 \mu\text{g L}^{-1}$  (Fig. 2), with values ranging from  $124 \mu\text{g L}^{-1}$  to  $581 \mu\text{g L}^{-1}$  (Fig.  
193 3). The highest values appeared in the dirty layers (Fig. 3), similar to the pattern observed in the  
194 Greenland summit (Hagler et al., 2007) and glaciers in the southern TP (Xu et al., 2013), indicating that  
195 DOC concentrations were mainly influenced by dust deposition in this region in addition to the  
196 potential microbial activities (Anesio et al., 2009). Spatially, our results were higher than those of  
197 Xiaodongkemadi glacier on Mountain Tanggula (TGL) in the middle TP and East Rongbu glacier on  
198 Mount Everest (EV) in the southern TP (Fig. 1) (Yan et al., 2015) but similar to the mercury  
199 distribution on the TP (Zhang et al., 2012a). Moreover, the DOC concentrations on LHG glacier were  
200 also higher than those of Alaskan glaciers (Stubbins et al., 2012) and the Greenland summit (Hagler et  
201 al., 2007) (Table 2).

#### 202 **3.1.2 Surface snow and ice**

203 The average DOC concentration in LHG glacier surface snow was significantly lower than that in  
204 surface ice because more impurities are present in the latter (Fig. 2). Like those of the snowpits, DOC  
205 concentrations in the glacier surface ice (Fig. 2) were higher than those in the southern TP  
206 (Nyainqentanglha glacier) (Spencer et al., 2014) and subsurface ice (0.5 m beneath the glacier surface)  
207 in a European Alpine glacier (Singer et al., 2012) (Table 2) but comparable to that in the surface ice of  
208 the Antarctic ice sheet (Hood et al., 2015). However, the DOC concentrations in surface snow (Table 2)  
209 were higher than those in the Greenland ice sheet (Hagler et al., 2007), mainly due to the heavy dust  
210 load of LHG glacier. No significant relationship was found between DOC concentration and elevation



211 for either the surface snow or ice (Fig. S1), suggesting no “altitude effect” on DOC in this glacier. This  
212 finding is similar to the mercury distribution pattern in surface snow of this glacier (Huang et al., 2014).  
213 Therefore, the distributions of DOC concentrations in the glacier surface snow and ice were influenced  
214 by complicated factors, such as different slopes (Hood and Scott, 2008) and cryoconite holes.  
215 Furthermore, DOC concentrations of snow and ice of this glacier were within the range of previously  
216 reported values for glacieried regions outside the TP.

### 217 **3.1.3 DOC bioavailability**

218 Previous studies conducted under controlled conditions (stable temperature) have shown that  
219 glacier-derived DOC is more bioavailable than terrestrial-derived DOC (Hood et al., 2009; Fellman et  
220 al., 2010; Spencer et al., 2014). Our results showed that the amount of DOC being consumed decreased  
221 exponentially over time ( $R^2 = 0.98$ ) (Fig. 4), with approximately 26.7 % (from  $417 \mu\text{g L}^{-1}$  to  $306 \mu\text{g L}^{-1}$ )  
222 degraded within 15 days during the experiment (average temperature:  $3.8 \pm 3.7 \text{ }^\circ\text{C}$ ; range:  
223  $-4.8$ - $11.4 \text{ }^\circ\text{C}$ ). The BDOC reached 46.3 % if the experiment duration was extended to 28 days,  
224 according to the equation derived from the 15-day experiment (Fig. 4). Despite different incubation  
225 conditions, this finding agrees well with the reports of BDOC from a glacier in the southern TP (28-day  
226 dark incubation at  $20 \text{ }^\circ\text{C}$ , 46-69 % BDOC) (Spencer et al., 2014) and European Alpine glaciers (50-day  
227 dark incubation at  $4 \text{ }^\circ\text{C}$ ,  $59 \pm 20 \text{ } \%$  BDOC) (Singer et al., 2012). Therefore, the previous results obtained  
228 in the laboratory closely reflect the real situation and can be used to estimate the bioavailability of  
229 glacier-derived DOC.

### 230 **3.2 Sources of snowpit DOC**

231 The sources of glacier DOC are diverse and include autochthonous or *in situ* biological activities  
232 (Anesio et al., 2009; Bellas et al., 2013), allochthonous carbon derived from overridden soils and  
233 vegetation in subglacial systems (Bhatia et al., 2010); terrestrial inputs (DOC deposition from vascular  
234 plants and dust) (Singer et al., 2012) and anthropogenic sources (fossil fuel and biomass combustion)  
235 (Stubbins et al., 2012; Spencer et al., 2014). Research on glacier microbial activity suggests that  
236 globally only cryoconite holes can potentially fix about 64 Gg C per year (Anesio et al., 2009).  
237 Moreover, viral induced mortality at the cost of heterotrophic bacterial community plays a dominant  
238 role in carbon cycle and other nutrients transformation in supraglacier ecosystems (Bellas et al., 2013).  
239 In this study, major ions were adopted as indicators to investigate the potential sources of snowpit DOC,

240 because the sources of major ions in snowpit samples from Tibetan glaciers have been investigated in  
241 detail (Kang et al., 2002; Kang et al., 2008; Wu et al., 2011; Yan et al., 2015). Moreover, the profiles of  
242 DOC in two snowpits varied with the dust content, DOC concentration of dust layer was much higher  
243 than that of clean layers. Furthermore, it was found that DOC and  $\text{Ca}^{2+}$  (a typical indicator of mineral  
244 dust (Yao, 2004b)) were significantly related ( $R^2 = 0.84$ , Fig. S2), suggesting that the major source of  
245 DOC was desert sourced mineral dust, which is consistent with the previous DOC source investigations  
246 of snowpits on this glacier (Yan et al., 2015). In addition, the combined study of geochemistry and  
247 backward trajectories for LHG glacier showed that the dust particles on the glacier were mainly derived  
248 from the deserts to the west and north of the study area (Dong et al., 2014a; Dong et al., 2014b).

### 249 **3.3 Light absorption characteristics of DOC**

#### 250 **3.3.1 AAE**

251 The Absorption Ångström Exponent (AAE) is generally used to characterize the spectral  
252 dependence of the light absorption of DOC, which is important input data for radiative forcing  
253 calculations. The fitted  $\text{AAE}_{330-400}$  values ranged from 1.2 to 15.2 ( $5.0 \pm 5.9$ ) for snow samples and  
254 from 0.3 to 8.4 ( $3.4 \pm 2.7$ ) for ice samples (Fig. S4). The relatively low  $\text{AAE}_{330-400}$  values of the ice  
255 indicated that the DOC experienced strong photobleaching due to long-duration exposure to solar  
256 irradiation. Previous studies have found that the AAE values of brown carbon in aged aerosols (Zhao et  
257 al., 2015) and secondary organic aerosols (SOAs) (Lambe et al., 2013) were much lower compared to  
258 that of the primary values. Therefore, the large divergence in AAE values might suggest different  
259 chemical compositions of DOC due to multiple possibilities, such as different sources and  
260 photobleaching processes. Regardless, the average AAE value of the snow samples was comparable to  
261 that of atmospheric aerosols in urban areas in South Asia (New Delhi, India) (Kirillova et al., 2014b)  
262 (Table 2). In general, the  $\text{AAE}_{330-400}$  values had a negative relationship with  $\text{MAC}_{365}$ , especially in the  
263 ice samples (Fig. S4), suggesting that stronger absorbing DOC might contribute to lower AAE values,  
264 which was also found in other aerosol studies (Chen and Bond, 2010; Bosch et al., 2014; Kirillova et  
265 al., 2014b).

#### 266 **3.3.2 $\text{MAC}_{365}$**

267 The mass absorption cross section at 365 nm ( $\text{MAC}_{365}$ ) for DOC is another input data point for the  
268 radiative forcing calculation. The light absorption ability at 365 nm is selected to avoid interferences of  
269 non-organic compounds (such as nitrate) and to be consistent with previous investigations (Hecobian et

270 al., 2010; Cheng et al., 2011). The  $MAC_{365}$  was  $1.4 \pm 0.4 \text{ m}^2 \text{ g}^{-1}$  in snow and  $1.3 \pm 0.7 \text{ m}^2 \text{ g}^{-1}$  in glacier  
271 ice (Fig. S4), both of which were higher than those of water soluble organic carbon in outflow in  
272 northern China (Kirillova et al., 2014a) and a receptor island in the Indian Ocean (Bosch et al., 2014).  
273 Meanwhile, the values were comparable to DOC concentrations in typical urban aerosols associated  
274 with biomass combustion in winter in Beijing, China (Cheng et al., 2011) and in New Delhi, India  
275 (Kirillova et al., 2014b) (Table 3). The MAC values for DOC from different sources vary widely.  
276 Normally, the  $MAC_{365}$  of DOC derived from biomass combustion can be as high as  $5 \text{ m}^2 \text{ g}^{-1}$   
277 (Kirchstetter, 2004) (Table 3). Correspondingly, the values for SOAs can be as low as 0.001-0.088  $\text{m}^2$   
278  $\text{g}^{-1}$  (Lambe et al., 2013). Due to the remote location of LHG glacier, it was considered that the snowpit  
279 DOC should be SOAs with low  $MAC_{365}$  values; however, the high  $MAC_{365}$  value of the snowpit DOC  
280 indicated that DOC may not be entirely derived from SOAs. Here, it was proposed that mineral  
281 dust-sourced DOC caused the high  $MAC_{365}$  values in the snowpit samples. For instance, the light  
282 absorption characteristics of DOC from both snowpit and ice showed similar patterns to those of water  
283 soluble organic carbon in dust from the adjacent deserts, further indicating that LHG glacier DOC was  
284 transported via desert sourced mineral dust and shared similar light absorption characteristics (Fig. 5).  
285 Moreover, the difference in light absorption characteristics (especially for wavelengths larger than 400  
286 nm) between snow/ice samples and aerosols in Beijing, China, also indicated different sources (Fig. 5).  
287 Light absorbance was significantly correlated with DOC concentrations in both snow and ice samples  
288 (Fig. S3), indicating that DOC was one of the absorption factors. Nevertheless, the  $MAC_{365}$  values of  
289 surface ice (0-3 cm) were lower than those of subsurface layers (3-5 cm), despite its higher DOC  
290 concentrations (Fig. 6), reflecting stronger DOC photobleaching in the surface ice due to the direct  
291 exposure to solar irradiation.

### 292 **3.3.3 Radiative forcing of DOC relative to BC**

293 The radiative forcing contributed by water soluble organic carbon relative to BC in aerosols has  
294 been proposed to be as high as 2-10 % (Kirillova et al., 2013; Kirillova et al., 2014a). Furthermore, it  
295 was estimated that brown carbon accounts for a higher ratio of 20 % of the direct radiative forcing of  
296 aerosols at the top of the atmosphere because BC concentrations decrease faster than brown carbon in  
297 the high-altitude atmosphere (Liu et al., 2014).

298 Our results showed that the relative radiative forcing caused by DOC relative to BC ranged from  
299 2.1 % to 30.4 % ( $9.5 \pm 8.4 \%$ ) for snowpit samples and from 0.01 % to 0.5 % ( $0.1 \pm 0.1\%$ ) for surface

300 ice samples (Fig. S4). The high radiative forcing ratio of snowpit samples was caused by its higher  
301 DOC/BC (0.65) than that of surface ice (0.012) (Fig. S5), and the low ratio of DOC/BC in surface ice  
302 was caused by enrichment of BC in surface glacier ice during the intensive ablation period (Xu et al.,  
303 2009). Snowpit samples can be approximately considered to be fresh snow; thus, it is concluded that  
304 radiative forcing caused by DOC is a non-ignorable contributor in addition to BC in reducing the  
305 albedo of a glacier when the glacier is covered by fresh snow.

### 306 **3.4 DOC export during the melt season**

307 The two-year average discharge-weighted DOC concentration was  $238 \pm 96 \mu\text{g L}^{-1}$  during the  
308 melting period, comparable with the proglacial streamwater of Mount Nyainqentanglha glacier in the  
309 southern TP (Spencer et al., 2014). Seasonally, high DOC concentrations appeared during the low  
310 discharge periods (May to July and September to October) (Fig. 7), suggesting that DOC  
311 concentrations were slightly enriched to some extent. However, there were no clear diurnal variations  
312 in the DOC concentrations with the discharge, suggesting that the discharge from different parts of the  
313 glacier was well mixed at the glacier terminus (Fig. S6).

314 The seasonal variations in DOC flux were similar to those of the discharge (Fig. 7), indicating  
315 that discharge (rather than DOC concentrations) played a dominant role in the DOC mass flux. Hence,  
316 the majority of the glacier DOC export occurred during the summer melting season. Over the whole  
317 melting season, the annual flux of DOC from LHG glacier was  $192 \text{ kg km}^{-2} \text{ yr}^{-1}$ , with peak DOC fluxes  
318 from mid-late July to late August (70 % of the annual flux). Combined with the value of BDOC  
319 determined above, at least  $3211 \text{ kg C yr}^{-1}$  was ready to be decomposed and returned to the atmosphere  
320 as  $\text{CO}_2$  within one month of its release, producing positive feedback in the global warming process.

321 When it comes to the entire TP, it is obvious that proglacial streamwater DOC concentrations  
322 (Table S2) showed similar spatial variation to that of snowpit DOC (Li et al., 2016), with high and low  
323 value appeared at north and south TP, respectively, reflecting good succession of proglacial  
324 streamwater DOC concentration to that of snowpit samples. Therefore, it was calculated that DOC flux  
325 in proglacial streamwater of the entire TP glacier was around 12.7-13.2 Gg C ( $\text{Gg} = 10^9 \text{ g}$ ) based on  
326 average proglacial streamwater DOC concentration of  $193 \mu\text{g L}^{-1}$  (Table S2) and annual glacial  
327 meltwater runoff in China of  $66\text{-}68.2 \text{ km}^3$  (Xie et al., 2006), which is higher than that of DOC  
328 deposition (5.6 Gg C) at glacial region of the TP (Li et al., 2016), agree well with the negative water  
329 balance of the glaciers of the TP. Therefore, the TP glaciers can be considered as a carbon source under

330 present environment condition.

#### 331 **4 Conclusions and implications**

332 The concentrations and light absorption characteristics of DOC on a typical glacier in the northern  
333 TP were reported in this study. The mean DOC concentrations of snowpit samples, fresh snow, surface  
334 ice and proglacial streamwater were  $332 \pm 132 \mu\text{g L}^{-1}$ ,  $229 \pm 104 \mu\text{g L}^{-1}$ ,  $426 \pm 270 \mu\text{g L}^{-1}$  and  $238 \pm 96$   
335  $\mu\text{g L}^{-1}$ , respectively. These values were slightly higher or comparable to those of other regions, such as  
336 the European Alps and Alaska (Singer et al., 2012; Fellman et al., 2015). DOC in the snowpit samples  
337 was significantly correlated with  $\text{Ca}^{2+}$ , a typical cation in mineral dust, indicating that mineral dust  
338 transported from adjacent arid regions made important contributions to DOC of the studied glacierized  
339 regions except autochthonous or *in situ* biological activities. In addition, the light absorption profile of  
340 the snowpit DOC was similar to that of dust from potential source deserts, providing further evidence  
341 of the influence of desert sourced mineral dust on snowpit DOC. Based on the previous published  
342 radiative forcing data of black carbon in snowpit of LHG (Ming et al., 2013), for the first time, it is  
343 estimated that the radiative forcing caused by snowpit DOC was  $0.43 \text{ W m}^{-2}$ , accounting for around 10 %  
344 of the radiative forcing caused by BC. Therefore, in addition to BC, DOC is also an important agent in  
345 terms of absorbing solar radiation in glacierized regions, especially when the glacier is covered by  
346 fresh snow. It has also been proven that water-insoluble organic carbon has stronger light absorption  
347 ability (Chen and Bond, 2010). Therefore, the total contribution of OC to light absorption in glacierized  
348 regions should be higher, which requires further study in the future. Wet deposition is the most  
349 effective way of removing carbonaceous matter from the atmosphere (Vignati et al., 2010), and the  
350 removal ratio of OC in remote areas is almost the same as that of BC after long-range transport from  
351 source regions (Garrett et al., 2011). Because snowpit samples directly reflect the wet and dry  
352 deposition of carbonaceous matter, it is assumed that the contribution of radiative forcing for water  
353 soluble organic carbon relative to BC in the atmosphere in glacierized regions should be close to that of  
354 the snowpit samples in this study.

355 Because proglacial streamwater from different parts of the glacier is well mixed, no clear diurnal  
356 variations in DOC concentrations have been found. Combined with discharge and the corresponding  
357 DOC concentration, it was calculated that approximate  $192.0 \text{ kg km}^{-2} \text{ yr}^{-1}$  of DOC was released from  
358 LHG glacier. It was also calculated that approximately 46.3 % of the DOC could be decomposed  
359 within 28 days; thus,  $3,211 \text{ kg C yr}^{-1}$  would return to the atmosphere as  $\text{CO}_2$ , producing positive

360 feedback in the warming process. Although the flux of DOC from the studied glacier is small, when it  
361 comes to the entire TP DOC flux from glaciers of the total TP was around 12.7-13.2 Gg C.

362

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**Table 1.** Sampling information for snow, ice and proglacial streamwater in this study.

Sample type	Sampling time	Resolution <sup>*</sup>	Sampling site	Number (n)	Index
Snowpit	30th July, 2014	5 cm	4989 m	15	DOC, absorbance, ions
Snowpit	25th August, 2015	5 cm	5050 m	23	DOC, absorbance, ions
Surface fresh snow	4th August, 2014	100 m	4450-4900 m	18	DOC
Surface ice	6th August, 2014	100 m	4350-4900 m	20	DOC
Surface snow	16th July, 2015	50 m	4350-4850 m	11	DOC
Surface ice	15th August, 2015	50 m	4350-4850 m	11	DOC
Surface ice	25th August, 2015	50 m	4350-4600 m	6	DOC, absorbance
Subsurface ice	25th August, 2015	50 m	4350-4600 m	5	DOC, absorbance
Proglacial streamwater	29th-30th July, 2014	2h (day),4h (night)	4210 m	17	DOC
Proglacial streamwater	20th May-9th October, 2015	Every day	4210 m	184	DOC

532 <sup>\*</sup> for snowpit it is vertical resolution, for surface ice and snow it is horizontal distance.

533

534

535 **Table 2.** Comparison of DOC concentrations in snow, ice and proglacial streamwater from the glacier in this study  
 536 and glaciers in other regions.

Sites	DOC concentration ( $\mu\text{g L}^{-1}$ )	Sample types	References
Laohugou glacier (LHG)	332 $\pm$ 132	Snowpit	This study
Tanggula glacier (TGL)	217 $\pm$ 143	Snowpit	Yan et al. (2015)
Mount Everest (EV)	153 $\pm$ 561	Snowpit	
Mendenhall Glacier, Alaska	190	snowpit	Stubbins et al. (2012)
Greenland ice sheet	401 - 57	Snowpit	Hagler et al. (2007)
Laohugou glacier (LHG)	229 $\pm$ 104	Surface snow	This study
Greenland ice sheet	111	Surface snow	Hagler et al. (2007)
Juneau Icefield, Southeast Alaska	100 - 300	Fresh snow/snowpits	Fellman et al. (2015)
Laohugou glacier (LHG)	426 $\pm$ 270	Surface ice	This study
Mount Nyainqentanglha Glacier	212	Glacier ice	Spencer et al. (2014)
Antarctic ice sheet	460 $\pm$ 120	Surface ice	Hood et al. (2015)
Alpine glacier	138 $\pm$ 96	Subsurface ice	Singer et al. (2012)
Laohugou glacier (LHG)	238 $\pm$ 96	Proglacial streamwater	This study
Mount Nyainqentanglha Glacier	262	Proglacial streamwater	Spencer et al. (2014)
Mendenhall Glacier, Alaska	380 $\pm$ 20	Proglacial streamwater	Stubbins et al. (2012)

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539 **Table 3.** Mass absorption cross section (MAC) and Absorption Ångström Exponent (AAE<sub>330-400</sub>) of ice and snow  
 540 from LHG glacier and aerosols from other regions.

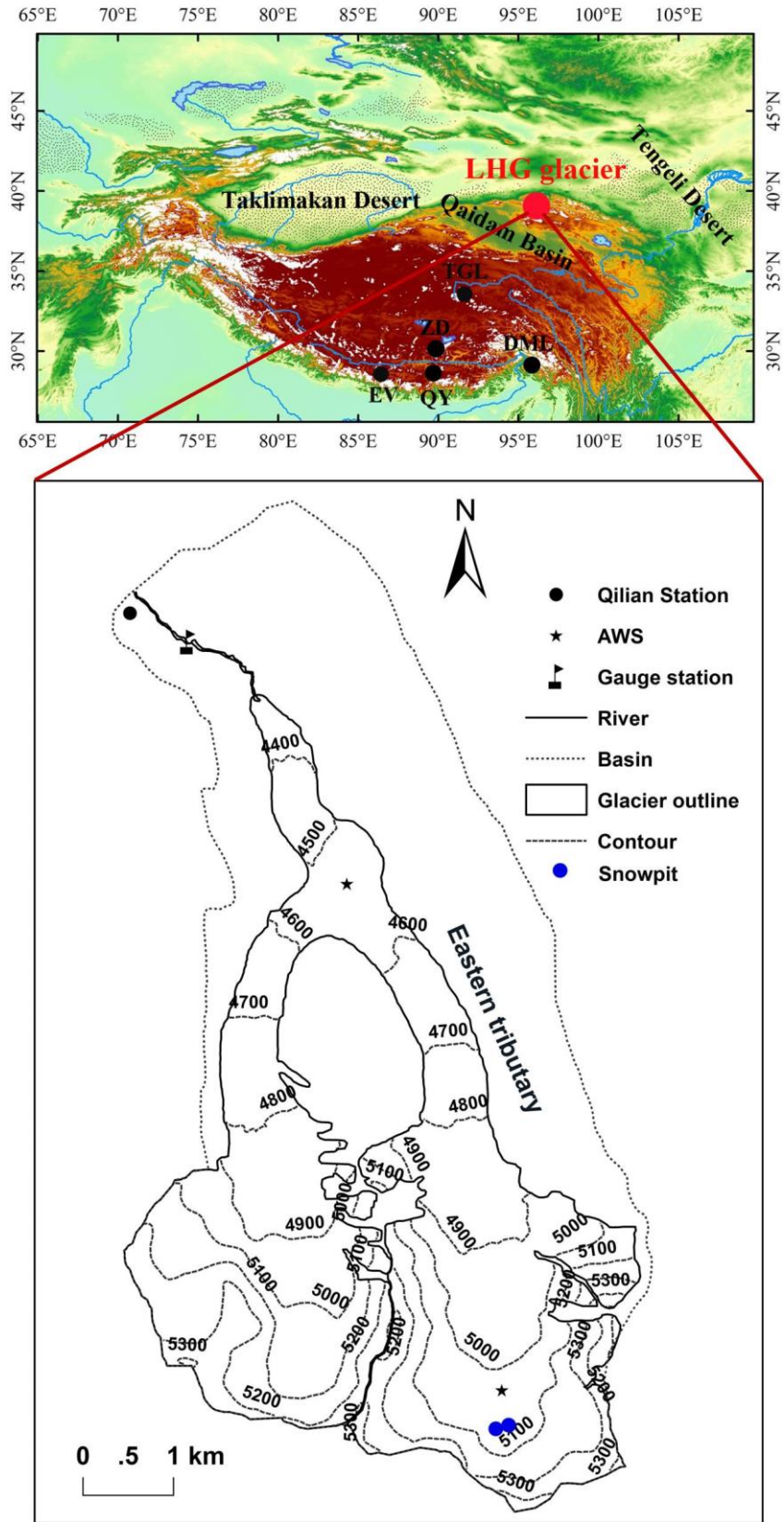
Site/Source	MAC (m <sup>2</sup> g <sup>-1</sup> )	AAE <sub>330-400</sub>	λ (MAC)	References
LHG glacier	1.4 ± 0.4 (snow)	5.0 ± 5.9 (snow)	365	This study
	1.3 ± 0.7 (ice)	3.4 ± 2.7 (ice)		
Biomass smoke	5.0	4.8	350	Kirchstetter et al. (2004)
Secondary organic aerosols	0.001 - 0.088	5.2 - 8.8	405	Lambe et al. (2013)
Wood smoke	0.13 - 1.1	8.6 - 17.8	400	Chen and Bond (2010)
HULIS, Arctic snow	2.6 ± 1.1	6.1 <sup>*</sup>	250	Voisin et al. (2012)
Beijing, China (winter)	1.79 ± 0.24	7.5	365	Cheng et al. (2011)
Beijing, China (summer)	0.71 ± 0.20	7.1	365	Cheng et al. (2011)

541 <sup>\*</sup> the wavelength range for AAE of this study is 300 - 550 nm

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544 Figure 1. Location map of LHG glacier No. 12.  
545 Figure 2. Average DOC concentrations of ice, snow and proglacial streamwater for LHG glacier.  
546 Figure 3. Variation in DOC concentrations in profiles of studied snowpits. The gray rectangles are dirty layers.  
547 Figure 4. Exponential decreases in DOC concentrations during the biodegradation experiment. Note: The blue point  
548 is calculated using equations derived from the experimental data (black point).  
549 Figure 5. Absorption spectra for DOC in snow and ice of LHG glacier and the dust from surrounding areas.  
550 Figure 6. Comparison of DOC concentrations (A) and  $MAC_{365}$  (B) between surface and subsurface ice.  
551 Figure 7. The discharge, DOC concentrations and fluxes exported from LHG glacier. Note: The concentrations  
552 with error bars include more than one sample on that day.  
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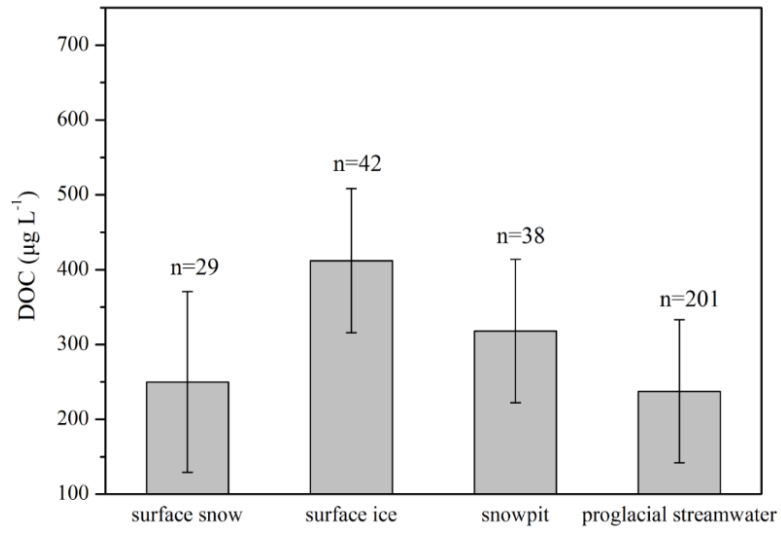


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Figure 1. Location map of Laohugou glacier No. 12.



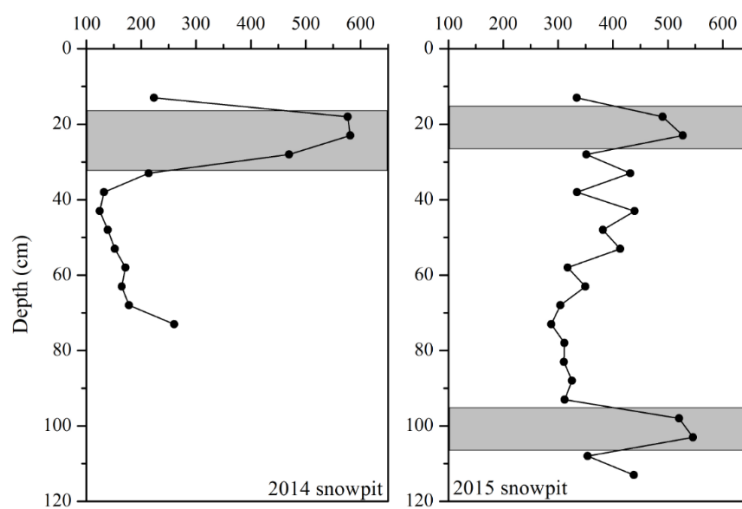
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**Figure 2.** Average DOC concentrations of ice, snow and proglacial streamwater for LHG glacier.

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563 **Figure 3.** Variation in DOC concentrations in profiles of studied snowpits. The gray rectangles are dirty  
564 layers.

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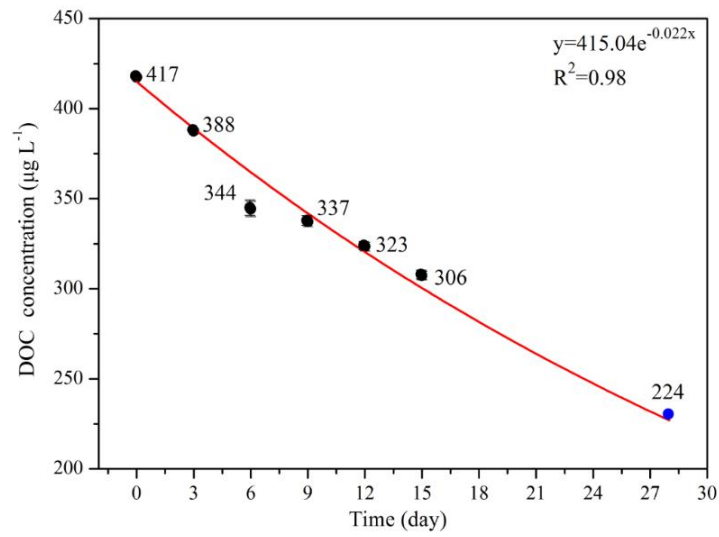
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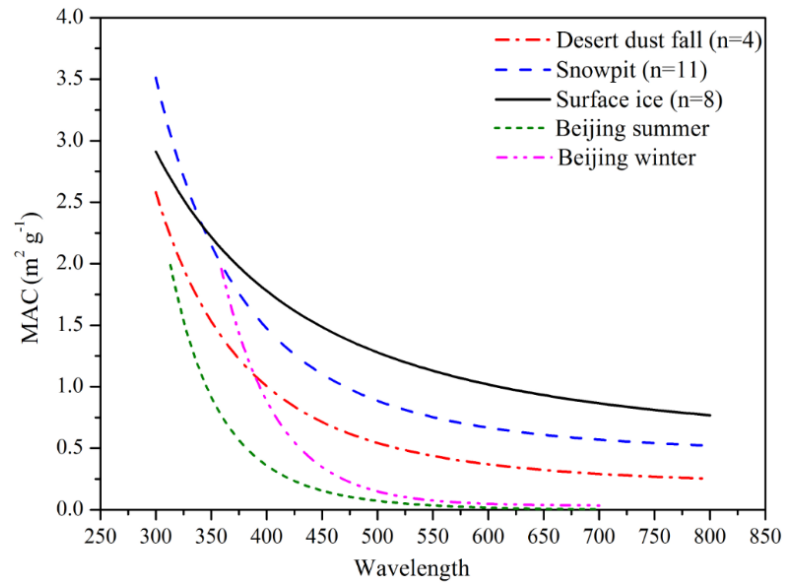
577 **Figure 4.** Exponential decreases in DOC concentrations during the biodegradation experiment. Note: The blue  
578 point is calculated using equations derived from the experimental data (black point). Mean values  $\pm$  standard

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deviations of duplicate treated samples are presented.

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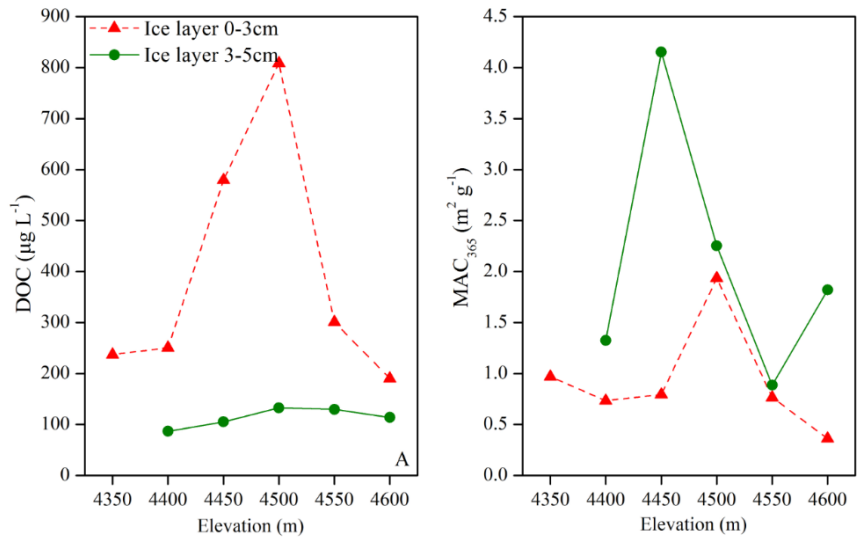
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**Figure 5.** Absorption spectra for DOC in snow and ice of LHG glacier and the dust from surrounding areas.

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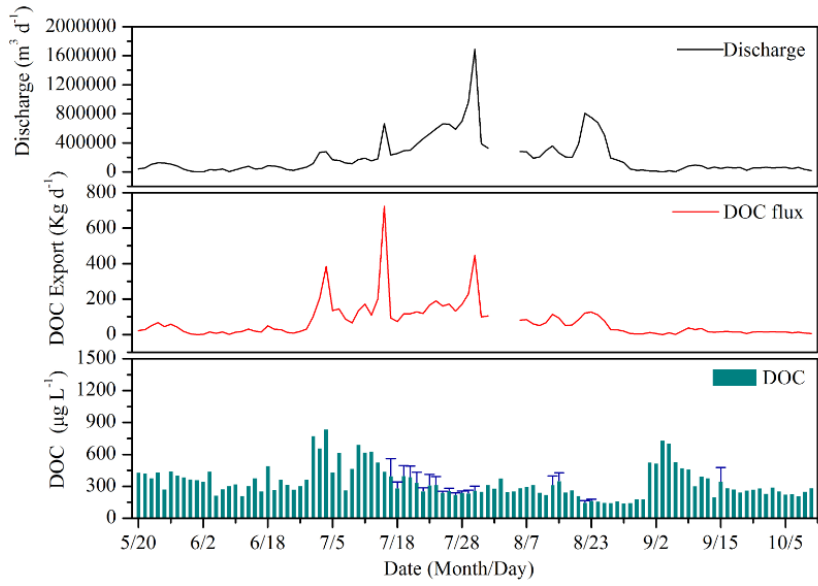
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**Figure 6.** Comparison of DOC concentrations (A) and MAC<sub>365</sub> (B) between surface and subsurface ice.

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592 **Figure 7.** The discharge, DOC concentrations and fluxes exported from LHG glacier. Note: The concentrations

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with error bars include more than one sample on that day.

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