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*Supplement of*

## **Can we retrieve a clear paleoclimatic signal from the deeper part of the EPICA Dome C ice core?**

**J.-L. Tison et al.**

*Correspondence to:* J.-L. Tison (jtison@ulb.ac.be)

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## 1 **Supplementary Material**

2

### 3 S1. Material and Methods

4

#### 5 *S1.1. Chemical analyses*

6 At the LGGE (Laboratoire de Glaciologie et Géophysique de l'Environnement),  
7 preliminary discontinuous samples were rinsed in 3 successive baths of ultrapure water.  
8 This decontamination procedure has been proved to be the most efficient even for  
9 organic traces in Antarctic ice (de Angelis et al., 2012) but is not suitable for high  
10 resolution sampling. Ice lamellae devoted to high resolution studies (both at LGGE and  
11 British Antarctic Survey - BAS) were thus decontaminated with a plane and cut into  
12 successive samples 1.5 to 2.5cm long, a dry cleaning procedure previously checked  
13 and intensively used for analyzing mineral ions in Antarctic firn and ice cores (Littot et  
14 al., 2002) . Concentrations were determined by ionic chromatography (IC) with a typical  
15 analytical uncertainty varying from 1 to 5%.

#### 16 *S1.2 Water Isotopes*

17 Measurements of the EDC  $\delta D$  and  $\delta^{18}O$  were conducted on continuous “bag samples”  
18 with a depth resolution of 0.55 m, using classical isotope ratio mass spectrometry  
19 (IRMS) techniques. The  $\delta D$  measurements were performed in France at Laboratoire  
20 des Sciences du Climat et l'Environnement (LSCE) using an automatic injection device  
21 and the uranium reduction technique, with a precision of 0.5 per mil. The  $\delta^{18}O$   
22 measurements of the whole core up to 3189.45 m were performed in Italy at the  
23 Department of Earth Sciences of the University of Parma (DST) and at the Department  
24 of Mathematics and Geosciences of the University of Trieste (DMG) using the  $CO_2$ -  
25 water equilibration technique and with a precision of 0.05 per mil. The  $\delta^{18}O$   
26 measurements of the bottom part have been conducted in Denmark at the Centre for Ice  
27 and Climate in Copenhagen using the  $CO_2$ -water equilibration technique. The data are  
28 reported against VSMOW values. Inter-comparison of reference waters among the  
29 involved laboratories were conducted over the analysis period and the same water standard  
30 was used in both LSCE and DST/DMG laboratories.

#### 31 *S1.3 Total gas content, Gas Mixing ratios and isotopes*

32 The measurements of the total gas content have been performed at LGGE using an  
33 original barometrical method implemented with an experimental setup called STAN  
34 (Lipenkov et al., 1995). This technique allows a precise evaluation of the pressure and  
35 temperature of the air extracted from an ice sample having a mass of 20-30 g by its  
36 melting-refreezing under vacuum in a volume-calibrated cell. After correction of the  
37 measured pressure for the partial pressure of saturated water vapour and of the  
38 calibrated volume for the volume occupied by refrozen bubble free ice, the gas content  
39  $V$  is calculated using the ideal gas law. The  $V$  values are then corrected for gas loss

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40 from air inclusions (i.e. gas hydrates and relaxation features such as gas cavities at  
41 depths under consideration) cut at the surface of the sample (Martinerie et al., 1990).  
42 The absolute precision of the STAN measurements has been estimated to be within  
43  $\pm 0.6\%$ . However the overall error of obtained  $V$  values amounts 1% because of the  
44 uncertainties in the cut-bubble correction. The replication of the results estimated by  
45 repeated measurements in the same horizontal slice of an ice core has been confirmed  
46 to be better than 1%.

47 For  $\text{CO}_2$  measurements at the University of Bern, ice samples of about 7 g were  
48 pulverized in a cooled and evacuated vacuum chamber using a needle cracker  
49 principle. After this dry extraction process, the extracted air was expanded from the  
50 vacuum chamber into an infrared laser spectrometer which is used to derive the  $\text{CO}_2$   
51 concentration. At the LGGE about 40g of ice were pulverized in a cooled vacuum  
52 chamber using a ball mill principle. In this system the extracted air is transferred into a  
53 GC in order to derive the  $\text{CO}_2$  concentration. A more detailed description of the systems  
54 used in Bern and at the LGGE is found in Siegenthaler et al. (2005) and Barnola et al.  
55 (1987), respectively. The  $\text{CO}_2$  data set shown in this paper has a resolution of about 2.2  
56 m and a precision of 1.6 - 2 ppmv (corresponding to mean Bern and LGGE precision,  
57 respectively), whereas up to 8 replicate measurements were performed.

58 For  $\text{CH}_4$  and  $\text{N}_2\text{O}$  measurements, ice samples of about 40 g (University of Bern) or 50 g  
59 (LGGE) were melted in sealed and evacuated glass containers. After refreezing of the  
60 samples the extracted ancient air is injected into a sampling loop and analyzed by gas  
61 chromatography. The gas chromatographs are equipped with a thermal conductivity  
62 detector (TCD, for total air amount), a flame ionization detector (FID, for  $\text{CH}_4$ ), and an  
63 electron capture detector (ECD) for  $\text{N}_2\text{O}$  (University of Bern only). A detailed description  
64 of this melt-refreezing method and the measurement systems can be found in Flückiger  
65 et al. (2004) and Chappellaz et al. (1997). The precision of the  $\text{CH}_4$  and  $\text{N}_2\text{O}$   
66 measurements is 10 and 5.6 ppbV respectively. Note that the  $\text{N}_2\text{O}$  measurements are  
67 potentially affected by in-situ production in the ice leading to elevated values (see Schilt  
68 et al., 2010). The mean resolution of the  $\text{CH}_4$  and  $\text{N}_2\text{O}$  measurements for the depth  
69 interval 3191 to 3259 m is 1.4 and 2.1 m, respectively.

70 The composition of the oxygen 18 in entrapped air, which gives access to the isotopic  
71 composition of the atmosphere,  $\delta^{18}\text{O}_{\text{atm}}$ , has been measured at LSCE using the melt-  
72 freezing method for air extraction followed by mass-spectrometry measurement as  
73 described in Landais et al. (2003) and Dreyfus (2008). The measurement precision for  
74 the 2004-2005 dataset is 0.02 ‰ (pooled standard deviation).

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### 75 S1.4 Crystal size

76 Crystal sizes in the Deep and Basal ice were determined in the field, using a simple  
77 version of the linear intercept method (Pickering, 1976). In this method, the number (N)  
78 of grain boundaries crossed by a random linear traverse of length (L) across the thin  
79 section is averaged over many traverses. The mean grain diameter (d) is then  
80 estimated as  $d = \bar{N}/\bar{L}$ . At NEEM, crystals with cross sections up to 40 cm were  
81 observed, with occasional bands of smaller crystals. We therefore simply counted the  
82 amount of crystals along a center line of a 3mm thick slab observed through polarized  
83 light, and divided by the length of the core (55cm).

### 84 S1.5 Dust

85 At the LGGE), a set of 24 discontinuous 7-cm long samples was selected. Ice was  
86 decontaminated through 3 successive baths of ultrapure water. Insoluble dust  
87 concentration and size distribution measurements were performed by Coulter Counter  
88 Multisizer IIe in a clean room setting. Each data represents the average of three  
89 consecutive measurements performed following ultrasonic treatment. A density of 2,5  
90 g/cm<sup>3</sup> was assumed for all particles in agreement with earlier studies. Analytical  
91 procedure followed in this study is identical to that described in Lambert et al. (2008).

### 92 S2. Validity of the comparison between previous “full glacial” periods and the Deep and 93 Basal ice layers

94 As discussed in section 3, we have chosen to compare the chemistry of the Deep and  
95 Basal ice layers to the one of the previous glacial climatic episodes, showing minimum  
96  $\delta D$  values. To achieve this, we have selected these successive “full glacial” episodes by  
97 isolating in the  $\delta D$  data set the periods with minimal values, using locations of  
98 increasing  $\delta$ -gradients as cutting points on both sides. Table 1 shows that the  $\delta D$  ranges  
99 obtained for the various selected periods are quite similar. However, these various  
100 glacial episodes obviously cover different depth and time intervals. Before we discuss  
101 and compare the mean values and the variability between those groups and the Deep  
102 and Basal ice layers, it is essential to show that no artificial “time smoothing” is induced  
103 by the methodology. For example, one might expect that, depending on the resolution of  
104 the data set, increasing the time window at a given resolution would damp the  
105 variability. In figure S1, we have plotted, for each chemical species considered and for  
106 each group of samples, their mean (Fig. S1a) and  $1\sigma$  (Fig. S1b) value as a function of  
107 the duration of the period, based on the EDC-3 time scale (Parrenin, 2007, extended for  
108 the Deep and Basal ice layer). Although there is some variability, no clear trend of the  
109 chemical signals is seen with duration of the episodes. Furthermore, if “time smoothing”  
110 was to occur, it should result in a decrease of the variability, at a given resolution, as we

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111 go down the EDC core, which is the reverse of what is observed comparing the Deep  
112 and Basal ice groups to the “full glacial” episodes above. Finally, the contrast between  
113 the signature of different elements within the Deep ice or within the Basal ice concerns  
114 samples that are strictly from the same time window, at the same resolution, so that it is  
115 sound to discuss it.