- Halogen-based reconstruction of Russian Arctic sea ice area from the Akademii Nauk ice core (Severnaya Zemlya)
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33 ABSTRACT

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35 The role of sea ice in the Earth climate system is still under debate, although it is known to influence 36 albedo, ocean circulation, and atmosphere-ocean heat and gas exchange. Here we present a 37 reconstruction of 1950 to 1998 A.D. sea ice in the Laptev Sea based on the Akademii Nauk ice core 38 (Severnaya Zemlya, Russian Arctic). The chemistry of halogens bromine (Br) and iodine (I) are 39 strongly active and influenced by sea ice dynamics, in terms of physical, chemical and biological 40 process. Bromine reacts on the sea ice surface in auto-catalyzing "Bromine explosion" events causing 41 an enrichment of the Br/Na ratio and hence a bromine excess (Brexc) in snow compared to that in 42 seawater. Iodine is suggested to be emitted from algal communities growing under sea ice. The results 43 suggest a connection between Brexc and spring sea ice area, as well as a connection between iodine 44 concentration and summer sea ice area. The correlation coefficients obtained between Brexc and spring 45 sea ice (r=0.44) as well as between iodine and summer sea ice (r=0.50) for the Laptev sea suggest that 46 these two halogens could become good candidates for extended reconstructions of past sea ice changes 47 in the Arctic.

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49 **1 INTRODUCTION**

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51 The rapid and unexpected decrease of Arctic sea ice during recent decades has highlighted the lack of 52 knowledge regarding the mechanisms controlling sea ice growth and decay. Sea ice affects albedo by 53 covering the relatively dark, energy-absorbing ocean waters with highly reflective ice (Francis et al., 54 2009). Sea ice formation is an important process for driving salinification of surface waters, thereby 55 promoting convection in polar regions (Holland et al., 2001). Sea ice also is an efficient barrier 56 between the ocean and atmosphere, limiting the effectiveness of ocean water to warm the polar 57 atmospheric boundary layer, as well as limiting both the exhalation of CO₂ from CO₂-rich upwelling 58 circum-Antarctic waters and the drawdown of atmospheric CO₂ by downwelling surface waters 59 (Dieckmann and Hellmer, 2010).

Satellite measurements have documented a rapid decrease in recent Arctic summer sea ice area (Comiso, 2011;Stroeve et al., 2007). During the last 10 years, the four lowest September sea ice minima of the last 35 years have been recorded (Arctic Sea-Ice Monitor – IJIS, www.ijis.iarc.uaf.edu) with the lowest sea ice area of 3.41 million km² recorded on 16th September 2012. This is 47% of the average sea ice minimum extent for the 1970-1990 period (7.2 million km²), but the latest oceanatmosphere coupled climate models are unable to replicate the rapid pace of Arctic sea ice retreat(Turner et al., 2012). Such limitations may result from poor parameterization of key physical sea ice processes because the only record available is the relatively short 35-year period of satellite observations.

69 Accurate reconstruction of sea ice variability before the satellite epoch is important for understanding 70 interactions between sea ice area and both the forcing and effects of climate changes (Wolff et al., 71 2006). These results also are important for improved model calibration. Many different approaches 72 have been proposed to reconstruct sea ice extent variability. Sediments in marine cores reflect sources 73 and so are used to show pathways of Arctic and sub-Arctic oceanic circulation (Darby, 2003). Ice-74 rafted debris in marine sediment records indicates the occurrence of floating ice (Lisitzin, 2002), while 75 sea ice-related palaeo-productivity can be inferred from the accumulated remains of microscopic 76 organisms and other biomarkers. Recently, highly branched isoprenoids (IP₂₅) in specific sea ice 77 diatoms in sediment cores have also been proposed and applied as indicators of past sea ice variability 78 (Belt et al., 2007; Müller et al., 2009; Xiao et al., 2013). Coastal records also help to understand the past 79 dynamics of sea ice, producing a clear signal in both coastal sediments and landforms (Polyak et al., 80 2010). Additionally, integration of various paleoclimate archives and historical observations allows a 81 broad reconstruction of past sea ice variability in the Arctic (Divine and Dick, 2006;Kinnard et al., 82 2011;Polyakov et al., 2003;Vinje, 2001). A weakness of the available reconstructions of past sea ice 83 area and variability, extending back more than a century, is that they are characterized by poor 84 temporal resolution and/or a limited regional significance.

85 Many atmospheric conditions are recorded in high temporal resolution in ice cores, such as dust 86 deposition, temperature, solar radiation and atmospheric gas concentration (Wolff et al., 2010) and so 87 they are employed extensively for reconstructing past climate (Petit et al., 1999). The absence of 88 sufficiently reliable and specific proxies has limited their application to reconstructions of sea ice 89 extent and variability. Methanesulphonic acid (MSA), a product of ocean algal emissions, has been 90 used to reconstruct past sea ice changes from both Antarctic and Arctic ice cores (Curran et al., 91 2003; Isaksson et al., 2005), however MSA is unstable and remobilized in ice cores over centennial to 92 millennial time scales (Röthlisberger et al., 2010). Sodium in ice cores reflects glacial-interglacial sea 93 ice variability but on shorter timescales is strongly influenced by meteorology as well as competing sea 94 ice and open ocean emission sources (Abram et al., 2013;Levine et al., 2014).

Recent studies of the halogen elements Bromine (Br) and Iodine (I) have shown potential for their use
as proxies of polar sea ice area extent in both Antarctic and Arctic regions (Spolaor et al., 2014;Sturges

97 and Barrie, 1988). Bromine is released into the atmosphere as a component of sea salt. An additional 98 source comes from so-called "bromine explosions" that are defined as an autocatalytic sequence of 99 chemical reactions able to generate gaseous bromine compounds such as BrO from bromide trapped in 100 the sea ice (Pratt et al., 2013; Vogt et al., 1996). This is supported by satellite measurements in polar 101 regions that show pronounced springtime increases in atmospheric BrO concentrations (Schonhardt et 102 al., 2012) associated with sea ice presence. These so-called "bromine explosions" (Simpson et al., 103 2015; Simpson et al., 2007), have the net effect of enriching Br beyond the seawater Br/Na ratio in the 104 snow deposits in the ice caps and subsequently ice cores (Spolaor et al., 2014; Sturges and Harrison, 105 1986; Sturges and Barrie, 1988). Some studies have found depleted Br compared to sea water 106 abundances in coastal areas (de Gois et al., 2015; Barrie et al., 2001). However all measurements 107 conducted on Arctic and Antarctic ice caps demonstrate a general enrichment of Br (Spolaor et al., 108 2014) compared to sea water abundances. Other processes can influence the Br excess detected in the 109 snow, however previous findings suggest that Bromine related sea ice chemistry could be the major 110 explanation.

111 Iodine emissions to the polar atmosphere mainly arise from oceanic biological production and 112 formation of volatile organo-iodine compounds (Atkinson et al., 2012). Laboratory analyses suggest 113 that sea salt iodine contributes less than 2% of total iodine deposition (Spolaor et al., 2013b;Sturges and 114 Barrie, 1988). Although it has been discovered that atmospheric iodine around Antarctica is produced 115 from algae growing on the underside of the relatively thin seasonal sea ice (Saiz-Lopez et al., 116 2015;Saiz-Lopez et al., 2007), satellite measurements do not show exceptional iodine activity above 117 Arctic sea ice (Spolaor et al., 2014), which may be due to differences in sea ice conditions between 118 both polar environments. In the Arctic, only above summer sea ice satellites are able to determine IO 119 emissions. Arctic boundary layer observations show sporadic enhanced atmospheric IO concentrations 120 related to the presence of ice-free open ocean conditions (Mahajan et al., 2010).

121 Here we present halogen records of the Akademii Nauk (AN) ice core (Opel et al., 2013) from 122 Severnaya Zemlya to assess their potential for the reconstruction of regional sea ice extend variability 123 in the Russian Arctic and to provide a new regional-scale sea-ice reconstruction, i.e. the easternmost 124 record of the Arctic. Severnaya Zemlya is located in the marine boundary layer and is surrounded by 125 winter Arctic sea ice. The AN ice core features annual resolution, and hence can be used to produce a 126 sensitive climate record for comparison to satellite, ship and land-based observations of sea ice area. 127 Combined with other circumpolar ice caps, this location allows the possibility to produce localized sea 128 ice reconstructions for the whole Arctic region. The bromine excess (Brexc) is expressed in terms of

concentrations (ng g^{-1}) and has been calculated by subtracting the seawater component from the total 129 130 bromine concentration using sodium as seawater proxy. Iodine concentrations have been used directly 131 without any seawater correction since the sea water contribution (using the average sea water 132 abundance) is less than 1%. These halogen data have been compared with summer and spring sea ice 133 areas from the Laptev and Kara seas, the two Arctic seas east and west of Severnaya Zemlya, 134 respectively. Our results suggest a strong connection between Brexc and spring sea ice changes in the Laptev Sea as well as a positive correlation between iodine and summer sea ice in the Laptev Sea. This 135 136 work continues investigations already done by our research group on the connection between halogens, 137 (I, Br) and sea ice changes (Spolaor et al., 2013a; Spolaor et al., 2013b). These data are the first 138 investigating the halogen climate signal in the Arctic in the last 50 years and shed new light on the 139 connections between halogen and past sea ice changes.

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141 2 DATA AND METHODS

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143 2.1 AKADEMII NAUK ICE CORE

A 724 m ice core from Akademii Nauk (AN) ice cap (80°31'N, 94°49'E, 760 m a.s.l., Figure 1) was 144 145 drilled from 1999 to 2001(Fritzsche et al., 2002), presenting the easternmost ice core record currently 146 available from the Arctic. Due to the relatively low altitude of the ice cap, the ice core shows evidence 147 of summer melt and infiltration processes (Opel et al., 2009) which may influence some of the 148 atmospheric records preserved in the ice (Fritzsche et al., 2005). Despite a mean annual air temperature 149 of -15.7°C (May 1999 to April 2000), surface melting occurs almost every year when temperatures 150 may rise above 0°C even at the ice cap summit and a considerable amount of Akademii Nauk ice core 151 consists of melt-layers and partly infiltrated firn (Opel et al., 2009). In the literature few studies have 152 reported the effect of meltwater percolation on the ice core climate signal. Pohjola et al., (Pohjola et al., 153 2002) studied the effect of percolation in the Lomonosovfonna ice core (Svalbard), a site that features 154 melting and infiltration conditions similar to the Akademii Nauk. Their results suggest that, though the 155 original seasonal climate signal could be disturbed especially for the anions associated with strong acids (NO₃⁻ and SO₄²⁻), most of the other chemical species and in particular the stable water isotopes 156 157 are less affected than the strong acids. In the Akademii Nauk ice core, stable isotope records still show 158 seasonal variations, indicating a minor impact of melting and refreezing only. Even though ions and 159 other species may be more affected a deep infiltration and redistribution will be obstructed by the 160 frequent occurrence of melt layers. Therefore, the Akademii Nauk ice core can be considered suitable for high-resolution (i.e. annual to multi-annual) reconstruction of paleoclimate and atmospheric aerosol loading as already shown for the past century (Weiler et al., 2005) and the past millennium (Opel et al., 2013). Bromine and iodine are photoactive species but recent findings suggest that Br is preserved in the snow after deposition, while iodine can be remobilized in the snow but still maintains the average annual climatic signal (Spolaor et al., 2014). The relatively high average accumulation rate of 0.46 m water equivalent per year, determined for the period 1956-1999, could also help to preserve the halogen signal in the snowpack.

The core chronology is based on counting of annual layers in stable water isotopes, constrained by the identification of reference horizons including the ¹³⁷Cs nuclear bomb test peak (1963 AD) and volcanic eruptions (Bezymianny 1956 AD). In this pilot study, we focus on the core section 0-29 m, representing the time period 1950-1999. The core chronology for the time period viewed here is well constrained by the detection of the 1956 AD volcanic eruption of Bezymianny (Kamchatka Peninsula) (Opel et al., 2013). Based on comparisons to other dating approaches (linear interpolation, age modeling) we estimate the dating uncertainties to be about ± 1 year, but definitely less than ± 3 years.

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176 2.2 HALOGENS ANALYSIS

177 Contiguous, longitudinal samples (1.0x0.033x0.033 m) were cut from the Akademii Nauk ice core and 178 shipped frozen to the Ultra-Trace Chemistry Laboratory at the Desert Research Institute for analyses 179 using a unique, continuous ice core measurement system (McConnell et al., 2002). Longitudinal 180 samples are melted consecutively on a carefully cleaned, engraved melter head that splits meltwater 181 from different parts of the sample cross-section into ultra-clean (innermost $\sim 10\%$), clean (next $\sim 20\%$) 182 and potentially contaminated (outermost part of the ice core \sim 70%) continuously flowing sample 183 streams. Elemental measurements are made on the ultra-clean sample stream, with ultra-pure nitric acid 184 added immediately after the melter plate to yield an acid concentration of $\sim 1\%$. The analytical system 185 includes two Thermo-Fisher Element II high-resolution Inductively Coupled Plasma Mass 186 Spectrometers operating in parallel and used to measure simultaneously >30 elements (McConnell et 187 al., 2014; Sigl et al., 2014) including Br, I, and Na, a Picarro L2130 water isotope analyser (Maselli et 188 al., 2013), a Droplet Measurement Technologies SP2 black carbon analyser (McConnell et al., 2007), 189 among other instruments for determination of ammonium, nitrate, hydrogen peroxide and other 190 chemical compounds (Pasteris et al., 2014). Effective depth resolution differs between the instruments 191 in the analytical system and operating parameters but in this study is estimated to be ~ 0.02 m for Br. I. 192 and Na, with all measurements exactly co-registered in depth. Detection limits are 0.1, 0.003, and 0.06

193 ng/g for Br, I, and Na, respectively.

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195 2.3 BACK TRAJECTORY CALCULATIONS

196 To understand the sources of air masses that influence the bromine and iodine deposition at the drill 197 site of the Akademii Nauk ice core we calculated back trajectories for the period covered by satellite 198 sea ice measurements (1979-2000). Three-day back trajectories for spring (March, April, May, MAM) 199 were chosen to investigate the bromine sources while six-day back trajectories for summer (June, July, 200 August; JJA) were chosen to investigate the iodine sources, where some long-life iodine organic 201 compounds and re-cycling processes can influence total iodine concentrations (Simpson et al., 2015). 202 Back Trajectories were calculated with the model HYSPLIT (HYbrid Single-Particle Lagrangian 203 Integrated Trajectory, Version September 2014) using NCEP/NCAR Reanalysis data from the National 204 Weather Service's National Centers for Environmental Prediction provided by the NOAA's Air 205 Resources Laboratory (Draxler and Hess, 1998). The arrival height was the drilling site at Akademii 206 Nauk ice cap located at 760 m a.s.l. Trajectories were calculated each 12 hours and vertical motion 207 mode was used. For a better comparison and an indicative estimation of the potential origin of air 208 masses, back trajectories of five-year periods were clustered for each three-month period (MAM and 209 JJA) for the years 1980 to 2000 (Figure 2).

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211 2.4 SEA ICE AREA AND ANOMALIES

212 Time series of the monthly mean sea ice area over the period January 1979 to December 2013 were 213 calculated for three source regions in the Arctic (Figure 1). These correspond to the Kara Sea (496,875 km²), the Laptev Sea (781,875 km²), and a subset of the Arctic Ocean (536,875 km²). Observations of 214 215 sea ice concentrations from passive microwave satellite radiometers were used as input data (Sea Ice 216 Concentrations from Nimbus-7 SMMR and DMSP SSM/I-SSMIS Passive Microwave Data) (Cavalieri 217 et al., 1996, updated yearly). Regional averages were produced for the sea ice concentration datasets 218 which are published at 25 km grid resolution resulting in a single sea ice concentration value for each 219 region every day. The time series were resampled to three-monthly averages for each region and 220 averaged by multiplying the mean sea ice concentration with the area of each region.

Typical uncertainties for sea ice concentration observations are between 5% and 10%, depending upon the time of year (greater uncertainty in summer) and the sea ice concentration range (greater uncertainty for lower ice concentrations). Although averaging over larger areas, such as those designated in Figure 1, will reduce the relative uncertainty we estimate the uncertainty of the sea ice data presented here to be no greater than 10%.

In addition to the satellite measurements we compare our results to the dataset produced by Polyakov et al. (Polyakov et al., 2003) reporting August sea ice anomalies in the Kara, Laptev, East Siberian and Chukchi Seas. The dataset was produced by compiling Russian historical records of fast ice locations in the Arctic seas from ship-based observations, hydrographic surveys and commercial shipping routes and aircraft-based observations. During World War II some missing data (1942-1945) were reconstructed using statistical regression models.

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3 RESULTS

3.1 Trajectories and sea ice area: the Laptev Sea basin influence

235 Air mass back trajectories suggest that the Laptev Sea basin is the predominant source for air masses 236 arriving at Akademii Nauk drill site during springtime (Figure 2). The percentage of the springtime air 237 masses originating in the Laptev basin range from a minimum of 44% (1976-1980) up to 53% (1991-238 1995), with the percentage of air masses originating from the Kara Sea and Arctic Ocean regions 239 defined here consistently lower. Hence, we consider the Laptev Sea basin to be the most important 240 source region for the spring climate signal present in the Akademii Nauk ice core. For the summer 241 period the sources are more variable with the Laptev Sea basin showing a majority of air mass sources 242 in three of the five 5-year periods investigated. During the decade 1981-1990, the percentage of 243 summer air masses from the Kara Sea exceeds those from the Laptev Sea (in particular between 1986-1990); however no associated changes have been detected in climate proxies such as δ^{18} O (Opel et al., 244 245 2009).

246 In addition to the back trajectories we calculate sea ice areas for the three assigned basins of the Arctic 247 Ocean and Laptev and Kara Sea regions. The results clearly demonstrate that the greatest variability of 248 sea ice area occurs in the Laptev Sea for both spring and summer sea ice. In particular the Arctic Ocean 249 region shows very small changes in summer minima and the production of first-year sea ice is hence 250 negligible compared to the other two basins. Seasonal changes in Kara Sea ice area are comparable to 251 those calculated for the Laptev Sea (Figure 3). However, considering that the air masses arriving at 252 Akademii Nauk in spring and summer originate primarily from the Laptev Sea and that this region 253 displays the greatest seasonal variability in sea ice area, we consider halogens concentration in the 254 Akademii Nauk ice core to be most likely dominated by sea ice variability in the Laptev Sea. 255 Nonetheless we also evaluate the possibility that the Kara Sea is an important secondary source of 256 halogens.

257 **3.2 Statistical analysis**

258 Given the results of the calculated airmass back trajectory and observed sea ice variability, we compare 259 the yearly average values of Brexc and I concentrations with summer and spring sea ice of Laptev and 260 Kara Seas. Because of the low variability detected in this part of the Arctic Ocean, this region was 261 excluded from statistical evaluation. Apart from the Polyakov anomalies, the parameters are 262 transformed to the logarithmic scale to reduce their asymmetry and thus improve on the adherence to 263 the normal distribution assumption used in the statistical analyses. Inspection of normal probability 264 plots confirms the absence of departures from the normality for the log-transformed parameters. 265 Furthermore, the series of log(I) is de-trended by subtracting a least-squares-fit straight line. 266 Autocorrelation and partial autocorrelation plots confirm the absence of serial correlation in the log-267 transformed parameters and in the detrended log(I).

268 Table 1 lists the correlations of the detrended log(I) and $log(Br_{exc})$ with various calculated parameters 269 of sea ice extent together with the p-values based on the unilateral t test for the correlation coefficient. 270 We consider correlations with a p-value smaller than 0.05 as an indication of a statistically significant 271 positive correlation. The data support a significant positive correlation between log(I) and the logarithm 272 of the Laptev Sea ice area both in Spring (r=0.50, df=20, p=0.009) and Summer (r=0.49, df=20, 273 p=0.011), while there is no evidence of correlation with the Kara Sea ice area in either season (Figure 274 4a). The results confirm also a positive correlation between $log(Br_{exc})$ and the logarithm of the Laptev 275 Sea spring sea ice (r=0.44, df=20, p=0.020) but not with Kara Sea spring sea ice (r=0.18, df=20, 276 p=0.205) (Figure 4b).

277 We also evaluate the correlation between log(Br_{exc}) and I with the Polyakov anomalies dataset. The 278 Polyakov anomalies dataset represents the summer (August) sea ice changes in the last century for 279 several Arctic basins including the Laptev and Kara Seas. The obtained results confirm the finding that $log(Br_{exc})$ is correlated with Laptev. The correlation between $log(Br_{exc})$ and the Polyakov anomalies 280 281 dataset is significant for both Laptev Sea sea ice (r=0.34, df=48, p-value=0.009) and Kara Sea sea ice 282 (r=0.31, df=48, p-value=0.015). As regards to iodine, data indicate a significant correlation between 283 log(I) and the Polyakov anomalies data for Laptev Sea sea ice (r=0.32, df=48, p-value=0.012) but not 284 for Kara Sea ice (r=0.17, df=48, p-value=0.116). These results, together with back trajectory 285 calculation, support the Laptev Sea as the main location of sea ice variability influencing Brexc and I in 286 the Akademii Nauk ice core.

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288 4 DISCUSSION

290 4.1 Brexc and Laptev Sea spring sea ice

291 Bromine explosions are defined as the autocatalytic sequence of chemical reactions able to generate gas 292 phase bromine species (such as BrO) and they occur mainly above seasonal sea ice (Figure 5). Bromine 293 explosions lead to enrichment of Br in snow deposition, well beyond the Br/Na ratio observed in 294 seawater. Bromine can be also emitted by other sources, such as biological activity or anthropogenic 295 emissions. In addition heterogeneous reactions on sea-salt particles and sea-salts on snow or sea-ice can 296 contribute to the total emission of Br from the bromine explosion. However satellite images and 297 previous results suggest that these processes are secondary to the sea ice-based bromine emission. 298 There are two calculations that can be used to quantify the influence of bromine explosions in 299 snowpack: bromine excess (Br_{exc}) and bromine enrichment (Br_{enr}). Both calculations use the same 300 Br/Na seawater concentration ratio of 0.006 (Turekian, 1968). Br_{exc} is calculated as $Br_{exc} = [Br] - [Na]$ x 0.006 and indicates how much bromine has been produced by the bromine explosion. Brenr is 301 calculated as $Br_{enr} = [Br] / ([Na] \times 0.006)$ and indicates the factor by which bromine has been enriched 302 303 beyond the seawater ratio. Comparing both Br indicators (Brexc and Brenr) reveals that they have a 304 similar trend over the past 50 years, with a peak in the 1970s and 1980s and a sharp decrease in the last 305 decade of the record (Figure 6). To investigate the connection between Br and sea ice we used Br_{exc} 306 since it allows quantification of the additional bromine fluxes produced by the bromine explosion. 307 Additionally, the Akademii Nauk ice core was subject to summer melt and hence may be susceptible to 308 artefacts based on the different percolation velocities of Br and Na. Air mass back trajectory data 309 indicate the Laptev Sea is the primary springtime source of air masses arriving at Akademii Nauk so we compare Br_{exc} with spring sea ice area in the Laptev Sea. It must be noted that bromine explosions 310 311 are promoted by first year sea ice (Simpson et al., 2007). Considering the strong sea ice recycling in the 312 Laptev Sea and the small amount of summer sea ice area compared to winter sea ice (Figure 3), we 313 consider the whole spring sea ice area to be an active site for the bromine explosion. A strong influence 314 of the bromine explosion also is expected due to the observation that only 15 to 30 % of sea ice is 315 present by late summer, implying that almost all spring sea ice is seasonal. Furthermore, Laptev Sea ice 316 can be exported by the Transpolar Drift through the Arctic Ocean to the Greenland Sea and North 317 Atlantic, leading to strong recycling of sea ice in the area (Xiao et al., 2013). Comparing the logtransformed Br_{exc} and Laptev Sea spring sea ice we find a positive correlation (r = 0.44). Over the 318 319 period for which observations are available, noting that both spring sea ice area and Brexc exhibit a 320 steadily decreasing trend (Figure 7a). In 1991 in particular, we observe a sharp decrease of spring sea 321 ice followed by decreasing production of Br_{exc}. The intensity of bromine explosion and the production 322 of BrO in the Arctic are strongly driven by sea ice presence, in this case spring sea ice. Log-323 transformed Brexc also was compared with a 50-year record of summer (August) sea ice anomalies for 324 the Laptev basin, calculated by Polyakov et al. (Polyakov et al., 2003) (Figure 7b). We found a positive 325 correlation between the two data series (r=0.31). We emphasize that the degree of correlation may be 326 influenced by the fact that we are not comparing $log(Br_{exc})$ to the sea ice area at the season of peak 327 emission. Although spring and summer sea ice areas may have different short-term trends, the general 328 trend is similar over the decadal period considered here. Although our results suggest that excess of Br 329 in snow deposition seem mainly driven by the changes in sea ice area, especially first year sea ice, we 330 also consider other factors that could contribute to the total nss-Br in the snow deposition. Bromine 331 explosions are catalyzed by acidity and light, and a change in the acidic condition of the Arctic 332 atmosphere could produce a change in the magnitude of bromine activation and hence bromine 333 deposition (Sander et al., 2006). Other process can interferes on bromine deposition. The recycling of 334 bromine in airborne sea salt aerosols could also impact the bromine concentrations. Quantification of 335 the impacts of acidity changes and other process on Br_{exc} (or Br_{enr}) deposition is beyond the scope of 336 this study and additional studies are necessary to better understand which factors (e.g., sea ice area, 337 acidity, wind pumping, temperature) are the most relevant. However the highly significant correlations and previous research suggests that sea ice is the main driver of Brexc. 338

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340 4.2 Iodine and Laptev Sea summer sea ice

341 While the main source of iodine in Antarctica has been attributed to sea ice algae during spring time 342 (Atkinson et al., 2012;Saiz-Lopez et al., 2007;Saiz-Lopez et al., 2015) and recently also to inorganic 343 emissions during winter time (Granfors et al., 2014), in the Arctic there is a lack of knowledge 344 regarding iodine sources. One of the main barriers to identifying Arctic iodine sources is that the 345 atmospheric concentrations of IO are close to satellite detection limits (Figure 5). Iodine could be 346 emitted also by heterogeneous reactions on sea ice, snowpack, or marine aerosol surfaces but the 347 general idea remains that the control of iodine atmospheric concentration is due to biological activities 348 associated with sea ice. The first measurements of iodine in Arctic aerosols (Sturges and Barrie, 1988) 349 found increasing concentrations during late spring and summer. More recently Mahajan et al. (2010) 350 detected iodine emissions in the form of IO, related to the presence of polynyas or areas free of sea ice 351 over Hudson Bay, Canada. These open water areas were identified as hot spots for iodine emission. 352 Arctic sea ice is thicker and less permeable than Antarctic sea ice, constituting a barrier between the 353 ocean and the atmosphere (Zhou et al., 2013). In this case it is likely that gas phase iodine produced 354 from sea ice algae could escape only from sea ice leads (cracks) or the open ocean surface (Saiz-Lopez 355 et al., 2015). The Akademii Nauk iodine signal is positively correlated with satellite observations of 356 Laptev Sea summer sea ice (r=0.49, Table 1 and Figure 8a) and with spring sea ice (r=0.50). 357 Concentrations of I also have been compared with summer (August) sea ice anomalies for the Laptev 358 basin calculated by Polyakov et al. (Polyakov et al., 2003) (Figure 8b). A significant positive 359 correlation (r=0.32, Table 1) was observed also in this case. No significant correlation was observed 360 with summer or spring sea ice area in the Kara Sea (Table 1), supporting the back-trajectory analysis 361 data. These results are consistent with iodine concentrations from a snow pit sampled at the NEEM ice 362 core site in northwest Greenland. The NEEM record displayed peak iodine concentrations in summer 363 (Spolaor et al., 2014), although only three annual cycles were sampled. In contrast, iodine 364 concentrations from Svalbard were more consistent with spring sea ice area (Spolaor et al., 2013a). The 365 results from Svalbard must be evaluated with caution because seasonal variability is less clear in the 366 Svalbard records and the influence of summer melting and iodine re-emission may be more significant 367 than for the Akademii Nauk ice core. The summer peak of iodine concentrations found in Arctic snow 368 and ice may depend upon both the production of iodine by sea ice algae and processes that control 369 release of iodine to the atmosphere. Arctic sea ice is an effective barrier to gas exchange in the ocean-370 atmosphere system especially in winter, when there is low ice permeability and minimal convection of 371 Arctic waters (Zhou et al., 2013). Ice permeability follows Arctic temperatures, with maximum 372 permeability in summer and particularly when the sea ice is warmer than -5°C. Water convection is 373 greatest during spring, driven by an unstable brine density profile. As suggested for the Antarctic (Saiz-374 Lopez et al., ACP, 2015), fracturing is an additional process that may enhance gas exchange, and hence 375 iodine emission from the Arctic Seas, particularly in the highly dynamic Laptev Sea. Phytoplankton 376 blooms occur mainly in late spring and summer (Ardyna et al., 2013;Kevin R. Arrigo et al., 2012), 377 responding to the availability of nutrients and the limited penetration of light through thick winter sea 378 ice. These factors ensure that the most efficient season for production and emission of iodine is the 379 Arctic summer. In summer, the decreased sea ice thickness, increased sea ice fracturing and 380 permeability, and enhanced light penetration and nutrient availability, could create conditions 381 necessary to sustain blooms of sea ice algae responsible for IO production (Kevin R. Arrigo et al., 382 2012; Allan et al., 2015). In years of decreased Arctic sea ice area, the available substratum for growing 383 algae is reduced and hence IO production is inhibited. This explanation fits well with the data available 384 from Greenland (Spolaor et al., 2014) and presented here for the Akademii Nauk ice core. Satellite

385 measurements of Arctic IO concentrations exceed detection limits only in the summer, and at that time 386 of maximum IO concentrations above summer sea ice. However, these results can only be used as 387 indications. Improved sensitivity of satellite-borne instruments would greatly enhance the detection and 388 seasonal variability of IO emission sources in the Arctic.

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390 5. CONCLUSIONS

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392 The halogens iodine and bromine reported here from the Akademii Nauk ice core from Severnaya 393 Zemlya offer a new perspective on the variability of sea ice extent in the Arctic. This work follows on 394 previous work to suggest a connection of bromine and iodine chemistry with sea ice changes (Spolaor et al., 2013a;Spolaor et al., 2014;Spolaor et al., 2013b). In particular, Brexc and Brenr have been linked 395 396 to seasonal sea ice area and here we report a connection between ice-core Brexc and spring sea ice in the 397 Laptev Sea largely because almost all Laptev Sea spring sea ice is seasonal (approximately 80%) and 398 so undergoes continuous renewal. Ice-core iodine appears to be connected with the summer sea ice 399 area; however, its sources are suggested to be biologically-mediated, which makes the interpretation 400 more difficult, and atmospheric concentrations of IO in the Arctic are close to satellite detection limits, 401 limiting the accurate characterization of IO emission sources. Further studies are necessary to better 402 identify the seasonal variability of this element and the impact of acidity on bromine reactivity in 403 Arctic. However, the significant correlation between nss-Br and sea ice during the last 50 years in the 404 Akademii Nauk ice-core record suggests that the nss-Br record primarily reflects changing sea ice 405 conditions in the Laptev sea, bringing important new information for the use of halogens in 406 paleoclimate studies and suggesting that the Akademii Nauk ice core may be a key archive for 407 reconstructing late-Holocene sea ice variations in the Russian Arctic.

408

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- 419
- 420

421 TABLE

Table 1. Correlations (r) of the detrended $\log(I)$ and $\log(Br_{exc})$ with the logarithm of the first year sea ice area in the Laptev and Kara seas for the period 1979-1999 and the Polyakov anomalies in the Laptev and Kara seas for the period 1950-1999 (denoted by an asterisk *) (Polyakov et al., 2003). Columns display the r-values and p-values. Bold numbers indicate the statistically significant correlations. No correlation has been calculated between log(Brexc) and summer sea ice since the Bromine explosion occur during spring time.

	log (Br _{exc})		log(I)	
	r	p-value	r	p-value
Laptev Sea spring	0,44	0,020	0,50	0,009
Kara Sea spring	0,18	0,205	0,04	0,433
Laptev Sea summer		—	0,49	0,011
Kara Sea summer			-0,03	0,561
Polyakov* - Laptev Sea	0,31	0,009	0,32	0,012
Polyakov* - Kara Sea	0,34	0,015	0,17	0,116

448 FIGURES

Figure 1. Arctic areas considered for sea ice calculations. Kara Sea (pink), Laptev Sea (dark blue) and
a subset of the Arctic Ocean (green). The location of the Akademii Nauk ice core drill site on
Severnaya Zemlya also is shown.

452

Figure 2. Air-mass back trajectories calculated for the period 1976 – 2000. Each panel represents the
5-year average for spring (MAM, left column) and summer (JJA, right column) seasons.

455

Figure 3. Sea ice area variation in the period 1979 - 2000 for the three regions defined in Figure 1; (a)
Arctic Ocean, (b) Kara Sea and (c) Laptev Sea. Laptev Sea ice shows the greatest seasonal variability.

459 Figure 4. Scatter plot of log(Br_{exc}) and Spring Laptev sea ice area (a - black squares) and detrended
460 Iodine and summer Laptev sea ice area (b - black circles).

461

Figure 5. Atmospheric column averages of BrO and IO in the Arctic between 2009 and 2011. In the
Arctic, IO concentrations are near the limit of detection by satellite (after Spolaor et al., 2014).

Figure 6. Comparison between bromine excess (Br_{exc}) concentrations, bromine enrichment (Br_{enr}) ratios and Sodium (ng g⁻¹) from Akademii Nauk ice core. Raw (grey line) and 3-year smoothed (black line) Br_{exc} concentrations and raw (pink line) and 3-year smoothed (red line) Br_{enr} are shown on logarithmic scales. Raw (light blue) and 3-year smoothed (dark blue) sodium data are also shown.

469

Figure 7. Br_{exc} compared with sea ice area during spring and summer in the Laptev Sea region. In both
panels, raw (grey line) and 3-year smoothed (black line) Br_{exc} data from Akademii Nauk ice core are
shown. The Laptev Sea spring sea ice data are based on satellite observations (blue line, panel a).
Laptev Sea summer sea ice anomalies (red line, panel b) are from Polyakov et al. (2003).

474

Figure 8. Iodine compared with Laptev Sea summer sea ice area. In both panels, raw (grey line) and 3year smoothed (black line) [I] data from Akademii Nauk ice core are shown. The Laptev Sea summer sea ice data are based on satellite observations (blue line, panel a) and the reconstruction of Polyakov et al. (2003) (red line, panel b).





488 FIGURE 3







518 FIGURE 5

552 553 FIGURE 6

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