

Dear Stephan Gruber,

We are grateful for your comments and suggestions on the manuscript. On behalf of my co-authors, I submit the revised version with the title "*Dissolved organic carbon (DOC) in Arctic ground ice*".

Attached you will find:

- response to Editor Review
- marked-up manuscript version

You will find all replies or changes that have been made below. Reviewer comments are cited in red font. Thank you very much for your time and efforts in this process and we look forward to seeing this manuscript accepted soon. Please, do not hesitate to contact me if you need further formation.

Best regards,  
Michael Fritz

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#### Response to Editor Review

Dear authors,

One reviewer asked about the difference between Holocene and late Pleistocene ice wedges. You have replied, but not included any changes in the text. It may be good to clarify at one place in your manuscript what the difference is, and possibly, why you chose a certain definition for late Pleistocene.

Reviewer#2 asked about the use of the terms "late glacial" and "late Pleistocene" and whether they mean the same. Our answer was that the *late Pleistocene* (MIS 2-5) spans a longer time than the *Lateglacial* (Bølling to Younger Dryas) and that in some occasions we had the possibility to narrow the age of the studied ice bodies. It is important to note that climate conditions ameliorated during the waning stages of the last glacial period so that thermokarst already started and vegetation changed before the onset of the Holocene. As this is important to carbon cycling, we added a clear differentiation of the discussed periods to the text:

**p6 L6:** Behind late Pleistocene we added the explanation "(Marine isotope stages 2-5)".

**p6 L16:** A footnote explains the term "Lateglacial".

"We refer to the Lateglacial as a stratigraphic and geochronological period at the transition between the Pleistocene and the Holocene. The Lateglacial spans the latest part of the Late Weichselian / Late Wisconsin glacial period. It includes the Bølling, the Older Dryas, the Allerød, and the Younger Dryas, from ca. 14,700 to 11,600 years before present (cf. de Klerk, 2004)."

- reference for *de Klerk (2004)* was added

P13L9 You write “We therefore suggest that incorporating DOC from non-massive ground-ice types would lead to a significant rise in DOC stocks in permafrost of at least one order of magnitude.” This implies that previous analyses excluded the pore-ice carbon. Is this really true?

Previous studies have never differentiated the permafrost organic carbon (OC) pool into particulate organic carbon (POC) and dissolved organic carbon (DOC). However, we try to make the point that such a differentiation would be useful in the future because the dissolved OC fraction, which is either in the ground ice or leached upon thaw, is easily available for microorganisms and easily transportable into remote ecosystems.

In the manuscript we estimate DOC (45.2 Tg) pools in Yedoma areas. This number is based on ice wedges only because we have concentrated on massive ice types. However, Pleistocene Yedoma-type deposits have a volumetric ground-ice content of 60-82 vol% (see Introduction), but only 48 vol% on average account for ice wedges (p.12). The rest is non-massive intrasedimental ice containing a so far unknown amount of DOC, which is usually combined with POC into permafrost OC due to the nature the analyses are currently done. We would like to keep the sentence because it is neither right nor wrong; it depends on the perspective and the reason to study permafrost carbon pools.

P4L5 delete “as mentioned above”

Changed accordingly.

P5L1 put the “and” at the end of the previous bullet

Changed accordingly.

P13L4 Please do not include the “unpublished values” as this would make them de-facto published without providing reproducibility. At least, do not show the numbers. You could just say “...have been found to be higher...”.

Changed accordingly.

“DOC concentrations in non-massive intrasedimental ice from Muostakh Island (Siberia) and the Yukon Coast (Canada) have been found to be much higher (Fritz, unpublished data).”

P17L6 “is accompanied”

Changed accordingly.

1 **Dissolved organic carbon (DOC) in Arctic ground ice**

2

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4 **Lantuit<sup>1,2</sup>**

5

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10

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12

1 **Abstract**

2 Thermal permafrost degradation and coastal erosion in the Arctic remobilize  
3 substantial amounts of organic carbon (OC) and nutrients which have been accumulated in  
4 late Pleistocene and Holocene unconsolidated deposits. Permafrost vulnerability to thaw  
5 subsidence, collapsing coastlines and irreversible landscape change is largely due to the  
6 presence of large amounts of massive ground ice such as ice wedges. However, ground ice  
7 has not, until now, been considered to be a source of dissolved organic carbon (DOC),  
8 dissolved inorganic carbon (DIC) and other elements, which are important for ecosystems and  
9 carbon cycling. Here we show, using biogeochemical data from a large number of different  
10 ice bodies throughout the Arctic, that ice wedges have the greatest potential for DOC storage  
11 with a maximum of 28.6 mg L<sup>-1</sup> (mean: 9.6 mg L<sup>-1</sup>). Variation in DOC concentration is  
12 positively correlated with and explained by the concentrations and relative amounts of  
13 typically terrestrial cations such as Mg<sup>2+</sup> and K<sup>+</sup>. DOC sequestration into ground ice was more  
14 effective during the late Pleistocene than during the Holocene, which can be explained by  
15 rapid sediment and OC accumulation, the prevalence of more easily degradable vegetation  
16 and immediate incorporation into permafrost. We assume that pristine snowmelt is able to  
17 leach considerable amounts of well-preserved and highly bioavailable DOC as well as other  
18 elements from surface sediments, which are rapidly frozen and stored in ground ice,  
19 especially in ice wedges, even before further degradation. We found that ice wedges in the  
20 Yedoma region represent a significant DOC (45.2 Tg) and DIC (33.6 Tg) pool in permafrost  
21 areas and a fresh-water reservoir of 4,200 km<sup>3</sup>. This study underlines the need to discriminate  
22 between particulate OC and DOC to assess the availability and vulnerability of the permafrost  
23 carbon pool for ecosystems and climate feedback upon mobilization.

24

## 1 **1 Introduction**

2 Vast parts of the coastal lowlands of Siberia, Alaska and Canada consist of  
3 unconsolidated organic-rich fine-grained deposits. These sediments, that occur as glacial  
4 and Yedoma-type sediments (including their degradation forms as thermokarst), are  
5 characterized by high ground-ice contents, both on a volumetric (vol%) and gravimetric  
6 (weight %) basis (Brown et al., 1997; Zhang et al., 1999; Grosse et al., 2013; Schirrneister et  
7 al., 2013). Yedoma deposits, which formed during the late Pleistocene cold stages in  
8 unglaciated Beringia (Schirrneister et al., 2013), for instance, are characterized by absolute  
9 ground-ice contents, excluding ice wedges, of 40-60 weight % (Schirrneister et al., 2011c).  
10 Ice wedges are one of the most common types of ground ice in permafrost. They form when  
11 thermal contraction cracks open in winter, which are periodically filled with snow meltwater  
12 in spring that quickly (re)freezes at negative ground temperatures to form ice veins and finally  
13 vertically foliated ice wedges. The ice wedges are themselves characterized by volumetric ice  
14 contents closing 100 vol% and make much of the subsurface in these Yedoma deposits.  
15 Recent calculations of ice-wedge volumes in East Siberian Pleistocene Yedoma and Holocene  
16 thermokarst deposits show contents of 48 vol% and 7 vol%, respectively (Strauss et al.,  
17 2013). Combining ice wedges and other ice types in Yedoma deposits gives a mean  
18 volumetric ground-ice content for those regions between 60 and 82 vol% (Zimov et al.  
19 2006a, b; Schirrneister et al., 2011b, c; Strauss et al., 2013). High ground-ice contents are  
20 also typical for coastal Alaska (43-89 vol%; Kanevskiy et al., 2011, 2013) and the western  
21 Canadian Arctic (50-60 vol%; French, 1998). The presence of massive ice (i.e. gravimetric ice  
22 content >250% on dry soil weight basis; cf. van Everdingen, 1998) and excess ice, which is  
23 visible ice that exceeds the pore space, is the key factor for the vulnerability of permafrost to  
24 warmer temperatures and mechanical disturbance, as ice melt will initiate surface subsidence  
25 and thermal collapse, also known as thermokarst (Czudek and Demek, 1970).

26 Permafrost soils hold approximately 50% of the global soil carbon pool (Tarnocai et al., 2009;  
27 Hugelius et al., 2014), mostly as particulate organic carbon (POC). These calculations of  
28 permafrost OC stocks, however, subtract the ground-ice content (Zimov et al. 2006a, b;  
29 Tarnocai et al. 2009; Strauss et al., 2013; Hugelius et al., 2013, 2014) and therefore disregard  
30 the OC, especially the amount of dissolved organic carbon (DOC), contained in large ground-  
31 ice bodies such as ice wedges and other types of massive ice. Although these numbers might  
32 be small compared to the POC stocks in peat and mineral soils, DOC from permafrost is  
33 chemically labile (Dou et al., 2008; Vonk et al., 2013a, b) and may directly enter local food

1 webs. Due to its lability, DOC can become quickly mineralized by microbial communities  
2 and photochemical reactions (Battin et al., 2008; Vonk et al., 2013a, b; Cory et al., 2014) and  
3 returned to the atmosphere when released due to permafrost degradation (Schuur et al., 2009;  
4 Schuur and Abbot, 2011).

5 ~~As mentioned above,~~ several studies have shed light on the POC stocks contained in  
6 permafrost (e.g. Zimov et al., 2006a; Tarnocai et al., 2009; Schirrmeister et al., 2011b; Strauss  
7 et al., 2013; Hugelius et al., 2013, 2014; Walter Anthony et al., 2014) and how much of these  
8 stocks is potentially mobilized due to thermal permafrost degradation and coastal erosion  
9 (Rachold et al., 2004; Jorgenson and Brown, 2005; Lantuit et al., 2009; McGuire et al., 2009;  
10 Ping et al., 2011; Schneider von Deimling et al., 2012; Vonk et al., 2012; Günther et al., 2013;  
11 2015; Wegner et al., in review). DOC fluxes have also been quantified in western Siberian  
12 catchments (Frey and Smith, 2005) and monitoring efforts of the large rivers draining  
13 permafrost areas and entering into the Arctic Ocean have provided robust estimations of the  
14 riverine DOC export (Raymond et al., 2007; McGuire et al., 2009). However, DOC stocks in  
15 permafrost ground ice and the resulting potential DOC fluxes in response to coastal erosion  
16 and thermal degradation are still unknown (Guo et al., 2007; Duo et al., 2008). At this  
17 moment, any inference about DOC stocks in permafrost and fluxes from permafrost is derived  
18 from measurements in secondary systems such as lake (e.g. Kling et al., 1991; Walter  
19 Anthony et al., 2014), river (e.g. Benner et al., 2004; Finlay et al., 2006; Guo et al., 2007;  
20 Raymond et al., 2007; Holmes et al., 2012) and ocean waters (e.g. Opsahl and Benner, 1997;  
21 Dittmar and Kattner, 2003; Cooper et al., 2005) or from laboratory experiments (Dou et al.,  
22 2008). In contrast, the purpose of this study was to sample and measure DOC at the source  
23 (i.e. ground ice in permafrost) directly, before it gets altered by natural processes such as  
24 exposition to the atmosphere, lithosphere and hydrosphere.

25  
26 Here, we present an Arctic-wide study on DOC stocks in ground ice, aiming at incorporating  
27 massive ground ice into the Arctic permafrost carbon budget. The specific objectives of our  
28 study are:

- 29 • to quantify DOC contents in different massive ground ice types,
- 30 • to calculate DOC stocks in massive ground ice at the Arctic level,
- 31 • to put ground-ice-related DOC stocks into the context of the terrestrial Arctic OC  
32 pools and fluxes, and

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- ~~and~~ to introduce relationships between organic and inorganic geochemical parameters, stable water isotopes, stratigraphy, and genetic and spatial characteristics to shed light on the origin of DOC and the processes of carbon sequestration in ground ice.

## 2 Study area and study sites

This study was carried out along the coastal lowlands of east Siberia, Alaska and northwest Canada (Fig. 1). All study sites, except for the Fairbanks area, are located within the zone of continuous permafrost. The sites cover a wide and representative range of geomorphological settings, terrain units and ground-ice conditions (Table 1). All studied ground-ice bodies were found in ice-rich unconsolidated Holocene and late Pleistocene (Marine isotope stages 2-5) deposits. Outcrops in permafrost were either accessible due to strong rates of coastal erosion along the ice-rich coasts forming steep exposures (Forbes, 2011) or were technically constructed for research purposes such as the CRREL Permafrost Tunnel in Barrow or for mining such as the Vault Creek Tunnel near Fairbanks, Alaska. Coastal outcrops in Siberia were dominated by large late Pleistocene ice wedges reaching up to 20 m in depth and up to 6 m in width (Schirrneister et al., 2011c). They formed syngenetically during periods of rapid sedimentation of Ice Complex deposits, also known as Yedoma (Schirrneister et al., 2013). Holocene epigenetic and syngenetic ice wedges of 1 – 6 m in depth and <1.0 – 3.5 m in width were encountered in exposed thermokarst depressions of Late-glacial<sup>1</sup> to Holocene origin and within the Holocene peaty cover deposits. Besides ice wedges, other types of massive ground ice were sampled, such as buried glacier ice, buried lake ice and a fossil snow patch (Fig. 2). In some cases, massive ground ice occupied as much as 90 vol% of 40 m coastal exposures eroding up to 10 m a<sup>-1</sup> (Lantuit et al., 2012). The focus of this paper is on massive ground ice; non-massive ice (in particular pore ice and intrasedimental ice such as ice lenses) was excluded from this first attempt to calculate DOC stocks in ground ice, because of the complex genetic processes associated with the interaction with enclosing sediment and the relatively small amount of ice relative to massive ice bodies. DOC in intrasedimental ice is, however, not considered to be insignificant.

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<sup>1</sup> We refer to the Lateglacial as a stratigraphic and geochronological period at the transition between the Pleistocene and the Holocene. The Lateglacial spans the latest part of the Late Weichselian / Late Wisconsin glacial period. It includes the Bølling, the Older Dryas, the Allerød, and the Younger Dryas, from ca. 14,700 to 11,600 years before present (cf. de Klerk, 2004).



## 1 **3 Material and methods**

### 2 **3.1 Laboratory analyses**

3 A total number of 101 ice samples from 29 ice bodies and 3 surface water samples  
4 from 3 thermokarst lakes were studied. Ice blocks were cut with a chain saw in the field and  
5 kept frozen until further processing with a band saw in a cold lab at  $-15^{\circ}\text{C}$  for removal of  
6 partially melted margins and cleaning of the edges. Samples  $\geq 50$  ml were thawed at  $4^{\circ}\text{C}$  in  
7 pre-cleaned (purified water) glass beakers covered with pre-combusted aluminium foil  
8 ( $550^{\circ}\text{C}$ ). Meltwater was filtered with gum-free syringes equipped with glass fibre filters  
9 (Whatman<sup>TM</sup> GF/F; pore size:  $0.7\ \mu\text{m}$ ) and acidified with  $20\ \mu\text{l}$   $\text{HCl}_{\text{suprapur}}$  (30%) to  $\text{pH}<2$  in  
10 order to prevent microbial conversion. DOC concentrations ( $\text{mg L}^{-1}$ ) were measured with a  
11 high-temperature ( $680^{\circ}\text{C}$ ) combustion total organic carbon analyzer (Shimadzu TOC-V<sub>CPH</sub>).  
12 Internal acidification is used to convert inorganic carbon into  $\text{CO}_2$ , which is stripped out of  
13 solution. Non-purgeable organic carbon compounds are combusted and converted to  $\text{CO}_2$  and  
14 measured by a non-dispersive infrared detector (NDIR). The device-specific detection limit is  
15  $0.4\ \mu\text{g L}^{-1}$ . For each sample, one measurement with three to five repetitions was performed  
16 and results were averaged.

17 Further analyses for hydrochemical characterization included pH, electrical conductivity,  
18 major anions and cations, and stable water isotopes ( $\delta^{18}\text{O}$ ,  $\delta\text{D}$ ). Stratigraphic investigations  
19 and stable water isotopes were used to differentiate between genetic ice types and to assess  
20 their approximate age (i.e. Holocene and late Pleistocene). Analyses of  $\delta^{18}\text{O}$  and  $\delta\text{D}$  were  
21 carried out with a mass spectrometer (Finnigan MAT Delta-S) using the water-gas  
22 equilibration technique (for further information see [Horita et al., 1989](#); [Meyer et al., 2000](#)).  
23 The isotopic composition is expressed in delta per mil notation ( $\delta$ , ‰) relative to the Vienna  
24 Standard Mean Ocean Water (VSMOW) standard. The reproducibility derived from long-  
25 term standard measurements is established with  $1\sigma$  better than  $\pm 0.1$  ‰ for  $\delta^{18}\text{O}$  and  $\pm 0.8$  ‰  
26 for  $\delta\text{D}$  ([Meyer et al., 2000](#)). Samples for ion analysis were passed through cellulose-acetate  
27 filters (Whatman<sup>TM</sup> CA; pore size  $0.45\ \mu\text{m}$ ). Afterwards, samples for the cation analyses were  
28 acidified with  $\text{HNO}_3_{\text{suprapur}}$  (65%) to prevent microbial conversion processes and adsorptive  
29 accretion, whereas samples for anion analyses were kept cool. The cation content was  
30 analysed by Inductively Coupled Plasma-Optical Emission Spectrometry (ICP-OES, Perkin-  
31 Elmer Optima 3000 XL), while the anion content was determined by Ion Chromatography  
32 (IC, Dionex DX-320). Hydrogen carbonate concentrations were measured by titration with

1 0.01 M HCl using an automatic titrator (Metrohm 794 Basic Titrino). Based on  $\text{HCO}_3^-$   
2 concentrations we approximated the dissolved inorganic carbon (DIC) concentrations using  
3 the molecular weights.

## 4 5 **3.2 Statistical methods**

### 6 **3.2.1 Principal Component Analysis (PCA)**

7 Principal component analysis (PCA) was used to summarize the variation in a biplot  
8 by reducing dimensionality of the data while retaining most of the variation in the data set  
9 (Jolliffe, 2002). Ordinally scaled variables (i.e. chemical data set) were log-transformed,  
10 centered and standardized except for pH,  $\delta^{18}\text{O}$ ,  $\delta\text{D}$ , latitude, and longitude not being log-  
11 transformed due the inter-sample invariance. Ice types (ice wedge, buried lake ice, basal  
12 glacier ice, snow pack ice, surface water), relative age (Pleistocene, Holocene, recent) were  
13 coded with dummy variables and were superimposed as inactive supplementary variables on  
14 the ordination plot to enable rough assumptions about the relationship between chemical  
15 composition, ground-ice formation and age. The whole data set was reduced to 92 samples  
16 and 23 variables by removing those containing missing values. PCA was performed with  
17 focusing on inter-species correlation and was implemented using CANOCO 4.5 software for  
18 Windows (ter Braak and Šmilauer, 2002).

### 19 20 **3.2.2 Univariate Tree Models (UTM)**

21 A powerful tool to explore the relationship between a single continuous response  
22 variable (DOC concentration) and multiple explanatory variables is a regression tree (Zuur et  
23 al., 2007). Tree models perform well with non-linearity and interaction between explanatory  
24 variables. UTM is used to find interactions missed by other methods and also indicate the  
25 relative importance of different explanatory variables. UTM was performed using the  
26 computing environment R and Brodgar 2.6.5 software for Windows (ter Braak and Šmilauer,  
27 2002).

## 1 4 Results

### 2 4.1 DOC and DIC concentrations

3 [Table 2](#) provides an overview of mean DOC and DIC concentrations and range for  
4 each ground-ice type. We found strong variations of DOC concentrations within and across  
5 individual ground-ice types. The highest DOC concentrations were found in ice wedges with  
6 a mean of 9.6 mg L<sup>-1</sup> and a maximum of 28.6 mg L<sup>-1</sup>. Late Pleistocene ice wedges were  
7 characterized by higher mean DOC concentrations than Holocene ones with 11.1 and  
8 7.3 mg L<sup>-1</sup>, respectively. Other ice types had average DOC concentrations between 1.8 and  
9 3.0 mg L<sup>-1</sup> and their range was narrower than in ice wedges ([Table 2](#), [Fig. 3](#)). Modern surface  
10 water gave DOC values between 5.5 and 5.8 mg L<sup>-1</sup>.

11 The highest DIC concentrations were found in modern surface water with on average  
12 22.6 mg L<sup>-1</sup> and a maximum of 40.2 mg L<sup>-1</sup> ([Table 2](#), [Fig. 3](#)). DIC concentrations were lower  
13 in ground ice but varied strongly across ice types. With 8.5 mg L<sup>-1</sup> late Pleistocene ice wedges  
14 were characterized by almost four times higher mean DIC concentrations than Holocene ones  
15 (2.2 mg L<sup>-1</sup>; [Fig. 3](#)). Buried glacier and lake ice had similar mean DIC concentrations (around  
16 9 mg L<sup>-1</sup>) but showed large ranges; from values around zero up to 25 mg L<sup>-1</sup>. Basal glacier  
17 ice, buried lake ice, and snow pack ice show mean DOC concentrations between 1.8 and  
18 3.0 mg L<sup>-1</sup>. For individual sample values see [Supplement Table S1](#).

19

### 20 4.2 Correlation matrix

21 With the help of a correlation matrix environmental processes and chemical  
22 relationships can be visualized that may help to explain the sequestration of DOC into ground  
23 ice. Pearson's correlation coefficients were calculated and plotted in a correlation matrix in  
24 order to assess the degree of association between DOC, chemical properties, stable water  
25 isotopes and spatial variables ([Fig. 4](#)). A strong positive correlation suggests a mutual driving  
26 mechanism whereas negative values imply an inverse association. Most importantly, DOC is  
27 positively related to the relative proportion of Mg<sup>2+</sup> in the cation spectrum (R=0.65). Further  
28 positive relations between DOC and other parameters, although less pronounced, involve K<sup>+</sup>  
29 (R=0.36), HCO<sub>3</sub><sup>-</sup> (R=0.36) and latitude (R=0.38). The only significantly negative relationship

1 with regard to DOC exists together with  $\text{Na}^+$  ( $R=-0.44$ ) (Fig. 4). Climate-driven parameters  
2 such as  $\delta^{18}\text{O}$ ,  $\delta\text{D}$ , and D-excess do not explain DOC concentrations.

3

#### 4 **4.3 Principal components**

5 The first two axes of the PCA explain 43.9% of the variation in the data (Fig. 5).  $\text{Cl}^-$   
6 and  $\text{Na}^+$  ions are positively correlated with the first axis in descending order of correlation,  
7 whereas  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ , and  $\text{HCO}_3^-$  ions and pH are negatively correlated. Parameters positively  
8 correlated with PCA axis 2 include information on the ice origin as Pleistocene and basal  
9 glacier ice. In contrast,  $\delta\text{D}$ ,  $\delta^{18}\text{O}$ , DOC concentration, and information on the ice origin as ice  
10 wedge and Holocene ground ice are negatively correlated with PCA axis 2. Variations in  
11  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$  concentration as well as information on latitude and longitude are not  
12 correlated with the first two PCA axes. The separation of ice samples in the PCA ordination  
13 plot leads to three distinct groups: (1) Holocene ice wedges and recent surface water samples  
14 are entirely negatively related to the second axis, whereas (2) Pleistocene ice wedges are  
15 entirely negatively related to the first axis. (3) Pleistocene basal glacier ice and buried lake ice  
16 is positively related to the second axis. This separation might be related to the different  
17 processes of ice formation and climate variation.

18  $\text{Na}^+$  and  $\text{Cl}^-$ -dominated samples represent Holocene ice wedges from coastal cliffs in east  
19 Siberia (Muostakh Island and Oyogos Yar). The majority of ice wedges with a terrestrial ion  
20 composition ( $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{HCO}_3^-$ ) are of late Pleistocene age in areas such as Mamontov  
21 Klyk, Bol'shoy Lyakhovsky Island, Yukon Coast and the Fairbanks area. The first axis  
22 probably separates samples with a strong marine impact at its upper end from those with more  
23 of a continental background. The second axis might represent climate conditions of formation.  
24 The majority of Pleistocene ice samples with a depleted stable water isotope composition  
25 show positive sample scores whereas Holocene ground ice being enriched in heavy stable  
26 water isotopes mostly shows negative sample scores and therefore plots in the lower part of  
27 the PCA (Fig. 5).

28

#### 1 **4.4 Univariate Tree Model (UTM)**

2 UTM (Fig. 6a) shows that differences in DOC concentrations can be explained according to  
3 inorganic geochemical properties. The first two nodes split on  $Mg^{2+}$  with a threshold value of  
4 16% of the cation spectrum. The next nodes split according to thresholds in  $K^+$  of 2.30 and  
5 2.65%, respectively (Fig. 6a). Threshold percentages presented here are based on the cation  
6 spectrum only. This means that all measured cations sum up to 100 %. This is statistically  
7 more robust than using individual sample concentrations which can have different  
8 magnitudes. We end up with four statistically significant groups (i.e. nodes) with different  
9 mean DOC concentrations ( $mg\ L^{-1}$ ) of each group, also showing the number of observations  
10 in each group (n). With the UTM information – that inorganic geochemistry explains the  
11 variability in DOC concentration – we can make assumptions about relations between carbon  
12 sequestration in different water types. DOC concentration is not independent from inorganic  
13 geochemical composition. Cross validation (Fig. 6b) confirms statistical significance of the  
14 model result.

15

16

## 1 5 Discussion

### 2 5.1 DOC stocks in ground ice and relevance to carbon cycling

3 While the riverine DOC export to the Arctic Ocean has been estimated to 33-34 Tg a<sup>-1</sup>  
4 (McGuire et al. 2009; Holmes et al., 2012), comparable numbers for the DOC input by coastal  
5 erosion and thermal permafrost degradation (also known as thermokarst) are not available yet.  
6 This knowledge gap includes the DOC contribution derived from melting ground ice from  
7 ice-rich permafrost. Table 2 provides an overview of DOC contents in different massive  
8 ground-ice types from the North American and Siberian Arctic. Because of their wide spatial  
9 distribution in Arctic lowlands and the measured DOC concentrations, we conclude that from  
10 massive ground-ice types ice wedges hold the greatest potential for DOC storage with a  
11 maximum of 28.6 mg L<sup>-1</sup>. This is in good agreement with DOC measurements in a so far  
12 limited number of ice wedges by Douglas et al. (2011) in Alaska and Vonk et al. (2013b) in  
13 east Siberia who showed DOC concentrations of 18.4 – 68.5 mg L<sup>-1</sup> (n=5) and 8.8 – 15 mg L<sup>-1</sup>  
14 (n=3), respectively.

15 Ulrich et al., (2014) have calculated maximum wedge ice volumes (WIV), which range from  
16 31.4 to 63.2 vol% for late Pleistocene Yedoma deposits and from 6.6 to 13.2 vol% for  
17 Holocene thermokarst deposits in east Siberia and Alaska. Strauss et al. (2013) have shown  
18 similar averages for WIV of 48 vol% in late Pleistocene Yedoma and 7.0 vol% for Holocene  
19 thermokarst deposits. Together with average DOC concentrations of 11.1 mg L<sup>-1</sup> (max. 28.6)  
20 this would lead to 5.3 g DOC per m<sup>3</sup> (max. 18.1) for late Pleistocene ice wedges in the upper  
21 late Pleistocene permafrost column (Table 3) and a DOC pool of 43.0 Tg DOC based on  
22 416,000 km<sup>2</sup> of undisturbed Yedoma in Beringia and a mean thickness of 19.4 m (Strauss et  
23 al., 2013). DOC stocks in ice wedges in Holocene thermokarst deposits are much lower with  
24 on average 0.51 g m<sup>-3</sup> and a maximum of 2.6 g m<sup>-3</sup> due to much lower WIV (cf. Ulrich et al.,  
25 2014) and slightly lower DOC concentrations (Table 3). With on average 2.2 Tg the Holocene  
26 ice wedge DOC pool is much lower than the late Pleistocene pool, mainly because of lower  
27 WIV, an average thickness of 5.5 m for thermokarst deposits and despite their greater extent  
28 (775,000 km<sup>2</sup>) than undegraded Yedoma deposits (Strauss et al., 2013). Even stronger  
29 differences are characteristic for DIC pools in late Pleistocene ice wedges (32.9 Tg) compared  
30 to Holocene ice wedges (0.66 Tg) in the same areas. Based on the above-mentioned spatial  
31 coverage of Yedoma and thermokarst deposits including sediment thickness and WIV, we

1 conclude that in the study area ice wedges represent a significant DOC (45.2 Tg) and DIC  
2 (33.6 Tg) pool in permafrost areas and a fresh-water reservoir of 4,200 km<sup>3</sup> (see Table 3).  
3 However, all types of non-massive intrasedimental ice, raising the total ground-ice volume to  
4 ~80% (Schirmer et al., 2011b; Strauss et al., 2013), are still excluded. ~~Unpublished~~ DOC  
5 concentrations in non-massive intrasedimental ice from Muostakh Island (Siberia) and the  
6 Yukon Coast (Canada) have been found to be much higher ~~show overall averages of 327 and~~  
7 ~~100 mg L<sup>-1</sup>, respectively~~ (Fritz, unpublished data). Higher DOC concentrations in  
8 intrasedimental ice than in massive ice are certainly due to the long-term contact of soil  
9 moisture with soil organic matter prior to freezing. We therefore suggest that incorporating  
10 DOC from non-massive ground-ice types would lead to a significant rise in DOC stocks in  
11 permafrost of at least one order of magnitude. However, a differentiation between particulate  
12 and dissolved OC in permafrost is not done yet, although the technical means via rhizon soil  
13 moisture sampling is already available on a cost- and time-efficient basis. Nevertheless, we  
14 are aware of the fact that DOC makes up a limited proportion of the whole permafrost carbon  
15 stocks. A cautious estimation of the ratio of DOC and POC is in the order of ~1/2000 if we  
16 consider about 2 wt% total organic carbon (TOC) in sediments (e.g. Schirmer et al.,  
17 2011b,c; Strauss et al., 2013) and about 10 mg L<sup>-1</sup> DOC in massive ground ice. This ratio will  
18 become much smaller if POC and DOC in the whole permafrost column would be  
19 differentiated, because TOC comprises both POC and DOC.

20

## 21 5.2 Carbon sequestration and origin in relation to inorganic geochemistry

22 The origin and sequestration process into ground ice seems to play an important role in  
23 the magnitude and bioavailability of DOC. Sequestration of OC into ground ice is a complex  
24 process that is dependent on water source, freezing process, organic matter origin and  
25 inorganic geochemical signature of the ambient water to form ground ice.

26 Figures 4 and 6a show that the electrical conductivity (i.e. total ion content) of ground ice is  
27 unrelated to DOC but that the ion composition and therefore the ion source seems to be  
28 relevant. Mg<sup>2+</sup> and K<sup>+</sup> are the most significant parameters for explaining variations in DOC  
29 concentrations (Fig. 6a). Higher Mg<sup>2+</sup> and K<sup>+</sup> fractions of the cations spectrum are positively  
30 related to higher DOC concentrations (Fig. 4). We recognize that in the node (group) with the  
31 highest average DOC concentrations ( $\bar{x} = 11.9 \text{ mg L}^{-1}$ , n=40) we find most of the Pleistocene  
32 ice wedges and to a lesser extent Holocene ice wedges (Fig. 6a). All study areas are

1 represented here. Both,  $\text{Mg}^{2+}$  and  $\text{K}^+$  have typically high shares in terrestrial water types  
2 because Mg and K are major elements in clay minerals and feldspars. In combination with  
3 terrestrial  $\text{HCO}_3^-$  and  $\text{Ca}^{2+}$  the mobility of  $\text{Mg}^{2+}$  is high in  $\text{Mg}/\text{Ca}(\text{HCO}_3)_2$  solutions ([Gransee](#)  
4 [and Fühns, 2013](#)).

5

6 Ice wedges are fed by meltwater from atmospheric sources that have been in contact with  
7 vegetation and sediments of the tundra surface before meltwater infiltrated the frost cracks in  
8 spring. By contrast, glacier ice, buried snow bank ice, and lake ice is primarily fed by  
9 atmospheric waters having less interaction with carbon and ion sources. Yet, the yellowish  
10 brown to gray late Pleistocene and the milky-white Holocene ice wedges have incorporated  
11 sediments and organic matter that originates from surface soils and vegetation debris that was  
12 carried along with the meltwater into the frost crack (e.g. [Opel et al., 2011](#)). Spring snow melt  
13 water interacts with the soil material leaching out carbon as it trickles downward toward the  
14 ice wedges. Also, since wedges may take thousands of years to form and the location of their  
15 upper surface changes with time, there are numerous spatial and temporal ways that deeper  
16 soil pore waters can get incorporated into the wedge ice. Leaching of DOC from relatively  
17 young surface organic matter takes place ([Guo et al., 2007](#); [Lachniet et al., 2012](#)) as well as  
18 dissolution of ions from sediment particles. Snow melt feeding ice wedges strongly attracts  
19 leachable components because of its initial purity. This might be the reason why especially  
20 ice wedges contain relatively high amounts of bioavailable DOC with low-molecular weight  
21 compounds that can be old but remained fresh over millennia ([Vonk et al., 2013b](#)).

22

23 Principal component analysis clusters ice wedges into two main groups along the first axis  
24 based on  $\text{Na}^+$  and  $\text{Cl}^-$  dominating Holocene ice wedges in modern coastal settings and  $\text{Mg}^{2+}$ ,  
25  $\text{Ca}^{2+}$  and  $\text{HCO}_3^-$  for Pleistocene ice wedges and Holocene ones being far from coasts ([Fig. 5](#)).  
26 This pattern depicts the competing influence of maritime and terrestrial/continental  
27 conditions. A similar differentiation of ice wedges (and all ground ice types) is done along the  
28 second PCA axis ([Fig. 5](#)). Differences in stable water isotopes indicate the predominant  
29 climate variations between the late Pleistocene and the Holocene which are also reflected in  
30 the landscape (i.e. distance to sea; maritime vs. continental). Distance from the coastline is  
31 crucial for the incorporation of marine-derived ions through aerosols such as NaCl via sea  
32 spray. While the Fairbanks area is the only site far inland, all other study sites except for  
33 Samoylov Island in the central Lena River Delta are coastal areas today. However, during the



1 late Pleistocene global sea level was lower and large parts of the shallow circum-arctic  
2 shelves were subaerially exposed. Present-day coastal sites were located up to hundreds of  
3 kilometers inland. Marine ion transport via sea spray is not expected to have played a role on  
4 inland sites but indeed since the rapid marine transgression during the Holocene that changed  
5 far inland sites into coastal ones. Input of sea spray is only relevant during the open-water  
6 season so that a prolonged ice cover during the late Pleistocene (Nørgaard-Pedersen et al.,  
7 2003; Bradley and England, 2008) should have further reduced the influx of sea salt.  
8 Additionally, sustained dry conditions (Carter, 1981; Alfimov and Berman, 2001; Murton,  
9 2009) probably increased eolian input of terrestrial material into ice wedges which is then  
10 directly mirrored in the hydrochemical signature.

11  
12 So far we have shown that coastal/maritime and terrestrial environmental conditions can be  
13 differentiated based on inorganic hydrochemistry and that terrestrial surface OC sources feed  
14 the DOC signal in ice wedges. DOC sequestration into ground ice is also dependent on active  
15 layer properties, vegetation cover, vegetation communities, and deposition rates. Long-term  
16 stable surfaces and relatively constant active layer depths will lead to substantially leached  
17 soil layers in terms of DOC (Guo and Macdonald, 2006) and inorganic solutes (Kokelj et al.,  
18 2002). Based on  $\Delta^{14}\text{C}$  values and  $\delta^{13}\text{C}$  ratios on DOC from soil leaching experiments and  
19 natural river water samples, Guo et al. (2007) have shown that intensive leaching of DOC  
20 from young and fresh plant litter and upper soil horizons occurs during the snowmelt period.  
21 Later in the season, DOC yields decreased in rivers draining permafrost areas, indicating that  
22 deepening of the active layer and leaching of deeper seasonally frozen soil horizons were  
23 accompanied by much lower concentrations of DOC due to the refractory and insoluble  
24 character of the remaining organic matter compounds. In addition, dissolved organic matter  
25 compounds in runoff into lakes and rivers can become rapidly degraded by microbial  
26 communities and photochemical reactions (Striegl et al., 2005; Olefeldt and Roulet, 2012;  
27 Cory et al., 2014). One destination of the fresh, young and therefore most bioavailable DOC  
28 components will be ice wedges (Vonk et al., 2013b), where the chemical character is  
29 preserved because of immediate freezing. This highlights the importance of ground ice and  
30 especially ice wedges as a vital source of bioavailable DOC.

31

### 5.3 DOC mobility and quality upon permafrost degradation

The absolute numbers of DOC in permafrost might be still small compared to the POC. However, POC from both peat and mineral soil has a relatively slow decomposition rate after thaw compared to DOC (Schuur et al., 2008). Organic matter from melting ground ice was shown to be highly bioavailable and can even enhance organic matter degradation of the host material by increased enzyme activity in ice wedge meltwater (Vonk et al., 2013b). Bioavailability experiments with Yedoma DOC from thaw streams fed by ice wedge meltwater in NE Siberia illustrated the rapid decomposability of Yedoma OC, with OC losses of up to 33% in 14 days (Vonk et al 2013a). Incubations with increasing amounts of ice wedge water in the Yedoma-water suspension enhanced DOC loss over time. Vonk et al. (2013b) concluded that ice wedges contain a DOM pool of reduced aromaticity and can be therefore regarded as an old but readily available carbon source with a high content of low-molecular weight compounds. Additionally, a co-metabolizing effect through high potential enzyme activity in ice wedges upon thaw leads to enhanced degradation rates of organic matter of the host material. When studying organic matter cycling in permafrost areas we have to abandon the paradigm, which holds true for temperate regions and Arctic oceanography, that old OC is refractory and that only young OC is fresh, bioavailable, and therefore relevant for foods webs and greenhouse gas considerations.

We suggest that reduced organic matter degradation during cold periods is the main reason why late Pleistocene syngenetic ice wedges have incorporated more DOC on average than Holocene ice wedges. Incorporation of soluble organic matter into ground ice might have been more effective than today due to various reasons. Ice Complex deposits in the coastal lowlands formed during the late Pleistocene cold period, when high accumulation rates of fine-grained sediments and organic matter were accompanied by rapidly aggrading permafrost (Hubberten et al., 2004). This means that organic matter is less decomposed because it was rapidly incorporated into perennially frozen ground and into the surrounding syngenetic ice wedges as the permafrost table rose together with the rising surface during deposition (Schirmer et al., 2011b). Also, colder annual air temperatures led to reduced decomposition rates of organic matter which originated from vegetation communities dominated by easily decomposable forbs (Willerslev et al., 2014) in contrast to resistant sedge-moss-shrub tundra vegetation since postglacial times (Andreev et al., 2011). Additionally, low precipitation and reduced runoff presumably retained more DOC in the landscape, ready to be transported into frost cracks.

1  
2 Guo et al. (2007) concluded that most of the DOC in Arctic rivers is derived from young and  
3 fresh plant litter and upper soil horizons. Leaching of deeper seasonally frozen soil horizons  
4 | ~~were~~ is accompanied by much lower DOC concentrations due to the refractory and insoluble  
5 character of the remaining organic matter compounds (Guo et al., 2007). DOC  
6 impoverishment in the active layer is logical as it is leached each season over a long time  
7 under modern climate conditions, where permafrost aggradation is much slower than during  
8 cold stages; if it happens at all. The quantity and quality of DOC pools in deeper permafrost is  
9 probably much higher because of – so far – suppressed remobilization. Dou et al. (2008)  
10 studied the production of DOC as water-extractable organic carbon (WEOC) yields from  
11 organic-rich soil horizons in the active layer and permafrost from a coastal bluff near Barrow  
12 (Alaska) facing the Beaufort Sea. Besides high DOC yields in the uppermost horizon (0-5 cm  
13 below surface) the second highest DOC yields derived from permafrost although the sampled  
14 horizon showed lower soil OC contents than others (Dou et al., 2008). Interestingly, higher  
15 fractions of low-molecular-weight DOC, which is regarded to be more bioavailable, were  
16 generally found at greater depths. This supports the view that permafrost deposits hold a great  
17 potential for mobilizing large quantities of highly bioavailable organic matter upon  
18 degradation. Coastal erosion and thermokarst often expose old and deep permafrost strata.  
19 Contained organic matter is directly exposed to the atmosphere and transferred into coastal  
20 and fresh-water ecosystems without degradation because of short travel and residence times.  
21 Therefore, Arctic coastal zones are supposed to receive high loads of bioavailable dissolved  
22 and particulate organic matter. Dou et al. (2008) used pure water (presumably MilliQ) and  
23 natural sea water as solvent for studying the production of DOC. It turned out that seawater  
24 extraction significantly reduced DOC yields which were attributed mainly to reduced  
25 solubility of humic substances due to the presence of polyvalent cations such as  $\text{Ca}^{2+}$  and  
26  $\text{Mg}^{2+}$  in seawater (Aiken and Malcolm, 1987). On the one hand Dou et al. (2008) invoked that  
27 a laboratory setup using pure water and dried/rewetted soil samples would lead to an  
28 overestimation of DOC input to the Arctic Ocean during coastal erosion. On the other hand  
29 and based on the large ground-ice volumes in coastal cliffs (Lantuit et al., 2012), we suggest  
30 that ice wedge meltwater with a low ion content is probably able to leach greater amounts of  
31 DOC from permafrost upon thaw than other natural surface water.

32

1 An open question remains how much DOC can be found in intrasedimental ice and how much  
2 DOC is produced upon degradation of old permafrost (e.g. late Pleistocene Yedoma type) for  
3 example as a result of coastal erosion. The answer to this question is crucial to follow the fate  
4 of permafrost organic matter upon re-mobilization. Additionally, robust estimations of carbon  
5 release are crucial for predicting the strength and timing of carbon-cycle feedback effects, and  
6 thus how important permafrost thaw will be for climate change this century and beyond.  
7

## 6 Conclusions and outlook

Ground ice in ice-rich permafrost deposits contains DOC, DIC and other nutrients, which are relevant to the global carbon cycle, arctic fresh-water habitats and marine food webs upon release.

The following conclusions can be drawn from this study:

- Ice wedges represent a significant DOC (45.2 Tg) and DIC (33.6 Tg) pool in the studied permafrost areas and a considerable fresh-water reservoir of 4,200 km<sup>3</sup>.
- Syngenetic late Pleistocene ice wedges have the greatest potential to host a large pool of presumably bioavailable DOC because of i) highest measured average DOC concentrations in combination with ii) their wide spatial (lateral, vertical) distribution in ice-rich permafrost areas and iii) the sequestration of fresh and easily leachable OC compounds.
- Increased incorporation of DOC into ground ice is linked to relatively high proportions of terrestrial cations, especially Mg<sup>2+</sup> and K<sup>+</sup>. This indicates that leaching of terrestrial organic matter is the most relevant process of DOC sequestration into ground ice.

Based on our results about the stocks and chemical behavior of DOC in massive ground-ice bodies we propose that further studies shall strive to:

- quantify DOC fluxes in the Arctic from thawing permafrost, melting ground ice and coastal erosion,
- differentiate between DOC and POC in permafrost including non-massive intrasedimental ice,
- quantify DOC production from permafrost in different stratigraphic settings and with different natural solvents to answer the question, what fraction of soil OC will be leached as DOC,
- assess the age and lability of DOC versus POC in permafrost and the potential impact on coastal food webs and fresh-water ecosystems.

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6 **Tables**

7

8 **Table 1.** Summary of study areas, study sites, stratigraphy of the host sediments, ground-ice inventory and the studied ice types.

Region	Location	Longitude	Latitude	Stratigraphy and host sediments	Ground-ice conditions (inventory, ground-ice types, sampled ice types marked in <i>italic</i> )	Reference
Western Laptev Sea	Cape Mamontov Klyk	117.2	73.6	<ul style="list-style-type: none"> <li>Fluvial bottom sands – Late Weichselian Ice Complex – <u>L</u>ate glacial to Holocene thermokarst deposits – Holocene valley deposits – Holocene cover deposits</li> <li>Yedoma hills (20-40 m a.s.l.) of ice-rich permafrost sequences with wide and deep syngenetic ice wedges separated by thermoerosional valleys and thermokarst depressions</li> </ul>	Ice-rich permafrost sequences with wide and deep syngenetic <i>late Pleistocene ice wedges</i>	Schirrmeister et al., 2008; Schirrmeister et al., 2011b; Boereboom et al., 2013
Lena Delta	Samoylov Island	126.4	72.4	<ul style="list-style-type: none"> <li>First terrace (0-10 m a.s.l.): early to late Holocene delta floodplain, along the main river channels in the central and eastern parts of the delta; fluvial facies from organic-rich sands to silty-sandy peats towards bottom-up</li> <li>Modern to late Holocene floodplain; alluvial facies from peaty sands to silty-sandy peats bottom-up</li> </ul>	Ice-rich permafrost with active and buried syngenetic <i>Holocene ice wedges</i>  Ice-rich permafrost with epigenetic <i>Holocene ice wedges</i>	Schwaborn et al., 2002; Schirrmeister et al., 2011a; Meyer et al., 2015
Eastern Laptev Sea	Muostakh Island	129.9	71.6	<ul style="list-style-type: none"> <li>Late-glacial and Holocene cover deposits on top of Ice Complex</li> <li>Middle to Late Weichselian Ice Complex</li> </ul>	Very ice-rich permafrost, <i>late Pleistocene ice wedges, Holocene ice wedges</i>	Schirrmeister et al., 2011b, c (and references therein), Günther et al., 2015
Dmitry Laptev	Oyogos Yar coast	143.5	72.7	<ul style="list-style-type: none"> <li>Alternation of wide thermokarst depressions (alases) and hills representing remnants of Ice-Complex deposits (Yedoma)</li> </ul>	Late Pleistocene and	Wetterich et al., 2009;



Strait				<ul style="list-style-type: none"> <li>Late-glacial to Holocene thermokarst deposits and on top of Ice Complex</li> <li>Taberite formed during Weichselian to Holocene transition</li> <li>Late Weichselian Ice Complex</li> <li>Middle Weichselian Ice Complex</li> </ul>	<p><i>Holocene ice wedges,</i></p> <p>All ice wedges were sampled at a coastal bluff at an elevation of about 10 m a.s.l. in a central alas depression</p>	<p>Opel et al., 2011; Schirrmeister et al., 2011b</p>
New Siberian Islands	Bol'shoy Lyakhovsky Island	143.9	73.2	<ul style="list-style-type: none"> <li>Late Holocene cover deposits and Holocene valley deposits</li> <li>Late-glacial to Holocene thermokarst deposits</li> <li>Taberite formed during Weichselian to Holocene transition</li> <li>Middle Weichselian Ice Complex</li> </ul>	<p><i>Late Pleistocene ice wedges</i></p>	<p>Meyer et al., 2002 ; Andreev et al., 2004, 2009; Schirrmeister et al., 2011b; Wetterich et al., 2011, 2014</p>
Northern Alaska	Barrow (CRREL Permafrost Tunnel)	-156.7	71.3	Buried ice-wedge system under about three meters of <u>L</u> ate-glacial to early Holocene ice-rich sediments	<p><i>Late-glacial ice wedges,</i> Holocene ice wedges</p>	<p>Sellman and Brown, 1973; Meyer et al., 2010a, b</p>
Interior Alaska	Fairbanks (Vault Creek Tunnel)	-147.7	65.0	Discontinuous permafrost. Late Pleistocene ice-rich silty, loess-like organic-rich sediments between 12-15 m thick with large intersecting ice wedges	<p><i>Late Pleistocene ice wedges,</i> Holocene ice wedges</p>	<p>Shur, et al. 2004; Meyer et al., 2008</p>
Yukon Coast	Komakuk Beach	-140.5	69.6	<ul style="list-style-type: none"> <li>Middle and late Holocene ice-rich peat, polygonal tundra</li> <li>Early Holocene thaw-lake sediments, peat, ice wedge casts</li> <li>Late Wisconsin (i.e. Late Weichselian) proluvial, alluvial, eolian deposits</li> </ul>	<p>Holocene ice wedges, <i>Holocene snow pack ice (fossil snow bank)</i></p>	<p>Rampton, 1982; Fritz et al., 2012</p>
Yukon Coast	Herschel Island	-139.1	69.6	<ul style="list-style-type: none"> <li>Retrogressive thaw slumps along the coast exposing massive ground ice and ice-rich sediments</li> </ul>	<p><i>Buried glacier ice of <math>\geq</math> 20 m thickness within</i></p>	<p>Mackay, 1959; Rampton,</p>

				<ul style="list-style-type: none"> <li>• Holocene cover deposits and slope material along steep coastal bluffs</li> <li>• Mixed origin of marine, near-shore and terrestrial deposits</li> <li>• Push end-moraine of Late Wisconsin age</li> </ul>	<p>Late Wisconsin diamicton</p> <p><i>Late Wisconsin ice wedges</i> truncated by mass movement and early Holocene thaw unconformity</p> <p>Epigenetic and anti-syngenetic <i>Holocene ice wedges</i></p> <p><i>Buried lake ice</i>, fossil snow bank ice</p>	<p><a href="#">1982; Fritz et al., 2011, 2012</a></p>
Yukon Coast	Roland Bay	-139.0	69.4	<ul style="list-style-type: none"> <li>• Retrogressive thaw slumps along the coast exposing massive ground ice and ice-rich sediments</li> <li>• Holocene cover deposits and slope material along steep coastal bluffs</li> <li>• Late Wisconsin diamicton</li> </ul>	<p><i>Late Wisconsin and Holocene ice wedges</i></p>	<p><a href="#">Rampton, 1982</a></p>
Yukon Coast	Kay Point	-138.2	69.2	<ul style="list-style-type: none"> <li>• Retrogressive thaw slumps along the coast exposing massive ground ice and ice-rich sediments</li> <li>• Holocene cover deposits and slope material along steep coastal bluffs</li> <li>• Moraine (ridge) of Late Wisconsin age</li> </ul>	<p>Presumably <i>Late Wisconsin buried glacier ice, Holocene ice wedges</i></p>	<p><a href="#">Rampton, 1982; Harry et al., 1985</a></p>

1 [Table 2](#). Summarized DOC and DIC concentrations of different massive ground-ice types. For  
 2 individual sample values see [Supplement Table S1](#).

Ice type	DOC			DIC			No. of ice bodies	No. of samples	Stratigraphic affiliation
	DOC Mean [mg L <sup>-1</sup> ]	DOC concentration range [mg L <sup>-1</sup> ]	No. of ice bodies	No. of samples	DIC Mean [mg L <sup>-1</sup> ]	DIC concentration range [mg L <sup>-1</sup> ]			
Ice wedge ice	9.6	1.6–28.6	22	72	4.7	0.3–19.8	21	66	Holocene, Late Pleistocene
Basal glacier ice	1.8	0.7–3.8	5	22	9.3	0.1–25.4	4	19	Late Pleistocene
Buried lake ice	2.0	0.3–5.2	1	6	8.8	0.3–22.9	1	6	Late Pleistocene
Snow pack ice	3.0	n.a.	1	1	n.a.	n.a.	0	0	Holocene
<i>Modern surface water</i>	5.6	5.5–5.7	3	3	22.6	5.0–40.2	3	3	<i>recent</i>

3 Three modern surface water samples are from three different water bodies representing thermokarst  
 4 ponds along the Yukon Coast.

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1 **Table 3.** DOC stocks and pools in late Pleistocene and Holocene permafrost containing ice  
 2 wedges (IW) based on calculated wedge-ice volumes (WIV) in Yedoma and thermokarst  
 3 basin deposits. All other ground-ice types, especially non-massive intrasedimental ice, are not  
 4 included.

	DOC concentration in Pleistocene IW mg L <sup>-1</sup>	DOC concentration in Holocene IW mg L <sup>-1</sup>	WIV in Pleistocene Yedoma deposits vol%	WIV in Holocene thermokarst deposits vol%	DOC stocks in Pleistocene permafrost <sup>c</sup> g m <sup>-3</sup>	DOC stocks in Holocene permafrost <sup>c</sup> g m <sup>-3</sup>	DOC pools in Pleistocene permafrost <sup>c,d</sup> Tg	DOC pools in Holocene permafrost <sup>c,d</sup> Tg
Min	2.4	1.6	16.7 <sup>a</sup>	1.0 <sup>a</sup>	0.4	0.02	3.2	0.07
Mean	11.1	7.3	48.0 <sup>b</sup>	7.0 <sup>b</sup>	5.3	0.51	43.0	2.2
Max	28.6	19.5	63.2 <sup>a</sup>	13.2 <sup>a</sup>	18.1	2.6	145.9	11.0

5 <sup>a</sup> WIV data by [Ulrich et al., 2014](#). <sup>b</sup> Mean WIV data by [Strauss et al., 2013](#). <sup>c</sup> This includes ice wedges  
 6 only. <sup>d</sup> According to [Strauss et al. \(2013\)](#) undisturbed Pleistocene Yedoma covers 416,000 km<sup>2</sup> with a  
 7 mean thickness of 19.4 m, whereas Holocene thermokarst deposits cover 775,000 km<sup>2</sup> with a mean  
 8 thickness of 5.5 m.

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1 **Figures**

2

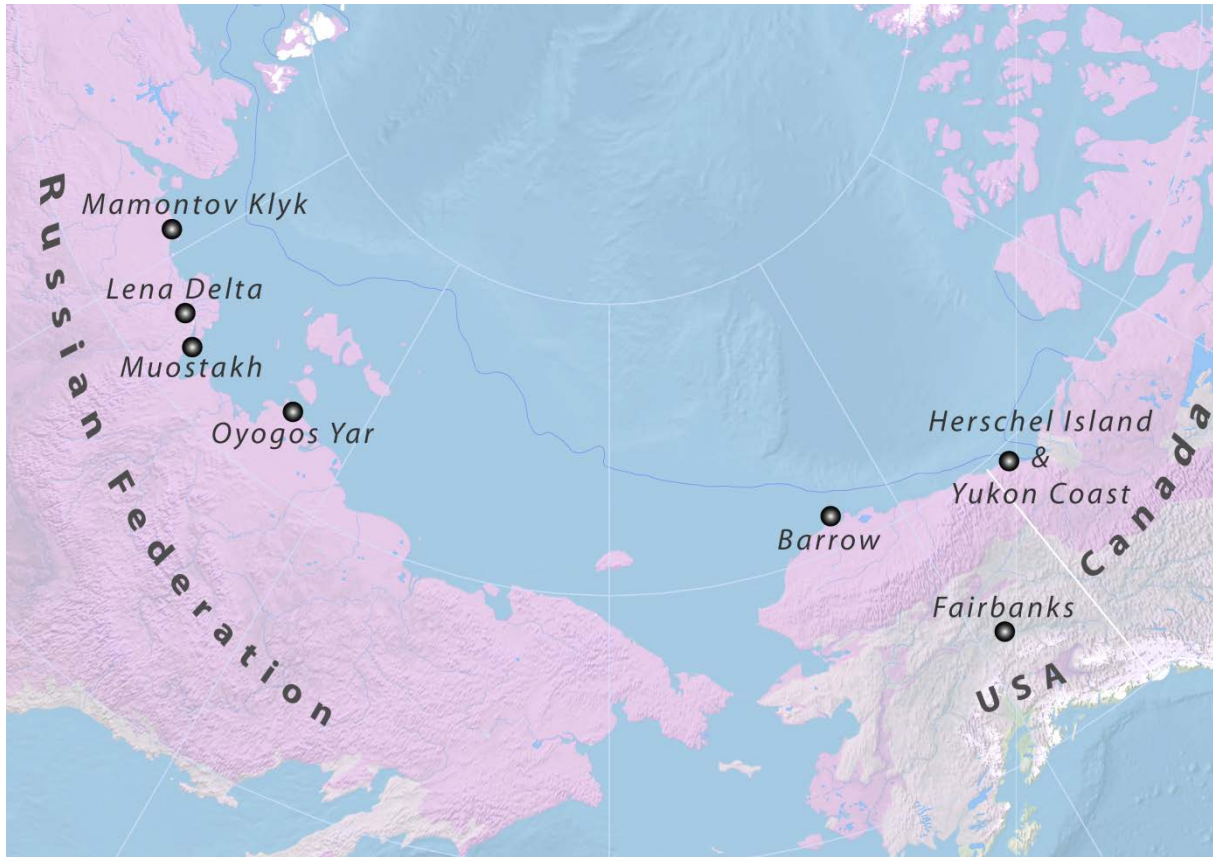


Figure 1. Study area and study sites (dots) for massive ground ice sampling in the Arctic lowlands of Siberia and North America. All study sites are located within the zone of continuous permafrost (dark purple), except for the Fairbanks area, which is the zone of discontinuous permafrost (light purple). Blue line in the Arctic Ocean marks the northerly extent of submarine permafrost according to [Brown et al. \(1997\)](#).

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4



Ice-rich late Pleistocene permafrost, Oyogos Yar, Russia



Ice-rich permafrost coast, Muostakh, Russia



Epigenetic Holocene ice wedge, Lena Delta (Samoylov Island), Russia



Syngenetic late Pleistocene ice wedge, Mamontov Klyk, Russia



Buried glacier ice, Herschel Island, Canada



Snow pack ice, Herschel Island, Canada



Buried lake ice, Herschel Island, Canada

Figure 2. Ground-ice conditions and examples of studied ground-ice types in the Siberian and North American Arctic. Place names are plotted on Fig. 1.

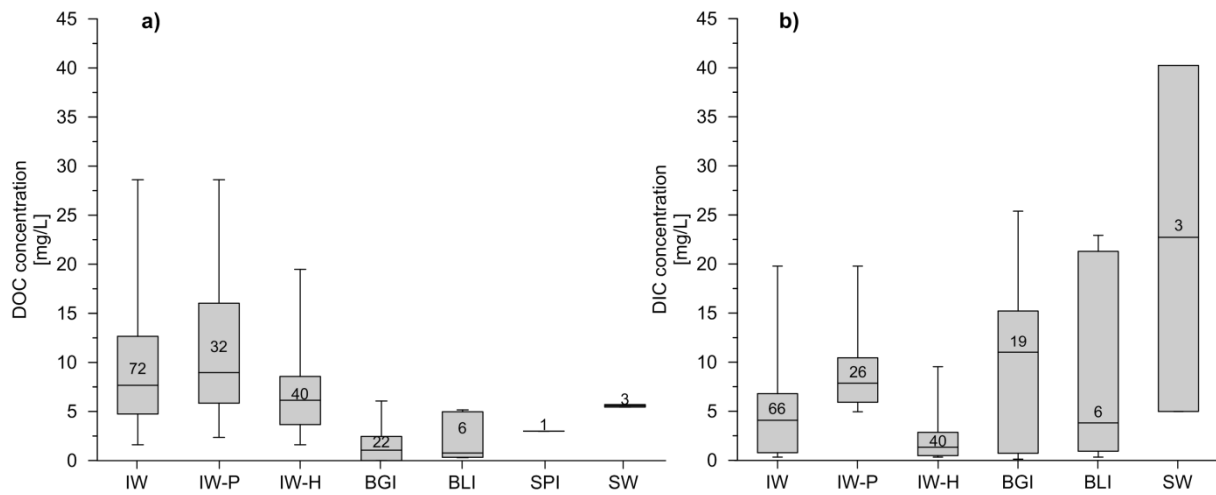


Figure 3. Boxplots of (a) DOC and (b) DIC concentrations in different massive ground-ice types. Plots show the number of samples in each category, minimum, maximum, median, 25 per cent-quartile and 75 per cent quartile as edge of boxes. IW: Ice wedges (all), IW-P: Pleistocene ice wedges, IW-H: Holocene ice wedges, BGI: Buried glacier ice, BLI: Buried lake ice, SPI: Snow pack ice, SW: Surface water. For individual sample values see [Supplement Table S1](#).

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DOC – unsorted correlation matrix

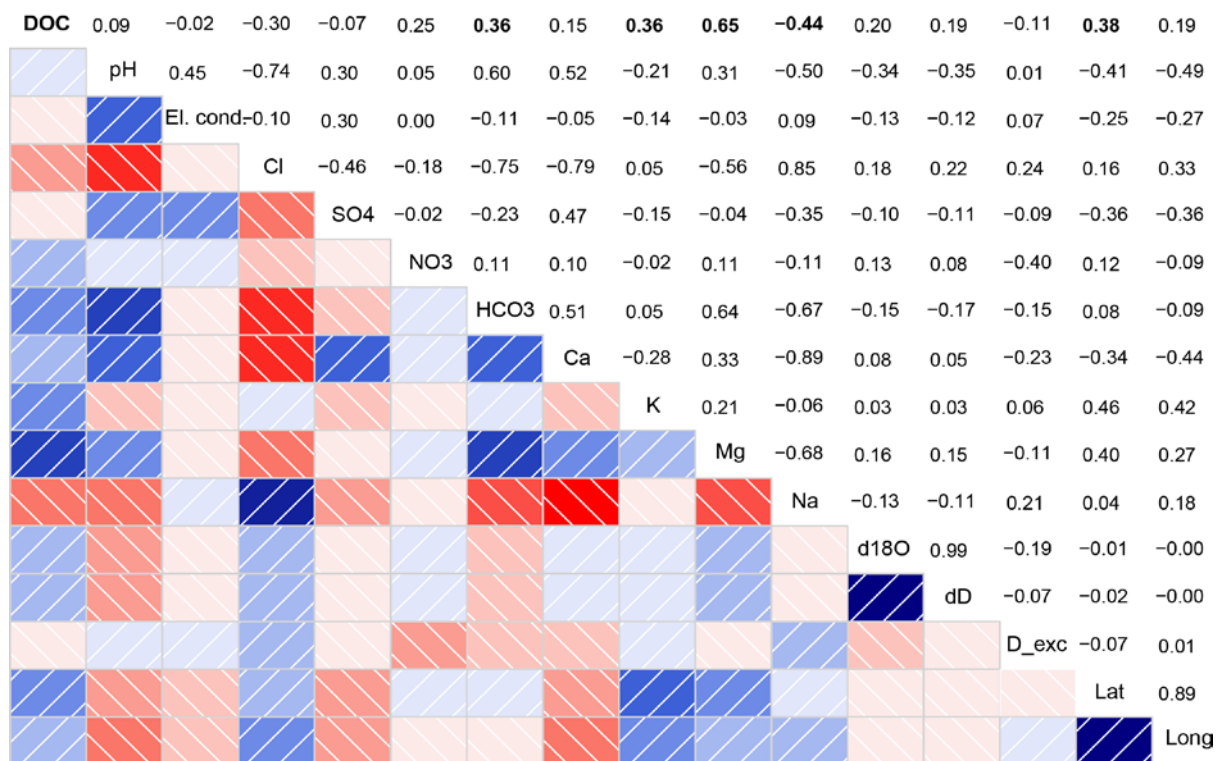


Figure 4. Correlation matrix. Correlations mentioned in the text are printed in bold. Strong positive correlations of paired variables are indicated by dark bluish colors, while strong anti-correlations are depicted in red. Hatching from the upper right to the lower left depict positive correlations, whereas negative correlations are reversely hatched for better perceptibility in a black-and-white print. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

1

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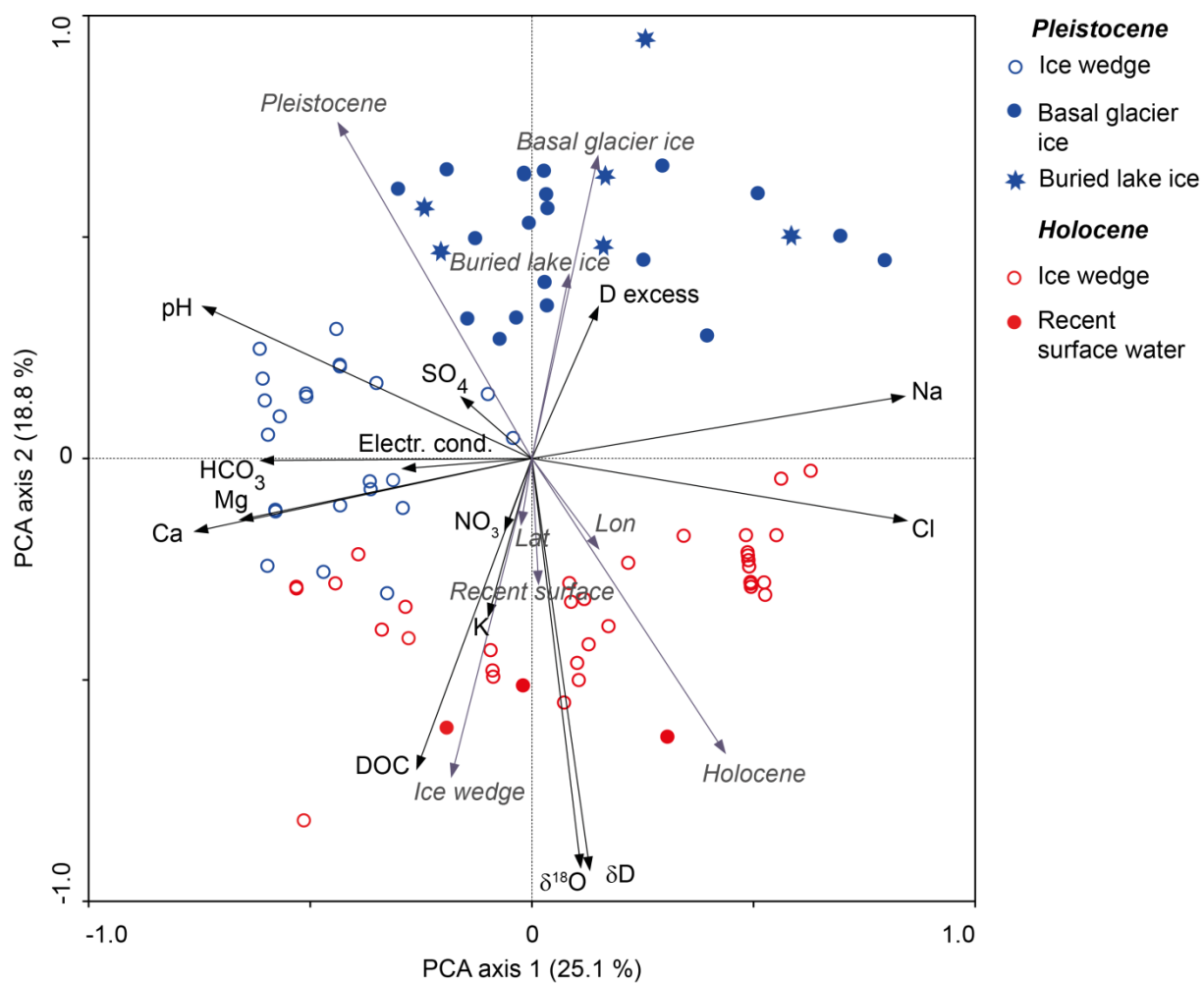


Figure 5. PCA biplot for ground-ice data. Inactive supplementary parameters (i.e. ice wedge, buried lake ice, basal glacier ice, snow pack ice, surface water, Pleistocene, Holocene, recent) are shown in grey italic. For individual sample values see [Supplement Table S1](#).

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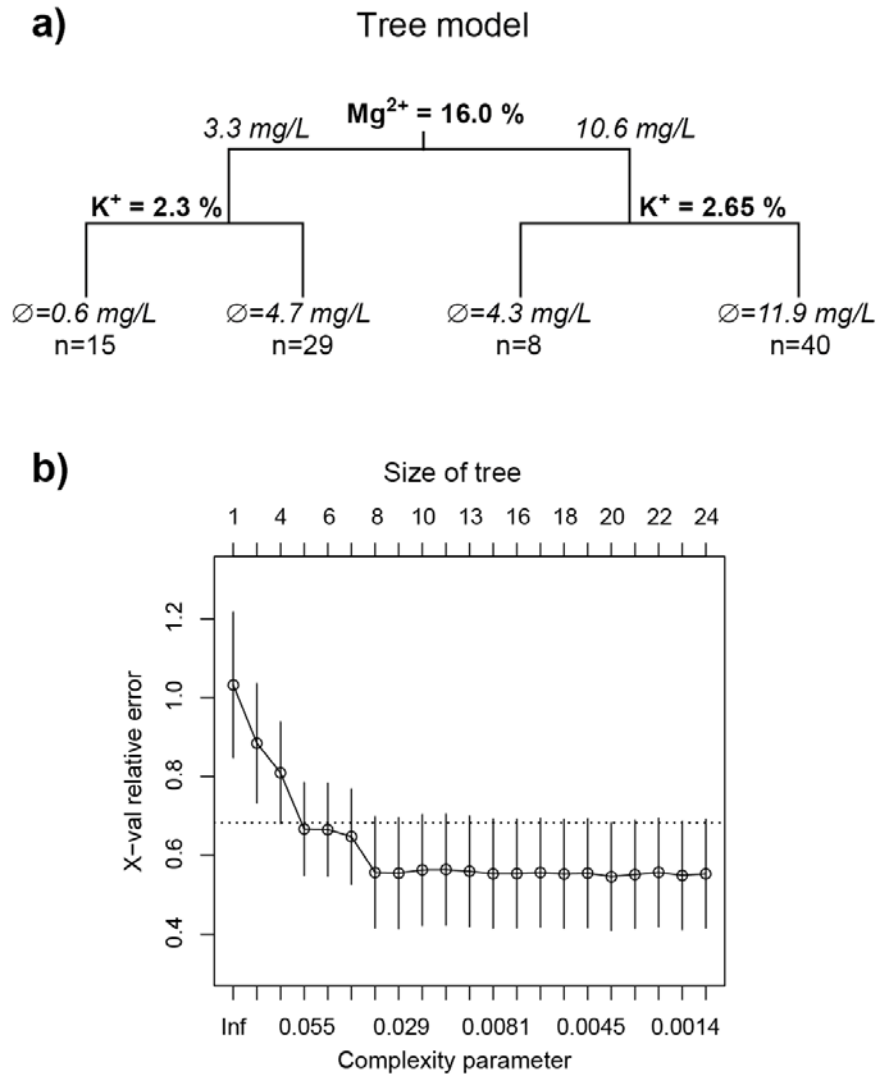


Figure 6. Univariate Tree Model (UTM) explains variability pattern in DOC concentration. a) Tree model focuses on DOC concentration as response variable. UTM uses 92 observations and a set of 22 explanatory variables.  $Mg^{2+}$  and  $K^+$  ions are most important to explain differences in DOC concentrations. Mean DOC concentrations of each group in  $mg L^{-1}$ . Number of observations in each group (n). b) Cross validation determines the statistically significant size of the tree model. The dotted line is obtained by the mean value of the errors (x-error) of the cross validations plus the standard deviation of the cross validations upon convergence. For individual sample values see [Supplement Table S1](#).